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## (54) Title: AMIDOAROMATIC RING SULFONAMIDE HYDROXAMIC ACID COMPOUNDS

### (57) Abstract

An amidoaromatic ring sulfonamide hydroxamic acid compound that *inter alia* inhibits matrix metalloprotease activity is disclosed, as are a treatment process that comprises administering a contemplated amidoaromatic ring sulfonamide hydroxamic acid compound in an MMP enzyme-inhibiting effective amount to a host having a condition associated with pathological matrix metalloprotease activity.

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# AMIDOAROMATIC RING SULFONAMIDE HYDROXAMIC ACID COMPOUNDS

### Description

### 5 Technical Field

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This invention is directed to proteinase
(protease) inhibitors, and more particularly to
amidoaromatic ring sulfonamide hydroxamic acid
compounds that, inter alia, exhibit activity as

10 inhibitors for matrix metalloproteinases, compositions
of proteinase inhibitors, intermediates for the
syntheses of proteinase inhibitors, processes for the
preparation of proteinase inhibitors and processes for
treating pathological conditions associated with

15 pathological matrix metalloproteinase activity.

### Background of the Invention

Connective tissue, extracellular matrix constituents and basement membranes are required components of all mammals. These components are the biological materials that provide rigidity, differentiation, attachments and, in some cases, elasticity to biological systems including human beings and other mammals. Connective tissues components include, for example, collagen, elastin, proteoglycans, fibronectin and laminin. These biochemicals makeup, or are components of structures, such as skin, bone, teeth, tendon, cartilage, basement membrane, blood vessels, cornea and vitreous humor.

Under normal conditions, connective tissue turnover and/or repair processes are controlled and in

equilibrium. The loss of this balance for whatever reason leads to a number of disease states. Inhibition of the enzymes responsible loss of equilibrium provides a control mechanism for this tissue decomposition and, therefore, a treatment for these diseases.

Degradation of connective tissue or connective tissue components is carried out by the action of proteinase enzymes released from resident tissue cells and/or invading inflammatory or tumor cells. A major class of enzymes involved in this function are the zinc metalloproteinases (metalloproteases).

The metalloprotease enzymes are divided into classes with some members having several different names in common use. Examples are: collagenase I 15 (MMP-1, fibroblast collagenase; EC 3.4.24.3); collagenase II (MMP-8, neutrophil collagenase; EC 3.4.24.34), collagenase III (MMP-13), stromelysin 1 (MMP-3; EC 3.4.24.17), stromelysin 2 (MMP-10; EC 3.4.24.22), proteoglycanase, matrilysin (MMP-7), 20 gelatinase A (MMP-2, 72kDa gelatinase, basement membrane collagenase; EC 3.4.24.24), gelatinase B (MMP-9, 92kDa gelatinase; EC 3.4.24.35), stromelysin 3 (MMP-11), metalloelastase (MMP-12, HME, human macrophage elastase) and membrane MMP (MMP-14). MMP is 25 an abbreviation or acronym representing the term Matrix Metalloprotease with the attached numerals providing differentiation between specific members of the MMP group.

The uncontrolled breakdown of connective tissue by metalloproteases is a feature of many

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pathological conditions. Examples include rheumatoid arthritis, osteoarthritis, septic arthritis; corneal, epidermal or gastric ulceration; tumor metastasis, invasion or angiogenesis; periodontal disease; proteinuria; Alzheimer's Disease; coronary thrombosis and bone disease. Defective injury repair processes also occur. This can produce improper wound healing leading to weak repairs, adhesions and scarring. These latter defects can lead to disfigurement and/or permanent disabilities as with post-surgical adhesions.

10 Matrix metalloproteases are also involved in the biosynthesis of tumor necrosis factor (TNF) and inhibition of the production or action of TNF and related compounds is an important clinical disease treatment mechanism. TNF- $\alpha$ , for example, is a cytokine 15 that at present is thought to be produced initially as a 28 kD cell-associated molecule. It is released as an active, 17 kD form that can mediate a large integer of deleterious effects in vitro and in vivo. For example, TNF can cause and/or contribute to the effects of 20 inflammation, rheumatoid arthritis, autoimmune disease, multiple sclerosis, graft rejection, fibrotic disease, cancer, infectious diseases, malaria, mycobacterial infection, meningitis, fever, psoriasis, 25 cardiovascular/pulmonary effects such as post-ischemic reperfusion injury, congestive heart failure, hemorrhage, coagulation, hyperoxic alveolar injury,

radiation damage and acute phase responses like those seen with infections and sepsis and during shock such as septic shock and hemodynamic shock. Chronic release

of active TNF can cause cachexia and anorexia. TNF can be lethal.

TNF-α convertase is a metalloproteinase involved in the formation of active TNF-α. Inhibition of TNF-α convertase inhibits production of active TNF-α. Compounds that inhibit both MMPs activity have been disclosed in WIPO International Publication Nos. WO 94/24140, WO 94/02466 and WO 97/20824. There remains a need for effective MMP and TNF-α convertase inhibiting agents. Compounds that inhibit MMPs such as collagenase, stromelysin and gelatinase have been shown to inhibit the release of TNF (Gearing et al. Nature 376, 555-557 (1994), McGeehan et al., Nature 376, 558-561 (1994)).

15 MMPs are involved in other biochemical processes in mammals as well. Included is the control of ovulation, post-partum uterine involution, possibly implantation, cleavage of APP (β-Amyloid Precursor Protein) to the amyloid plaque and inactivation of  $\alpha_1\text{-protease}$  inhibitor  $(\alpha,\text{-PI})_+$  Inhibition of these 20 metalloproteases permits the control of fertility and the treatment or prevention of Alzheimers Disease. In addition, increasing and maintaining the levels of an endogenous or administered serine protease inhibitor 25 drug or biochemical such as  $\alpha_1$ -PI supports the treatment and prevention of diseases such as emphysema, pulmonary diseases, inflammatory diseases and diseases of aging such as loss of skin or organ stretch and resiliency.

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Inhibition of selected MMPs can also be desirable in other instances. Treatment of cancer and/or inhibition of metastasis and/or inhibition of angiogenesis are examples of approaches to the treatment of diseases wherein the selective inhibition of stromelysin (MMP-3), gelatinase (MMP-2), or collagenase III (MMP-13) are the relatively most important enzyme or enzymes to inhibit especially when compared with collagenase I (MMP-1). A drug that does not inhibit collagenase I can have a superior therapeutic profile. Osteoarthritis, another prevalent disease wherein it is believed that cartilage degradation in inflamed joints is at least partially caused by MMP-13 released from cells such as stimulated chrondrocytes, may be best treated by administration of 15 drugs one of whose modes of action is inhibition of MMP-13. See, for example, Mitchell et al., J. Clin. Invest., 97:761-768 (1996) and Reboul et al., J. Clin. Invest., 97:2011-2019 (1996).

Inhibitors of metalloproteases are known.

Examples include natural biochemicals such as tissue inhibitor of metalloproteinase (TIMP), α<sub>2</sub>-macroglobulin and their analogs or derivatives. These are high molecular weight protein molecules that form inactive complexes with metalloproteases. An integer of smaller peptide-like compounds that inhibit metalloproteases have been described. Mercaptoamide peptidyl derivatives have shown ACE inhibition in vitro and in vivo. Angiotensin converting enzyme (ACE) aids in the production of angiotensin II, a potent pressor

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substance in mammals and inhibition of this enzyme leads to the lowering of blood pressure.

Thiol group-containing amide or peptidyl amide-based metalloprotease (MMP) inhibitors are known as is shown in, for example, WO95/12389, WO96/11209 and U.S. 4,595,700. Hydroxamate group-containing MMP inhibitors are disclosed in a number of published patent applications such as WO 95/29892, WO 97/24117, and EP 0 780 386 that disclose carbon backboned compounds, and WO 90/05719, WO 93/20047, WO 95/09841 and WO 96/06074 that disclose hydroxamates that have a peptidyl back-bones or peptidomimetic backbones, as does the article by Schwartz et al., Progr. Med. Chem., 29:271-334(1992) and those of Rasmussen et al., Pharmacol. Ther., 75(1): 69-75 (1997) and Denis et al., Invest. New Drugs, 15(3): 175-185 (1997).

One possible problem associated with known MMP inhibitors is that such compounds often exhibit the same or similar inhibitory effects against each of the MMP enzymes. For example, the peptidomimetic hydroxamate known as batimastat is reported to exhibit IC50 values of about 1 to about 20 nanomolar (nM) against each of MMP-1, MMP-2, MMP-3, MMP-7, and MMP-9. Marimastat, another peptidomimetic hydroxamate was reported to be another broad-spectrum MMP inhibitor with an enzyme inhibitory spectrum very similar to batimastat, except that marimastat exhibited an IC50 value against MMP-3 of 230 nM. Rasmussen et al., Pharmacol. Ther., 75(1): 69-75 (1997).

Meta analysis of data from Phase I/II studies using marimastat in patients with advanced, rapidly

progressive, treatment-refractory solid tumor cancers (colorectal, pancreatic, ovarian, prostate).indicated a dose-related reduction in the rise of cancer-specific antigens used as surrogate markers for biological activity. The most common drug-related toxicity of marimastat in those clinical trials was musculoskeletal pain and stiffness, often commencing in the small joints in the hands, spreading to the arms and shoulder. A short dosing holiday of 1-3 weeks followed by dosage reduction permits treatment to continue. Rasmussen et al., Pharmacol. Ther., 75(1): 69-75 (1997). It is thought that the lack of specificity of inhibitory effect among the MMPs may be the cause of that effect.

In view of the importance of hydroxamate MMP inhibitor compounds in the treatment of several diseases and the lack of enzyme specificity exhibited by two of the more potent drugs now in clinical trials, it would be a great benefit if hydroxamates of greater enzyme specificity could be found. This would be 20 particularly the case if the hydroxamate inhibitors exhibited limited inhibition of MMP-1 that is relatively ubiquitous and as yet not associated with any pathological condition, while exhibiting quite high inhibitory activity against one or more of MMP-2, MMP-9 or MMP-13 that are associated with several pathological conditions. The disclosure that follows describes one family of hydroxamate MMP inhibitors that exhibit those desirable activities.

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The present invention provides compounds and their pharmaceutically acceptable salts effective as inhibitors of matrix metalloprotease enzyme activity; the provision of such compositions that are effective for the inhibition of metalloproteases (MMPs) believed 5 to be implicated in diseases and disorders involving uncontrolled breakdown of connective tissue. diseases and disorders (pathological conditions) include, for example, rheumatoid arthritis, 10 osteoarthritis, septic arthritis, corneal, epidermal or gastric ulceration, snake bite, tumor metastasis, growth, invasion or angiogenesis, periodontal disease, proteinuria, Alzheimer's Disease, multiple sclerosis, coronary thrombosis and bone disease. Also contemplated are the provision of processes for 15 preparing such compositions; the provision of processes for treating pathological conditions associated with abnormal matrix metalloprotease activity. A contemplated process effective for treating such pathological conditions acts by selective inhibition of 20 metalloproteases associated with such conditions with minimal side effects resulting from inhibition of other proteases whose activity is necessary or desirable for normal body function.

Briefly, therefore, the present invention is directed to a compound of Formula I or a pharmaceutically acceptable acid or base addition salt of a compound of Formula I, as well as a pharmaceutical composition of a compound of Formula I or a pharmaceutically acceptable acid or base addition salt 30 of a compound of Formula I, and also a process for

treating conditions associated with pathological matrix metalloprotease activity comprising administering a matrix metalloprotease inhibitor in an effective dosage to a host suffering from such condition.

5 The present invention relates to a compound of Formula I:

HO 
$$R^{13}$$
  $R^{4}$   $R^{3}$   $(0)_{n}$ 

wherein:

n is an integer zero, 1 or 2;

W is independently selected from the group consisting of  $-NR^5COR^6$ ,  $-NR^5S(O)_ZR^7$  where z is zero, 1, or 2,  $-NR^5COOR^8$ ,  $-NR^5COOR^8R^9$  and  $-NR^{11}R^{12}$ ;

R<sup>1</sup> is cycloalkylene, arylene or

15 heteroarylene;

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R2 is selected from the group consisting of a hydrogen (hydrido), alkyl, aralkyl, heteroaralkyl, cycloalkylalkyl, heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl, hydroxycarbonylalkyl, aroylalkyl, and heteroaroylalkyl group, -(CH<sub>2</sub>)x-NR<sup>11</sup>R<sup>12</sup>, or -(CH<sub>2</sub>)x-C(O)NR<sup>11</sup>R<sup>12</sup>, wherein x is an integer from zero to 6; R3 is selected from the group consisting of a hydrogen (hydrido), alkyl, aryl, aralkyl, thioalkyl, heteroaralkyl, heteroaryl, alkoxyalkoxyalkyl, trifluoromethylalkyl, alkoxycarbonylalkyl,

aralkoxycarbonylalkyl, hydroxycarbonylalkyl, alkoxyalkyl, heterocycloalkylalkyl, aryloxyalkyl, alkylthioalkyl, arylthioalkyl, heteroarylthioalkyl group, or a sulfoxide or sulfone of any of said thiocontaining groups, a -(CH<sub>2</sub>)x-C(O)NR<sup>11</sup>R<sup>12</sup> group, wherein x is an integer from zero to 6, and a -(CH<sub>2</sub>)y-W group, wherein y is an integer from 1 to 6 and W is defined above:

cr R<sup>2</sup> and R<sup>3</sup> together with the atom

10 chain to which they are attached form a 3-8 membered ring;

 $R^4$  is a hydrogen (hydrido) or  $C_1$ - $C_4$  alkyl group;

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group;

R<sup>5</sup> is a hydrogen (hydrido) or C<sub>1</sub>-C<sub>4</sub> alkyl

R6 is selected from the group consisting of a hydrogen (hydrido), cycloalkyl, heterocycloalkyl, aryl, heteroaryl, aralkyl, heteroaralkyl, cycloalkylalkyl, heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl group, and a  $-(CH_2)x-NR^{11}R^{12}$  group wherein x is an 20 integer from zero to 6. The aryl or heteroaryl groups of R6 are optionally substituted (unsubstituted or substituted) with one or more substituents independently selected from the group consisting of a 25 halo, C1-C4 alkyl, C1-C4 alkoxy, nitro, cyano, hydroxy, carboxy, hydroxycarbonylalkyl, -(CH2)x-NR11R12, wherein x is an integer from zero to 6, trifluoromethyl, alkoxycarbonyl, aminocarbonyl, thio, alkylsulfonyl, carbonylamino, aminosulfonyl, alkylsulfonamino,

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alkoxyalkyl, cycolalkyloxy, alkylthioalkyl or alkylthio;

a) or  $\mathbb{R}^5$  and  $\mathbb{R}^6$  together with the atom chain to which they are bonded form a 5- to 7-membered a cyclic amide or imide that is substituted or unsubstituted;

b) or  $R^5$  and  $R^7$  together with the atom chain to which they are bonded form a 5- to 7-membered a cyclic sulfonamide that is substituted or unsubstituted;

 ${\tt R}^7$  is selected from the group consisting of  ${\tt R}^6$  and alkyl;

 $R^8$  and  $R^9$  are independently selected from the group consisting of  $R^6$  and alkyl, or  $R^8$  and  $R^9$  together with the depicted nitrogen atom form a 5- to 7-membered ring containing zero or one heteroatom that is oxygen, nitrogen or sulfur;

R11 and R12 are independently selected from the group consisting of a hydrogen (hydrido), alkyl, aryl, aralkyl, heteroaryl, heteroaralkyl, cycloalkyl, alkanoyl, aralkanoyl, and heteroaralkanoyl group, or R11 and R12 taken together form a 5 to 8-membered heterocyclo or heteroaryl ring; and

 $R^{13}$  is a hydrogen (hydrido) or  $C_1$ - $C_6$  alkyl 25 group.

The present invention also relates to a compound of Formula II:

HO 
$$R^{13}$$
  $CH_2$   $R^4$   $CH_2$   $R^4$   $CH_2$   $R^4$   $R$ 

wherein:

m is an integer from 1 to 6; ....

5 W,  $R^1$ ,  $R^2$  and  $R^{13}$  have the meanings described above;

 $R^4$  is a hydrogen (hydrido) or  $C_1\text{-}C_4$  alkyl group, as before;

or  $\mathbb{R}^4$  and W of  $-(CH_2)x-W$  together with the atom chain to which they are attached form a 4-8-membered ring.

Another particular embodiment of the invention relates to a compound of Formula III:

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HO 
$$R^3$$
  $R^4$   $R^3$   $(0)_2$ 

wherein  $\mathbb{R}^2$ ,  $\mathbb{R}^3$ ,  $\mathbb{R}^4$ ,  $\mathbb{R}^6$  and  $\mathbb{R}^{13}$  are as described above.

One particular embodiment of the invention relates to a compound of Formula IV:

HO 
$$R^3$$
  $R^4$   $R^3$   $(0)_2$ 

wherein  ${\tt R}^2,~{\tt R}^3,~{\tt R}^4,~{\tt R}^8$  and  ${\tt R}^{13}$  are as defined previously.

5 A further particular embodiment of the invention relates to a compound of Formula V:

HO 
$$R^{3}$$
  $R^{4}$   $R^{3}$   $\left( \begin{array}{c} \\ \\ \\ \\ \end{array} \right)_{2}$ 

wherein  $\mathbb{R}^2$ ,  $\mathbb{R}^3$ ,  $\mathbb{R}^4$ ,  $\mathbb{R}^7$  and  $\mathbb{R}^{13}$  are as defined previously.

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Yet another particular embodiment of the invention relates to a compound of Formula VI:

wherein  $R^2$ ,  $R^3$ ,  $R^4$ ,  $R^8$ ,  $R^9$  and  $R^{13}$  are as defined previously.

A still further particular embodiment of the invention relates to a compound of Formula VII:

$$R^{14}O$$

$$\downarrow_{R^{13}} R^4$$

$$\downarrow_{R^3} R^3 \left( \begin{array}{c} \\ \\ \\ \\ \end{array} \right)_{n}$$

$$VII$$

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... wherein:

 $_{\rm n,\ W,\ R^{\rm 1},\ R^{\rm 2},\ R^{\rm 3},\ R^{\rm 4}}$  and  $_{\rm R^{\rm 13}}$  are as defined previously, and

R14 is selected from the group consisting of a hydrido, C1-C6 alkyl, aryl, substituted aryl, arylalkyl, substituted arylalkyl, alkanoyl, cycloalkylcarbonyl, aralkanoyl, aroyl, and heterocyclylcarbonyl group. R14 is preferably a hydrido group, in which case a compound of formula VII becomes a compound of formula I, or an acyl group such as an alkanoyl, cycloalkylcarbonyl, aralkanoyl, aroyl, and heterocyclylcarbonyl group.

A contemplated compound contains an asymmetric carbon atom at the alpha-position so that 20 enantiomeric, d and l or R and S, forms of each compound exist. A particularly preferred stereoconfiguration for a contemplated enantiomeric compound is shown generically below in Formulas IA and VIIA, wherein R<sup>3</sup> is hydrido and not depicted, and W, n, the depicted R groups are as defined before.

HO 
$$R^{13}$$
  $R^{4}$   $R^{14}$   $R^{14}$ 

In the above formulas, the dashed line represents a bond that extends beneath the plane of the page,

5- whereas the solid wedge-shaped line represents a bond that extends above the plane of the page, as is usual in stereochemical depictions.

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As utilized herein, the term "alkyl", alone or in combination, means a straight-chain or branchedchain alkyl radical containing from 1 to about 12, preferably from 1 to about 10, carbon atoms. Examples of such radicals include methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, sec-butyl, tert-butyl, pentyl, iso-amyl, hexyl, octyl and the like. The term "alkenyl", alone or in combination, means a straightchain or branched-chain hydrocarbon radial having one or more double bonds and containing from 2 to about 12 carbon atoms preferably from 2 to about 10 carbon atoms. Examples of suitable alkenyl radicals include ethenyl (vinyl), 2-propenyl, 3-propenyl, 1,4pentadienyl, 1,4-butadienyl, 1-butenyl, 2-butenyl, 3butenyl, decenyl and the like. The term "alkynyl", alone or in combination, means a straight-chain hydrocarbon radical having one or more triple bonds and containing from 2 to about 12 carbon atoms preferably from 2 to about 10 carbon atoms. Examples of alkynyl

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radicals include ethynyl, 2-propynyl, 3-propynyl, decynyl, 1-butynyl, 2-butynyl, 3-butynyl, and the like.

The term "carbonyl", alone or in combination, means a -C(=0) - group wherein the remaining two bonds (valences) can be independently substituted. The term "thiol" or "sulfhydryl", alone or in combination, means a -SH group. The term "thio" or "thia", alone or in combination, means a thiaether group; i.e., an ether group wherein the ether oxygen is replaced by a sulfur atom.

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The term "amino", alone or in combination, means an amine or -NH2 group whereas the term monosubstituted amino, alone or in combination, means a substituted amine -N(H)(substituent) group wherein one hydrogen atom is replaced with a substituent, and disubstituted amine means a -N(substituent)2 wherein two hydrogen atoms of the amino group are replaced with independently selected substituent groups. Amines, amino groups and amides are classes that can be designated as primary (I°), secondary (II°) or tertiary 20 (III°) or unsubstituted, mono-substituted or disubstituted depending on the degree of substitution of the amino nitrogen. Quaternary amine (IV°) means a nitrogen with four substituents (-N+(substituent)4) that is positively charged and accompanied by a counter 25 ion or N-oxide means one substituent is oxygen and the group is represented as (-N+(substituent)3-0-), i.e., the charges are internally compensated.

The term "cyano", alone or in combination, 30 means a -C-triple bond-N (-CN) group. The term

"azido", alone or in combination, means a -N-double bond-N-double bond-N (-N=N=N) group.

The term "hydroxyl", alone or in combination, means a -OH group. The term "nitro", alone or in combination, means a -NO2 group.

The term "azo", alone or in combination, means a -N=N- group wherein the bonds at the terminal positions are independently substituted. The term "hydrazino", alone or in combination, means a -NH-NH-group wherein the remaining two bonds (valences) are independently substituted. The hydrogen atoms of the hydrazino group can be replaced, independently, with substituents and the nitrogen atoms can form acid addition salts or be quaternized.

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The term "sulfonyl", alone or in combination, means a  $-S(=0)_2$ - group wherein the remaining two bonds (valences) can be independently substituted. The term "sulfoxido", alone or in combination, means a  $-S(=0)_1$ - group wherein the remaining two bonds (valences) can be independently substituted. The term "sulfonylamide", alone or in combination, means a  $-S(=0)_2$ -N= group wherein the remaining three bonds (valences) can be independently substituted. The term "sulfinamido", alone or in combination, means a  $-S(=0)_1$ N= group wherein the remaining three bonds (valences) can be independently substituted. The term "sulfenamide", alone or in combination, means a -S-N= group wherein the remaining three bonds (valences) can be independently substituted.

The term "alkoxy", alone or in combination, means an alkyl ether radical wherein the term alkyl is

as defined above. Examples of suitable alkyl ether radicals include methoxy, ethoxy, n-propoxy, isopropoxy, n-butoxy, iso-butoxy, sec-butoxy, tert-butoxy and the like. The term "cycloalkyl", alone or in combination, means an alkyl radical which contains from about 3 to about 8 carbon atoms and is cyclic. The term "cycloalkylalkyl" means an alkyl radical as defined above which is substituted by a cycloalkyl radical containing from about 3 to about 8, preferably from about 3 to about 6, carbon atoms. Examples of such cycloalkyl radicals include cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and the like.

The term "aryl", alone or in combination,

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means a phenyl, indenyl or naphthyl radical that 15 optionally carries one or more substituents selected from alkyl, alkoxy, halogen, hydroxy, amino, nitro and the like, such as phenyl, p-tolyl, 4-methoxyphenyl, 4-(tert-butoxy)phenyl, 4-fluorophenyl, 4-chlorophenyl, 4hydroxyphenyl, 1-naphthyl, 2-naphthyl, and the like. The term "aralkyl", alone or in combination, means an 20 alkyl radical as defined above in which one hydrogen atom is replaced by an aryl radical as defined above, such as benzyl, 2-phenylethyl and the like. The term "aralkoxycarbonyl", alone or in combination, means a radical of the formula -C(0)-O-aralkyl in which the 25 term "aralkyl" has the significance given above. An example of an aralkoxycarbonyl radical is benzyloxycarbonyl. The term "aryloxy" means a radical of the formula aryl-O- in which the term aryl has the significance given above. The term "aromatic ring" in 30 combinations such as substituted-aromatic ring

sulfonamide, substituted-aromatic ring sulfinamide or substituted-aromatic ring sulfenamide means aryl or heteroaryl as defined above.

The terms "alkanoyl" or "alkylcarbonyl", alone or in combination, means an acyl radical derived from an alkylcarboxylic acid, examples of which include acetyl, propionyl, butyryl, valeryl, 4-methylvaleryl, and the like. The term "cycloalkylcarbonyl" means an acyl group derived from a monocyclic or bridged 10 cycloalkylcarboxylic acid such as cyclopropanecarbonyl, cyclohexanecarbonyl, adamantanecarbonyl, and the like, or from a benz-fused monocyclic cycloalkylcarboxylic acid which is optionally substituted by, for example, alkanoylamino, such as 1,2,3,4-tetrahydro-2-naphthoyl, 2-acetamido-1,2,3,4-tetrahydro-2-naphthoyl. 15 "aralkanoyl" or "aralkylcarbonyl" mean an acyl radical derived from an aryl-substituted alkylcarboxylic acid such as phenylacetyl, 3-phenylpropionyl (hydrocinnamoyl), 4-phenylbutyryl, (2-naphthyl)acetyl, 4-chlorohydrocinnamoyl, 4-aminohydrocinnamoyl, 4-20 methoxyhydrocinnamoyl and the like.

The terms "aroyl" or "arylcarbonyl" means an acyl radical derived from an aromatic carboxylic acid. Examples of such radicals include aromatic carboxylic acids, an optionally substituted benzoic or naphthoic acid such as benzoyl, 4-chlorobenzoyl, 4-carboxybenzoyl, 4-(benzyloxycarbonyl)benzoyl, 1-naphthoyl, 2-naphthoyl, 6-carboxy-2 naphthoyl, 6-(benzyloxycarbonyl)-2-naphthoyl, 3-benzyloxy-2-naphthoyl, 3-hydroxy-2-naphthoyl, 3-(benzyloxyformamido)-2-naphthoyl, and the like.

The heterocyclyl (heterocyclo) or heterocycloalkyl portion of a heterocyclylcarbonyl, heterocyclyloxycarbonyl, heterocyclylalkoxycarbonyl, or heterocyclyalkyl group or the like is a saturated or partially unsaturated monocyclic, bicyclic or tricyclic heterocycle that contains one or more hetero atoms selected from nitrogen, oxygen and sulphur, which is optionally substituted on one or more carbon atoms by a halogen, alkyl, alkoxy, oxo group, and the like, and/or on a secondary nitrogen atom (i.e., -NH-) by an 10 alkyl, aralkoxycarbonyl, alkanoyl, aryl or arylalkyl or on a tertiary nitrogen atom (i.e. =N-) by oxido and which is attached via a carbon atom. The tertiary nitrogen atom with three substituents can also form a N-oxide (=N(O)-) group. The heteroaryl portion of a 15 heteroaroyl, heteroaryloxycarbonyl, or a heteroaralkoxy carbonyl group or the like is an aromatic monocyclic, bicyclic, or tricyclic heterocycle that contains the hetero atoms and is optionally substituted as defined above with respect to the definition of heterocyclyl. 20 Examples of such heterocyclyl and heteroaryl groups are pyrrolidinyl, piperidinyl, piperazinyl, morpholinyl, thiamorpholinyl, pyrrolyl, imidazolyl (e.g., imidazol 4-yl, 1-benzyloxycarbonyl-imidazol-4-yl, pyrazolyl, pyridyl, pyrazinyl, pyrimidinyl, furyl, 25 tetrahydrofuryl, thienyl, triazolyl, oxazolyl, oxadiazoyl, thiazolyl, thiadiazoyl, indolyl (e.g., 2indolyl, quinolinyl, (e.g., 2-quinolinyl, 3-quinolinyl, 1-oxido-2-quinolinyl), isoquinolinyl (e.g., 1isoquinolinyl, 3-isoquinolinyl), tetrahydroquinolinyl (e.g., 1,2,3,4-tetrahydro-2-quinolyl),

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1,2,3,4-tetrahydroisoquinolinyl (e.g., 1,2,3,4-tetrahydro-1-oxo-isoquinolinyl), quinoxalinyl, ß-carbolinyl, 2-benzofurancarbonyl, benzothiophenyl, 1-,2-,4- or 5-benzimidazolyl, and the like.

5 The term "cycloalkylalkoxycarbonyl" means an acyl group derived from a cycloalkylalkoxycarboxylic acid of the formula cycloalkylalkyl-O-COOH wherein cycloalkylalkyl has the significance given above. The term "aryloxyalkanoyl" means an acyl radical of the formula aryl-O-alkanoyl wherein aryl and alkanoyl have 10 the significance given above. The term "heterocyclyloxycarbonyl" means an acyl group derived from heterocyclyl-O-COOH wherein heterocyclyl is as defined above. The term "heterocyclylalkanoyl" is an acyl radical derived from a heterocyclyl-substituted 15 alkane carboxylic acid wherein heterocyclyl has the significance given above. The term "heterocyclylalkoxycarbonyl" means an acyl radical derived from a heterocyclyl-substituted alkane-O-COOH wherein heterocyclyl has the significance given above. 20 The term "heteroaryloxycarbonyl" means an acyl radical derived from a carboxylic acid represented by heteroaryl-O-COOH wherein heteroaryl has the significance given above.

The term "aminocarbonyl" alone or in combination, means an amino-substituted carbonyl (carbamoyl) group derived from an amino-substituted carboxylic acid wherein the amino group can be a primary, secondary or tertiary amino group containing substituents selected from hydrogen, and alkyl, aryl, aralkyl, cycloalkyl, cycloalkylalkyl radicals and the

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like. The term "aminoalkanoyl" means an acyl group derived from an amino-substituted alkanecarboxylic acid wherein the amino group can be a primary, secondary or tertiary amino group containing substituents independently selected from hydrogen, alkyl, aryl, aralkyl, cycloalkyl, cycloalkylalkyl radicals and the like.

The term "halogen" means fluorine, chlorine, The term "haloalkyl" means an alkyl bromine or iodine. radical having the significance as defined above 10 wherein one or more hydrogens are replaced with a halogen. Examples of such haloalkyl radicals include chloromethyl, 1-bromoethyl, fluoromethyl, difluoromethyl, trifluoromethyl, 1,1,1-trifluoroethyl and the like. The term perfluoroalkyl means an alkyl 15 group wherein each hydrogen has been replaced by a fluorine atom. Examples of such perfluoroalkyl groups, in addition to trifluoromethyl above, are perfluorobutyl, perfluoroisopropyl, perfluorododecyl and perfluorodecyl. 20

### Description of the Preferred Embodiments

In accordance with the present invention, it has been discovered that certain novel substituted-aromatic sulfonamide hydroxamic acid compounds are effective for inhibition of matrix metalloproteases ("MMPs") believed to be associated with uncontrolled or otherwise pathological breakdown of connective tissue. In particular, it has been found that these substituted-aromatic ring sulfonamide hydroxamic acid, substituted-aromatic ring sulfinamide hydroxamic acid

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or substituted-aromatic ring sulfenamide hydroxamic acid compounds are effective for inhibition of collagenase Type III (MMP-13), which is believed to be particularly destructive to tissue if present or 5 generated in abnormal quantities or concentrations. Moreover, it has been discovered that many of these novel sulfur-nitrogen bonded compounds are selective in the inhibition of MMP-13 and/or other MMPs associated with diseased conditions without excessive inhibition 10 of those collagenases essential to normal bodily function such as tissue turnover and repair or other zinc proteases. More particularly, it has been found that many of the substituted-aryl- or substitutedheteroaryl-sulfonamide hydroxamic acids of the invention are selective for MMP-13 with limited or 15 minimal effect on MMP-1.

Set forth in Table 1 to Table 8 inclusive and in Example 1 to Example 15 inclusive are several series 20 of preferred classes of compounds.

TABLE 1

HO 
$$R_3$$
 H  $O$   $O$ 

Example	R <sub>2</sub>	$R_3$	R <sub>4</sub>
4	~ NO	H <sub>3</sub> C CH <sub>3</sub>	Br Br
5	$\sim$ $\sim$ $\sim$ $\sim$ $\sim$	H <sub>3</sub> C CH <sub>3</sub>	СH,3
6		CH <sub>3</sub>	————он
7	$\sim$ N $\sim$ 0	CH <sub>3</sub>	S
8	NO	CH <sub>3</sub>	
9		——CH <sub>3</sub>	СН,
10		CH <sub>3</sub>	— № т
11		—-CH <sub>3</sub>	NH <sub>2</sub>

Table 1 (continued)

	N	0 , 1	
12		N N	СН
13		SSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSSS	ОН
14		H <sub>3</sub> C CH <sub>3</sub>	s
15		——CH <sub>3</sub>	CH3
16	~~~~°	CH <sub>3</sub>	ОН
17	~~~~°	CH <sub>3</sub>	S

5 .

TABLE 2

HO 
$$R_3$$
 H  $R_3$  H  $R_3$  H  $R_3$  H  $R_4$ 

Example 18	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
l i.		l i	$\wedge$
	N 0		
		H <sub>3</sub> C CH <sub>3</sub>	
		3	
19			
		H <sub>3</sub> C CH <sub>3</sub>	CO₂H
20		CH <sub>3</sub>	$\sim$
1	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\		
		CH <sub>3</sub>	\\
21	$\sim$	CH <sub>3</sub>	
1	N O	——	
·		CH₃	, o
22		CH <sub>3</sub>	
<u> </u>	N D		s
		CH <sub>3</sub>	ĺ N ≥
23	/==N	—CH <sub>3</sub>	
1			
		·	
24		——CH <sub>3</sub>	
		0,1,3	
			CO <sub>2</sub> H
			2.
25		——CH <sub>3</sub>	N N
	<i>→</i>		
1			

Table 2 (continued)

26	N I	0 .	
26		NET	
27	N	s	
28	~~~~°	H <sub>3</sub> C CH <sub>3</sub>	
29	~~~	——CH₃	
30	\	CH <sub>3</sub>	
31	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	СН3	S N

Example	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
32		1	
J.	$\sim$ $\sqrt{b}$		
-	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\		
		H <sub>3</sub> C CH <sub>3</sub>	0
33			—CH₃
	N O		
		H <sub>3</sub> C CH <sub>3</sub>	
34		CH <sub>3</sub>	
	N D		
1		CH₃	
1			
35		CH <sub>3</sub>	CH <sub>3</sub>
	N D	—	O_CH3
		CH <sub>3</sub>	
36		CH <sub>3</sub>	
	N 6	/ °	
		CH₃	
	N. N.	CH CH	-0
37	/\\	CH <sub>3</sub>	
			0'
38	/=N	CH <sub>3</sub>	
39	/=N,	——CH <sub>3</sub>	
			(
			S
		<u> </u>	S CI

Table 3 (continued)

40		~ h	
41		S	CH <sub>3</sub>
42		H <sub>3</sub> C CH <sub>3</sub>	
43	~~~~°	——СН <sub>3</sub>	CH
44		СН3	
45	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	CH <sub>3</sub>	CH <sub>3</sub>

Example	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
46	$\sim$ N $\bigcirc$ 0	H <sub>3</sub> C CH <sub>3</sub>	NH H
47	~~~	H <sub>3</sub> C CH <sub>3</sub>	_k
48	~~~	СН3	_K
49	NO	СН3	
50		CH <sub>3</sub>	——NH <sub>2</sub>
51		——CH <sub>3</sub>	N. H
52		CH <sub>3</sub>	
53		——CH <sub>3</sub>	-H-(-)

Table 4 (continued)

54		NH NH	_r()
55		S	H
56		H <sub>3</sub> C CH <sub>3</sub>	N H
57	~~~	CH <sub>3</sub>	) H
58	~~~°	СН3	_N_O
59	~~~°	CH <sub>3</sub>	_N

TABLE 5

HO 
$$R_3$$
 H O O

Example	R²	R <sup>3</sup>	R <sup>4</sup>
4		——CH₃	Br
5		H₃C CH₃	Br
6	~~~	CH <sub>3</sub>	Br
7			→ Br
8		——CH₃	Br
9		H <sub>3</sub> C CH <sub>3</sub>	-Br
10		CH <sub>3</sub>	-Br
11		N N N	Br

Table 5 (continued)

12	——CH₃	—CH₃	Br
13	—CH₃	H <sub>3</sub> C CH <sub>3</sub>	———Br
14	—CH₃	CH <sub>3</sub> CH <sub>3</sub>	————Br
15	—−CH <sub>3</sub>	——————————————————————————————————————	————Br
16	CH <sub>3</sub>	—CH₃	———Br
17	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	<b>─</b> Br
18	CH <sub>3</sub>	CH₃ CH₃	Br
19 /	CH <sub>3</sub>	→ p · · · ·	Br
20		—CH <sub>3</sub>	————Br
21	, in the second	H <sub>3</sub> C CH <sub>3</sub>	-Br
22	, in the second	CH <sub>3</sub>	————Br

Table 5 (continued)

23		~ h	Br
24		CH <sub>3</sub>	CH <sub>3</sub>
25		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
26		CH <sub>3</sub>	CH <sub>3</sub>
27	~~~	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	CH <sub>3</sub>
28		CH₃	CH <sub>3</sub>
29		H₃C CH₃	CH <sub>3</sub>
30		CH₃ —CH₃	CH <sub>3</sub>
31		M N	CH <sub>3</sub>
32	CH <sub>3</sub>	——CH₃	CH <sub>3</sub>
33	—-CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>

Table 5 (continued)

		011	
34	—-CH₃	CH₃	СН3
		CH₃	ĊH₃
35	СН <sub>3</sub>		-CH <sub>3</sub>
			ĊH <sub>3</sub>
36	CH <sub>3</sub>	—CH₃	CH <sub>3</sub>
	CH <sub>3</sub>		CH <sub>3</sub>
37	CH₃		CH <sub>3</sub>
	CH₃	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
38	CH₃	CH₃	CH <sub>3</sub>
	CH₃	—CH₃	CH <sub>3</sub>
39	ÇH₃		CH <sub>3</sub>
	CH₃	→ p · · · ·	CH₃
40	g g	—CH <sub>3</sub>	CH <sub>3</sub>
		<del>"</del> '	CH₃
41	9		
			CH <sub>3</sub>
		H <sub>3</sub> C CH <sub>3</sub>	ĊH₃
42	, j	CH₃	CH₃
		CH₃	CH <sub>3</sub>
43	)	Î L	CH <sub>3</sub>
		✓ Å Å Å	CH <sub>3</sub>
	<u>_</u> 6		, and the second
44	N O	——CH₃	CH <sub>3</sub>
			CH <sub>3</sub>

Table 5 (continued)

45	NO	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
46		CH <sub>3</sub>	CH <sub>3</sub>
47		~ h	CH <sub>3</sub>
48		—-CH₃	CH <sub>3</sub>
49		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
50		CH <sub>3</sub>	CH <sub>3</sub>
51			CH <sub>3</sub>
52	—-CH₃	—CH₃	CH <sub>3</sub>
53	—−CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
54	——CH₃	CH <sub>3</sub>	CH <sub>3</sub>
55	СН <sub>3</sub>	~ h	CH <sub>3</sub>

Table 5 (continued)

56	ÇH₃	—CH <sub>3</sub>	
	7'''3	0113	CH <sub>3</sub>
			CH₃
57	CH₃		
5/	CH₃		CH <sub>3</sub>
	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	ĊH₃
58	ÇH₃	CH <sub>3</sub>	CH <sub>3</sub>
		<u> </u>	
	CH <sub>3</sub>	`—CH₃	ĊH <sub>3</sub>
59	ÇH₃	Î	
		~_N~~	CH₃
,	CH <sub>3</sub>	n	ĆH <sub>3</sub> .
60	Ŷ	-—CH₃	<u></u>
1		-	CH₃
			ĊH₃
	, °		
61	Î		—⟨CH₃
		$\downarrow$	
İ		H <sub>3</sub> C CH <sub>3</sub>	ĊH <sub>3</sub>
62	ρ	CH₃	/=\
			CH <sub>3</sub>
	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	—CH₃	CH₃
	\ <u>\</u>	<b>U</b> .13	
63	0	9	CH <sub>3</sub>
		$\backslash \backslash $	ا کے
ļ	1 1	H	ĊH <sub>3</sub>
		011	
64		—CH₃	
	N O		OH.
65		1	
	N O		<b>—</b> ()—он
66		H <sub>3</sub> C CH <sub>3</sub>	
""	N O		— ()—он
		CH <sub>3</sub>	

Table 5 (continued)

67	NO	~ h	-С-
68	N	——CH₃	————он
69		H <sub>3</sub> C CH <sub>3</sub>	ОН
70		CH <sub>3</sub> —CH <sub>3</sub>	ОН
71			———ОН
72	CH <sub>3</sub>	—CH₃	———он
73	——CH₃	H₃C CH₃	————он
74	CH <sub>3</sub>	CH <sub>3</sub>	-С->-он
75	CH <sub>3</sub>		-С->-он
76	CH <sub>3</sub>	——CH₃	-С-ОН
77	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	————он

Table 5 (continued)

78	ÇH₃	,CH₃	
	CH <sub>3</sub>	CH <sub>3</sub>	— С
79	CH <sub>3</sub>	N N N N N N N N N N N N N N N N N N N	——ОН
80		CH₃	————он
81		H₃C CH₃	———ОН
82		CH₃ —CH₃	————он
83		H N	————ОН
84		CH <sub>3</sub>	N=S
85	- NO	H <sub>3</sub> C CH <sub>3</sub>	S N=
86		CH <sub>3</sub>	S
87		N N N N N N N N N N N N N N N N N N N	S N=

Table 5 (continued)

88		CH <sub>3</sub>	N=S
89		H <sub>3</sub> C CH <sub>3</sub>	S N=
90		CH <sub>3</sub>	S N=
91			S N=
92	——CH₃	—CH₃	
93	——CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	S
94	——CH₃	CH <sub>3</sub>	S
95	——CH₃	M N	S
96	CH <sub>3</sub>	CH <sub>3</sub>	S N=
97	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	S N=
98	CH <sub>3</sub>	CH <sub>3</sub>	S

Table 5 (continued)

99	CU	0 1 1	
·	CH <sub>3</sub>	~ h	S N=
100		CH <sub>3</sub>	S N=
101		H <sub>3</sub> C CH <sub>3</sub>	N=S
102		CH <sub>3</sub>	S N
103			S N=
104		——CH₃	-
105	NO	H <sub>3</sub> C CH <sub>3</sub>	
106		CH <sub>3</sub> —CH <sub>3</sub>	
107		~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	
108		CH <sub>3</sub>	

Table 5 (continued)

109		H <sub>3</sub> C CH <sub>3</sub>	
110		CH <sub>3</sub>	
111			
112	—CH₃	CH <sub>3</sub>	
113	CH₃	H <sub>3</sub> C CH <sub>3</sub>	
114	——CH₃	CH <sub>3</sub> CH <sub>3</sub>	Z Z
115	CH <sub>3</sub>	~ H	
116	CH <sub>3</sub>	CH <sub>3</sub>	
117	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
118	CH <sub>3</sub>	CH₃ CH₃	
119	CH <sub>3</sub>	J. J.	

Table 5 (continued)

120		—−CH <sub>3</sub>	
121		H <sub>3</sub> C CH <sub>3</sub>	
122		CH <sub>3</sub>	
123			
124		CH₃	——H <sub>3</sub> CH₃
125	~~~	H <sub>3</sub> C CH <sub>3</sub>	—————————————————————————————————————
126		CH <sub>3</sub>	——— Н сн₃
127			——Н сн₃
128		CH <sub>3</sub>	
129		H <sub>3</sub> C CH <sub>3</sub>	—— Н сн₃
130		CH <sub>3</sub>	— СН₃

Table 5 (continued)

131	N		——————————————————————————————————————
132	——CH₃	—CH₃	—{_}Н_сн₃
133	——CH₃	H₃C CH₃	- $        -$
134	——CH₃	CH <sub>3</sub> —CH <sub>3</sub>	———H CH₃
135	CH <sub>3</sub>		——H <sub>CH₃</sub>
136	CH <sub>3</sub>	——CH₃	
137	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	——H CH₃
138	CH <sub>3</sub>	CH <sub>3</sub>	H CH₃
139	CH <sub>3</sub>	~ h	H CH <sub>3</sub>
140		—−CH <sub>3</sub>	——H CH₃
141	l i	H <sub>3</sub> C CH <sub>3</sub>	— Д СН₃

Table 5 (continued)

142	Ö	,CH₃	/\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
		· — ( - )	_ <b>√_</b> }_Н <sub>_</sub> сн₃
		CH₃	
140			
143	Ĭ	, i i	— <b>√</b> У–₩_сн₃
	~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	~ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	\_ \_ \
	<u>_</u>		
144		—CH₃	
·	N O		
145		1	NH <sub>2</sub>
	N O		
		H₃C CH₃	NH <sub>2</sub>
146		,CH <sub>3</sub>	0,0
	N O	_	
		CH <sub>3</sub>	NH <sub>2</sub>
147	$\sim$	Ŷ l	
	NO		<del>(</del> ' ')S.'
148		—CH₃	NH <sub>2</sub>
	<b></b>	Ci 13	
1			NH <sub>2</sub>
149	( <u></u>	<u>                                     </u>	
		H <sub>3</sub> C CH <sub>3</sub>	NH <sub>2</sub>
150	/==N	CH₃	0
		—	<del></del>
		CH <sub>3</sub>	NH <sub>2</sub>
151	/=N	Ŷ	0 - S, 0
		, , , , , , , , , , , , , , , , , , ,	NH <sub>2</sub>
152	CH <sub>3</sub>	CH <sub>3</sub>	
			NH <sub>2</sub>

Table 5 (continued)

153	OII I		
153	—CH₃		-
154	—CH₃	H <sub>3</sub> C CH <sub>3</sub>	
	O1 13		
		CH₃	NH <sub>2</sub>
155	——CH₃		
		1 V 1 V V V V V V V V V V V V V V V V V	NH <sub>2</sub>
156	ÇH₃	CH <sub>3</sub>	
			NH <sub>2</sub>
157	CH <sub>3</sub>	1	
-37	) I I 3		
	CH₃	H <sub>3</sub> C CH <sub>3</sub>	NH <sub>2</sub>
158	ÇH₃	CH <sub>3</sub>	
			NH <sub>2</sub>
159	CH₃	CH₃	
133	CH <sub>3</sub>		
	CH <sub>3</sub>	H	NH <sub>2</sub>
160	9	—-СH <sub>3</sub>	
	, , , , , ,		NH <sub>2</sub>
			14.12
161	) 		-\(\)_\_\s\(\)\_\S\(\)
	N N		NH <sub>2</sub>
ļ		H <sub>3</sub> C CH <sub>3</sub>	
162	Î	CH <sub>3</sub>	0.00
	N N		NH <sub>2</sub>
		CH₃	
163	<u> </u>		\$=0
			NH <sub>2</sub>
	\ <u>\</u>		

Table 5 (continued)

164		CH <sub>3</sub>	———СH₃
165	$\sim$ NO	H <sub>3</sub> C CH <sub>3</sub>	——————————————————————————————————————
166		CH <sub>3</sub>	———CH₃
167	$\sim$ N $_{\odot}$	→ H − − − − − − − − − − − − − − − − − −	—————————————————————————————————————
168		CH₃	-CH <sub>3</sub>
169		H <sub>3</sub> C CH <sub>3</sub>	-CH <sub>3</sub>
170		CH <sub>3</sub> CH <sub>3</sub>	——————————————————————————————————————
171		→ H	-CH <sub>3</sub>
172	——CH <sub>3</sub>	——CH₃	——————————————————————————————————————
173	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	—СН3
174	CH <sub>3</sub>	CH <sub>3</sub>	—СН3

Table 5 (continued)

175	——CH <sub>3</sub>	✓ MINING N	(CH <sub>3</sub>
176	CH <sub>3</sub>	CH <sub>3</sub>	-CH <sub>3</sub>
177	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
178	CH <sub>3</sub>	CH <sub>3</sub>	——————————————————————————————————————
179	CH <sub>3</sub>		-CH <sub>3</sub>
180		—−CH <sub>3</sub>	——————————————————————————————————————
181		H <sub>3</sub> C CH <sub>3</sub>	-CH <sub>3</sub>
182		CH <sub>3</sub>	CH <sub>3</sub>
183	, in the second	M N	——————————————————————————————————————
184		——CH₃	$\langle s \rangle$
185		H <sub>3</sub> C CH <sub>3</sub>	s ·

Table 5 (continued)

186	~~\o\_\o\_\o\_	CH <sub>3</sub>	\(\sigma\)
187		~ p	$rac{s}{s}$
188		—CH₃	S
189		H <sub>3</sub> C CH <sub>3</sub>	\s\s\
190		CH <sub>3</sub>	$\searrow^{S}$
191		~ A	√ <sup>S</sup> >
192	—-СН₃	——CH₃	\s\s\
193	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	S
194	—CH₃	CH <sub>3</sub>	S
195	——CH₃	The state of the s	\s\s\
196	CH <sub>3</sub>	──CH <sub>3</sub>	$\sim$

Table 5 (continued)

107		<del></del>	.S.
197	CH₃		$\langle \rangle$
	`CH₃	H <sub>3</sub> C CH <sub>3</sub>	
198	ÇH₃	CH₃	_ \S\
	CH <sub>3</sub>	CH <sub>3</sub>	
199	ÇH₃	Ŷ	_ S
	CH <sub>3</sub>	A	
200	Q	—CH <sub>3</sub>	_ \S\
	N		<u> </u>
	, °		
201	o I		~\s\
			\\ /
		H₃C CH₃	
202	9	CH₃	
1		CH₃	
	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	O113	
203	=0	0 !	~\s\
			\\
1		"	Time and the second
204		CU	
203	$\sim$ N $_{\rm O}$	—CH₃	
205		 	
	N O		
206		H <sub>3</sub> C CH <sub>3</sub>	
206		CH <sub>3</sub>	
	N O		<del> </del>
207		CH <sub>3</sub>	
207	1 / \		
	- N 0	l ~ h	

Table 5 (continued)

208	$\sim$	—−CH <sub>3</sub>	— N
209	N	H <sub>3</sub> C CH <sub>3</sub>	N
210		CH <sub>3</sub>	
211		N N N N N N N N N N N N N N N N N N N	—√_N
212	—CH₃	—CH₃	—⟨□N
213	CH₃	H <sub>3</sub> C CH <sub>3</sub>	
214	——CH₃	CH <sub>3</sub> —CH <sub>3</sub>	—√N
215	—−CH <sub>3</sub>	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	
216	CH <sub>3</sub>	——CH <sub>3</sub>	
217	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
218	CH <sub>3</sub>	CH <sub>3</sub>	

Table 5 (continued)

219	ÇH₃	0	/=\
	CH₃	~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	— ( )i
220		СН <sub>3</sub>	
221		H <sub>3</sub> C CH <sub>3</sub>	
222		CH <sub>3</sub> CH <sub>3</sub>	
223		A A	
224		——CH₃	CH <sub>3</sub>
225		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
226		CH <sub>3</sub>	CH <sub>3</sub>
227			CH <sub>3</sub>
228		CH <sub>3</sub>	CH <sub>3</sub>
229		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>

Table 5 (continued)

230	N T	,CH₃	∕=\ CH₃
	<i>_</i> \		
		CH₃	CH <sub>3</sub>
231	N	ОП3 9	∕=\ CH₃
231			
		~ \h ~ \	CH <sub>3</sub>
232		CU	<b>/</b> ⇒\ βH₃
232	——CH₃	—−CH <sub>3</sub>	
			CH <sub>3</sub>
233	—CH <sub>3</sub>		/ <del>−</del> \ CH <sub>3</sub>
	·	H <sub>3</sub> C CH <sub>3</sub>	°CH₃
234	CH₃	CH₃	/ <del>−</del> \ CH₃
		$\overline{}$	CH <sub>3</sub>
		CH₃	
235	—-CH <sub>3</sub>		CH <sub>3</sub>
		l ~ l	CH <sub>3</sub>
236	ÇH₃	—CH₃	∠ CH <sub>3</sub>
		_	
	CH <sub>3</sub>		CH <sub>3</sub>
237	ÇH₃		∠ CH <sub>3</sub>
			CH <sub>3</sub>
	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
238	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>
			CH <sub>3</sub>
	CH <sub>3</sub>	CH₃	i
239	CH <sub>3</sub>		CH <sub>3</sub>
		1 ~ h ~	CH <sub>3</sub>
240	CH <sub>3</sub>		∠ CH <sub>3</sub>
430		—-CH₃	
1			CH <sub>3</sub>
	1 1	1	

Table 5 (continued)

241		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
242		CH <sub>3</sub>	CH <sub>3</sub>
243			CH <sub>3</sub>
244		—CH₃	—————————————————————————————————————
245	NO	H <sub>3</sub> C CH <sub>3</sub>	—————————————————————————————————————
246		CH <sub>3</sub> CH <sub>3</sub>	—————————————————————————————————————
247		- H	—Сто
248		—−CH <sub>3</sub>	——С СН₃
249		H <sub>3</sub> C CH <sub>3</sub>	—————————————————————————————————————
250		CH <sub>3</sub> CH <sub>3</sub>	-\(\)-\(\)-\(\)CH <sub>3</sub>
251		A A	—————————————————————————————————————

Table 5 (continued)

	0.1		
252	—CH₃	—CH₃	-√СН₃
253	—CH₃	$\rightarrow$	-CH3
254	. —CH₃	H <sub>3</sub> C CH <sub>3</sub>	—————————————————————————————————————
		CH₃	
255	—CH₃		CH3
256	CH <sub>3</sub>	—CH₃	
257	CH <sub>3</sub>	H₃C CH₃	— СН₃
258	CH <sub>3</sub>	CH <sub>3</sub>	——С>—О, СН₃
259	CH₃ CH₃	N N N	—————————————————————————————————————
260		CH₃	———Сн₃
261		H <sub>3</sub> C CH <sub>3</sub>	—————————————————————————————————————
262		CH <sub>3</sub> CH <sub>3</sub>	—————————————————————————————————————

Table 5 (continued)

263		~	——С—СН3
264		—-СH <sub>3</sub>	о-сн₃
265		H <sub>3</sub> C CH <sub>3</sub>	о-сн₃
266		CH <sub>3</sub>	о-сн₃
267	$\sim$ N $\sim$ 0	NH N	О О О СН₃
268		CH₃	о−сн₃
269		H <sub>3</sub> C CH <sub>3</sub>	O-CH <sub>3</sub>
270		CH <sub>3</sub>	—————————————————————————————————————
271		N N N	0-CH <sub>3</sub>
272	——CH <sub>3</sub>	CH₃	-CH <sub>3</sub>
273	——CH₃	H <sub>3</sub> C CH <sub>3</sub>	—————————————————————————————————————

Table 5 (continued)

274	——CH <sub>3</sub>	CH <sub>3</sub>	————о-сн <sub>3</sub>
275	—CH₃	N N	O-CH <sub>3</sub>
276	CH <sub>3</sub>	—-CH₃	O-CH <sub>3</sub>
277	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	О−СН₃
278	CH <sub>3</sub>	CH <sub>3</sub> CH <sub>3</sub>	O-CH <sub>3</sub>
<b>27</b> 9	CH <sub>3</sub>	NH N	O-CH <sub>3</sub>
280		——CH₃	О-сн₃
281		H <sub>3</sub> C CH <sub>3</sub>	о-сн₃
282		CH₃ CH₃	O-CH <sub>3</sub>
283		The state of the s	о-сн₃
284		CH <sub>3</sub>	-CH <sub>3</sub>

Table 5 (continued)

286  286  CH <sub>3</sub> 290  CH <sub>3</sub> C			,	<del></del>
286  CH <sub>3</sub> 288  CH <sub>3</sub>	285	$\sim$ NO		
286  N CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 288  288  N CH <sub>3</sub>			H <sub>3</sub> C CH <sub>3</sub>	ĊH <sub>3</sub>
288  289  N  CH <sub>3</sub>	286		CH <sub>3</sub>	
288	287		0 1	
289  N  H <sub>3</sub> C  CH <sub>3</sub> CH <sub>3</sub> 290  N  CH <sub>3</sub>		N O	, N	
289  N  H <sub>3</sub> C  CH <sub>3</sub> CH <sub>3</sub> 290  N  CH <sub>3</sub>	288	/N	CH <sub>3</sub>	/=\
289  N  H <sub>3</sub> C  CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 291  N  CH <sub>3</sub> 294  CH <sub>3</sub>		\	_	
290  N CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 291  N CH <sub>3</sub> CH <sub></sub>	1			CH₃
290  CH <sub>3</sub>	289	/==N		
291  291  CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 291  CH <sub>3</sub> 294  CH <sub>3</sub>				-\\ \rangle o\ch3
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			H <sub>2</sub> C CH <sub>2</sub>	CH <sub>2</sub>
291	290			
291	250			
291 — CH <sub>3</sub> —				
292 — CH <sub>3</sub> — CH <sub>3</sub> — O_CH <sub>3</sub> CH <sub>3</sub> 293 — CH <sub>3</sub> — O_CH <sub>3</sub> CH <sub>3</sub> CH <sub>3</sub> 294 — CH <sub>3</sub> — O_CH <sub>3</sub> CH <sub>3</sub>				СН3
292 — CH <sub>3</sub> —	291	/ <del></del> N	l i	- CH2
292 — CH <sub>3</sub> —				
293 — CH <sub>3</sub> 294 — CH <sub>3</sub>				CH₃
293 — CH <sub>3</sub> 294 — CH <sub>3</sub>	292	——CH <sub>3</sub>	—CH₃	
293 — CH <sub>3</sub> 294 — CH <sub>3</sub>				
294 — CH <sub>3</sub> —				ĊH₃
294 — CH <sub>3</sub> —	293	CH <sub>3</sub>		
294 — CH <sub>3</sub> — CH <sub>3</sub> — O CH <sub>3</sub> — CH <sub>3</sub>				
294 — CH <sub>3</sub> — CH <sub>3</sub> — O CH <sub>3</sub> — CH <sub>3</sub>			H <sub>3</sub> C CH <sub>3</sub>	ĊH₃
295 — CH <sub>3</sub> — CH <sub>3</sub> — CH <sub>3</sub> — CH <sub>3</sub>	294	CH <sub>3</sub>	CH <sub>3</sub>	
295 — CH <sub>3</sub>			l —	CH <sub>3</sub>
295 — CH <sub>3</sub>			CH <sub>3</sub>	ĊH₃
	295	——CH <sub>3</sub>		
H CH <sub>3</sub>				-CH3
			Н	ĊH₃

Table 5 (continued)

			<del></del>
296	CH <sub>3</sub>	CH <sub>3</sub>	O CH <sub>3</sub>
297	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	O CH <sub>3</sub>
298	CH <sub>3</sub>	CH <sub>3</sub>	
299	CH <sub>3</sub>		O CH <sub>3</sub>
300		——CH₃	O CH <sub>3</sub>
301		H <sub>3</sub> C CH <sub>3</sub>	O CH <sub>3</sub>
302		CH <sub>3</sub> CH <sub>3</sub>	-CH <sub>3</sub>
303		y j	O_CH <sub>3</sub>

Table 6

HO 
$$R_3$$
 H O O

5

Example	R <sup>2</sup>	R <sup>3</sup>	R*
304		——CH₃	
305	~~~	H <sub>3</sub> C CH <sub>3</sub>	
306		CH <sub>3</sub>	
307		THE CONTRACT OF THE CONTRACT O	
308		——CH₃	
309		H <sub>3</sub> C CH <sub>3</sub>	
310		CH <sub>3</sub>	
311		N N	

Table 6 (continued)

312	CH <sub>3</sub>	CH <sub>3</sub>	
313	—−CH <sub>3</sub>		
		H₃C CH₃	
314	—CH₃	CH₃	
		CH₃	
315	CH₃		
316	CH <sub>3</sub>	CH <sub>3</sub>	
	CH₃		
317	CH₃		
	CH₃	H <sub>3</sub> C CH <sub>3</sub>	
318	CH₃	CH₃	
	CH <sub>3</sub>	—CH₃	
319	CH <sub>3</sub>	> >> >	
	CH <sub>3</sub>	, in	
320	l l	—CH₃	
321			
		H <sub>3</sub> C CH <sub>3</sub>	
322	l , l	CH <sub>3</sub>	
		CH₃	

Table 6 (continued)

323		Î İ	
		→ H	
324		—CH₃	CO <sub>2</sub> H
325		H <sub>3</sub> C CH <sub>3</sub>	CO <sub>2</sub> H
326		CH <sub>3</sub> —CH <sub>3</sub>	CO₂H
327			CO₂H
328		—−CH <sub>3</sub>	CO <sub>2</sub> H
329		H <sub>3</sub> C CH <sub>3</sub>	CO <sub>2</sub> H
330		CH <sub>3</sub>	CO <sub>2</sub> H
331		The state of the s	CO <sub>2</sub> H
332	—−CH <sub>3</sub>	——CH₃	CO <sub>2</sub> H
333	——СН <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CO <sub>2</sub> H

Table 6 (continued)

334	CH <sub>3</sub>	CH <sub>3</sub>	CO <sub>2</sub> H
335	—−CH <sub>3</sub>	~ H	CO <sub>2</sub> H
336	CH <sub>3</sub>	—−CH <sub>3</sub>	CO <sub>2</sub> H
337	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CO <sub>2</sub> H
338	CH <sub>3</sub>	CH <sub>3</sub> —CH <sub>3</sub>	CO <sub>2</sub> H
339	CH <sub>3</sub>	H - Z	CO <sub>2</sub> H
340		——CH <sub>3</sub>	CO <sub>2</sub> H
341		H <sub>3</sub> C CH <sub>3</sub>	CO <sub>2</sub> H
342		CH <sub>3</sub>	CO <sub>2</sub> H
343		H N	CO <sub>2</sub> H
344		CH <sub>3</sub>	

Table 6 (continued)

345			
		H <sub>3</sub> C CH <sub>3</sub>	
346		CH <sub>3</sub>	
347	~~~		
348		—CH₃	
349		H <sub>3</sub> C CH <sub>3</sub>	
350		CH <sub>3</sub>	
351		l	
352	——CH₃	—CH₃	
353	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
354	—-СH <sub>3</sub>	CH <sub>3</sub>	
355	——CH₃	The state of the s	

Table 6 (continued)

356	CH <sub>3</sub>	—−CH <sub>3</sub>	
357	CH <sub>3</sub>	H₃C CH₃	
358	CH₃	CH <sub>3</sub>	
359	CH <sub>3</sub>		
360		—CH₃	
361		H₃C CH₃	
362		CH <sub>3</sub> —CH <sub>3</sub>	
363		H	
364	-NO	CH <sub>3</sub>	O CH <sub>3</sub>
365		H <sub>3</sub> C CH <sub>3</sub>	O CH <sub>3</sub>

Table 6 (continued)

366		CH₃	
	NO	CH <sub>3</sub>	O CH <sub>3</sub>
367			O CH3
368	N	CH <sub>3</sub>	O CH <sub>3</sub>
369		H <sub>3</sub> C CH <sub>3</sub>	O CH <sub>3</sub>
370		CH <sub>3</sub> CH <sub>3</sub>	O-CH <sub>3</sub>
371			O CH <sub>3</sub>
372	——CH₃	——CH₃	O CH <sub>3</sub>
373	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	O CH <sub>3</sub>
374	. ——CH <sub>3</sub>	CH <sub>3</sub>	O CH <sub>3</sub>

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Table 6 (continued)

375	СН <sub>3</sub>	0 1	
	O1 13		
		H	✓ o
			CH <sub>3</sub>
376	CH₃	—CH₃	
	<b>\</b>		
	CH₃		CH <sub>3</sub>
377	CH <sub>3</sub>		
	CH <sub>3</sub>	H₃C CH₃	V 10 CH₃
378	CH₃	CH₃	
	`CH₃	CH₃	V `O CH₃
379	ÇH₃	0	
	`CH₃		CH <sub>3</sub>
380	O II	—CH₃	
	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\		
1			°O CH₃
381	9		
		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
382	9	CH₃	Ong Ong
		CH₃	CH <sub>3</sub>
383	0	Q 1	CH <sub>3</sub>
		H	
L			CH <sub>3</sub>

Table 6 (continued)

384	-\ _N 0	——CH <sub>3</sub>	
			N=√S
385			
386		H <sub>3</sub> C CH <sub>3</sub>	N=/°
386		CH₃	S
387		CH <sub>3</sub>	N=/
	$\left  \begin{array}{c} \\ \\ \end{array} \right $	N N N N N N N N N N N N N N N N N N N	N N N
388	N	—CH₃	
			N=\^
389	/=N		
		H <sub>3</sub> C CH <sub>3</sub>	N=\^
390	N	CH₃	
		CH <sub>3.</sub>	N=\N
391	N		
		) H	N=/S
392	CH <sub>3</sub>	—−CH <sub>3</sub>	
			N=√ <sup>5</sup>
393	CH₃		
		H <sub>3</sub> C CH <sub>3</sub>	N=/3
394	—-CH₃	CH <sub>3</sub>	
		CH₃	N=/3

Table 6 (continued)

395	CH₃		N = S
396	CH₃ CH₃	—CH₃	N⇒S
397	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	N S
398	CH <sub>3</sub>	CH <sub>3</sub>	N N
399	CH <sub>3</sub>		N N N N N N N N N N N N N N N N N N N
400		—CH₃	N N
401		H₃C CH₃	Z Z
402		CH <sub>3</sub>	N N N N N N N N N N N N N N N N N N N
403		H N	N=S
404	NO	—−CH <sub>3</sub>	N

Table 6 (continued)

405	NO	H₃C CH₃	N
406	NO	CH <sub>3</sub>	N
407		M N N N N N N N N N N N N N N N N N N N	N
408		——СН <sub>3</sub>	
409		H <sub>3</sub> C CH <sub>3</sub>	
410		CH <sub>3</sub> CH <sub>3</sub>	
411		The state of the s	
412	—CH₃	—−CH <sub>3</sub>	N N
413	—CH₃	H <sub>3</sub> C CH <sub>3</sub>	N
414	CH <sub>3</sub>	CH <sub>3</sub>	Z Z
415	CH <sub>3</sub>	A P	N

Table 6 (continued)

416	ÇH₃	CH <sub>3</sub>	^ ^
	CH₃	-7.5	N
417	CH <sub>3</sub>		
	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	N N
418	CH <sub>3</sub>	CH <sub>3</sub>	
	CH <sub>3</sub>	CH <sub>3</sub>	N N
419	CH₃		
	CH <sub>3</sub>		Ň.
420		——CH₃	
			Ň
421	.		
		H₃C CH₃	Ų,
422		CH₃	
		—CH₃	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
423	0	0 _2	
		A TO THE STATE OF	N N
424		—CH <sub>3</sub>	~
	- N O		NH
425			
		H <sub>3</sub> C CH <sub>3</sub>	VNH NH

Table 6 (continued)

426	$\sim$ NO	CH <sub>3</sub>	NH
427		~ h	NH
428	N	CH <sub>3</sub>	NH
429		H <sub>3</sub> C CH <sub>3</sub>	NH
430		CH <sub>3</sub> —CH <sub>3</sub>	NH
431		N N	NH
432	——CH₃	—CH₃	NH
433	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	NH
434	—-CH₃	CH₃ —CH₃	NH
435	——CH <sub>3</sub>	H N	NH
436	CH <sub>3</sub>	CH <sub>3</sub>	NH

Table 6 (continued)

437	ÇH₃	ı	
	, , ,	·	-Y
	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	\_\NH
438	ÇH₃	CH₃	~
1		$\overline{}$	NH
420	CH <sub>3</sub>	CH₃	<u> </u>
439	CH₃		
	CH₃	→ jä, → /	Му
440	O II	—-CH <sub>3</sub>	^
	N N		NH
			<b>~</b>
441	0		<b>△</b>
	N N		NH NH
		H <sub>3</sub> C CH <sub>3</sub>	
442	Î	CH₃	~~
	✓ " į		NH
		CH₃	~
443	Î	0	~~
	, i	N N	NH
			~
444		CH₃	
	\ \\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\		VN-CH₃
445		,	CH <sub>3</sub>
333	N O		
		H <sub>3</sub> C CH <sub>3</sub>	N-CH <sub>3</sub>
446		CH <sub>3</sub>	
	N D	—	N.
		CH <sub>3</sub>	N CH <sub>3</sub>

Table 6 (continued)

447	NO	~ h	N <sub>CH3</sub>
448		CH <sub>3</sub>	N-CH <sub>3</sub>
449	N	H <sub>3</sub> C CH <sub>3</sub>	N-CH <sub>3</sub>
450		CH <sub>3</sub>	N-CH <sub>3</sub>
451		ZI	N <sub>CH3</sub>
452	—−CH <sub>3</sub>	—CH₃	N <sub>CH3</sub>
453	—CH₃	H₃C CH₃	N <sub>CH3</sub>
454	CH <sub>3</sub>	CH₃ CH₃	N, CH3
455	CH <sub>3</sub>	H N	N <sub>CH3</sub>
456	CH <sub>3</sub>	—−CH <sub>3</sub>	N-CH <sub>3</sub>
457	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	N <sub>CH3</sub>

Table 6 (continued)

458	ÇH₃	,CH₃	
130	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\		$/\gamma$
		211	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
	CH <sub>3</sub>	CH₃	N-CH <sub>3</sub>
459	ÇH₃	ĵ l	
			l N
	CH₃	.,	N-CH <sub>3</sub>
460	Î	CH <sub>3</sub>	
- v	, N		VN CH₃
	0		
461			
	, , , ,		VN-CH.
,		H <sub>3</sub> C CH <sub>3</sub>	VN_CH <sub>3</sub>
462	9	CH₃	<b>✓</b>
		<b>─</b> <	
		<mark>`</mark> —СН₃	VN-CH₃
463			
463			
		\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	N <sub>CH3</sub>
			CH <sub>3</sub>
464		—CH₃	ÇH₃
	N O		
			CH <sub>3</sub>
465			ÇH₃
	- N 0		_ ∧ Ň
		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
466		CH₃	ÇH₃
	N O	—	
		CH <sub>3</sub>	CH <sub>3</sub>
467		0 1	ÇH₃
	N O		I A
		H	CH <sub>3</sub>
468	N	CH <sub>3</sub>	ÇH₃
			N CH <sub>3</sub>
			0113

Table 6 (continued)

	· · · · · · · · · · · · · · · · · · ·		
469		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
470	$\sim$	CH <sub>3</sub>	CH <sub>3</sub>
471			ÇH <sub>3</sub> Λ. Ν. CH <sub>3</sub>
472	—CH₃	—−CH <sub>3</sub>	CH <sub>3</sub> N CH <sub>3</sub>
473	CH₃	H₃C CH₃	CH <sub>3</sub>
474	——CH₃	CH <sub>3</sub> CH <sub>3</sub>	ÇH <sub>3</sub> N CH <sub>3</sub>
\ <b>475</b>	CH <sub>3</sub>	H Z	CH <sub>3</sub>
476	CH <sub>3</sub>	—−CH <sub>3</sub>	CH <sub>3</sub>
477	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
478	CH <sub>3</sub>	CH <sub>3</sub>	ÇH₃ ✓V <sub>CH3</sub>
479	CH <sub>3</sub>	N N	ÇH <sub>3</sub>

Table 6 (continued)

480	0 1	——CH <sub>3</sub>	ÇH₃
		On 13	^N_CH₃
481		H <sub>3</sub> C CH <sub>3</sub>	ÇH <sub>3</sub> N CH <sub>3</sub>
482		CH <sub>3</sub> CH <sub>3</sub>	CH <sub>3</sub>
483			CH <sub>3</sub> N CH <sub>3</sub>
484	~~~	—-CH₃	OH
485		H <sub>3</sub> C CH <sub>3</sub>	ОН
486	NO	CH <sub>3</sub>	ОН
487		L L	ОН
488		CH <sub>3</sub>	ОН
489		H <sub>3</sub> C CH <sub>3</sub>	ОН

Table 6 (continued)

490	N	CH <sub>3</sub>	
		CH <sub>3</sub>	ОН
491		→ A	ОН
492	——CH₃	—CH₃	OH
493	—CH₃	H₃C CH₃	ОН
494	——CH₃	CH <sub>3</sub>	ОН
495	——CH₃		ОН
496	CH <sub>3</sub>	——CH₃	→ OH
497	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	Ŏ Ŏ
498	CH <sub>3</sub>	CH <sub>3</sub>	OH OH
499	CH <sub>3</sub>	N N N N N N N N N N N N N N N N N N N	OH
500	i	CH <sub>3</sub>	ОН

Table 6 (continued)

501	0	- · · · · · · · · · · · · · · · · · · ·	^^
		H <sub>3</sub> C CH <sub>3</sub>	ОН
502	9	CH₃	
		$\longrightarrow$	
		CH₃	ОН
503			
	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\		OH OH
504	_N 0	—−CH <sub>3</sub>	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
		Ť	
505			OH
506		H <sub>3</sub> C CH <sub>3</sub>	OH
300	N O		
		CH₃	
507			HO
		l ~ t	
508	/==N	—-CH <sub>3</sub>	OH
F 00	- N		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
509			OH
		H <sub>3</sub> C CH <sub>3</sub>	
510	/=N	CH₃	OH
<u> </u>		CH₃	~ 011
511	N		OH
L		<u> </u>	<u> </u>

Table 6 (continued)

512	—CH₃	——CH <sub>3</sub>	OH
513	—−CH <sub>3</sub>		OH
		H <sub>3</sub> C CH <sub>3</sub>	
514	—СH <sub>3</sub>	CH₃	OH
		CH₃	
515	CH <sub>3</sub>	) N	OH
		l ~ t	
516	CH₃ /	—CH₃	OH
	CH₃		
517	ÇH₃		OH
	CH₃	u.c	
518	ÇH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	OH
519	CH₃ ÇH₃	CH₃	OH
		\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	
520	CH₃		011
520		—−CH <sub>3</sub>	OH
521	)		OH
		H <sub>3</sub> C CH <sub>3</sub>	
522		CH <sub>3</sub>	ОН
J.2.			
	N	CH₃	

Table 6 (continued)

523	9	0	OH
		N N N	
524		CH₃	O-CH <sub>3</sub>
525		H₃C CH₃	О-СН₃
526		CH <sub>3</sub>	0-CH <sub>3</sub>
527			О-СН3
528		——CH₃	0-CH <sub>3</sub>
529		H <sub>3</sub> C CH <sub>3</sub>	O-CH <sub>3</sub>
530		CH <sub>3</sub>	O-CH <sub>3</sub>
531		**************************************	O-CH <sub>3</sub>
532	CH <sub>3</sub>	CH₃	O-CH <sub>3</sub>
533	—CH₃	H <sub>3</sub> C CH <sub>3</sub>	O-CH <sub>3</sub>

Table 6 (continued)

534	—CH₃	CH <sub>3</sub>	O-CH <sub>3</sub>
		CH₃	
535	CH <sub>3</sub>		O-CH <sub>3</sub>
536	CH <sub>3</sub>	—CH₃	0-сн <sub>3</sub>
537	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	O-CH <sub>3</sub>
538	CH₃ CH₃	CH <sub>3</sub>	O-CH <sub>3</sub>
539	CH <sub>3</sub>		О-СН3
540		—−CH <sub>3</sub>	O-CH <sub>3</sub>
541		H <sub>3</sub> C CH <sub>3</sub>	O-CH <sub>3</sub>
542		CH <sub>3</sub> CH <sub>3</sub>	O-CH <sub>3</sub>
543		The state of the s	O-CH <sub>3</sub>
544	NO	—−CH <sub>3</sub>	

Table 6 (continued)

545		1	ш
	NO	H <sub>3</sub> C CH <sub>3</sub>	
546		CH <sub>3</sub>	
547		The state of the s	
548		CH₃	
549		H₃C CH₃	
550		CH <sub>3</sub>	
551			
552	——CH₃	——CH₃	
553	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
554	—CH₃	CH <sub>3</sub>	
555	——CH₃	H N	

Table 6 (continued)

556	ÇН₃	CH <sub>3</sub>	L
330	Ch3	Спз	~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
	CH <sub>3</sub>		$\smile$
557	CH₃		
	CH₃	H₃C CH₃	
558	CH₃	CH <sub>3</sub>	<b>✓ H</b>
	CH <sub>3</sub>	CH <sub>3</sub>	
559	CH₃ /		/\\I
	CH <sub>3</sub>	) d	igcup
560	0	——CH <sub>3</sub>	ZI
	N		
561	9	l	<b>↑</b> #
	N N		
		H <sub>3</sub> C CH <sub>3</sub>	)
562	Î	CH₃	\   
		CH <sub>3</sub>	
	Ö	5113	
563			
		H	

Table 7

HO 
$$R_3$$
 H O O

5

Example	R <sup>2</sup>	R <sup>3</sup>	R*
564	$\sim$ N $\sim$ 0	СH <sub>3</sub>	
565		H <sub>3</sub> C CH <sub>3</sub>	T)
566		CH <sub>3</sub>	
567			
568		——CH₃	
569		H <sub>3</sub> C CH <sub>3</sub>	
570		CH₃ —CH₃	
571	N	H N	T)
572	—CH₃	CH <sub>3</sub>	T)

Table 7 (continued)

573	——CH₃	H <sub>3</sub> C CH <sub>3</sub>	
574	——CH₃	CH <sub>3</sub>	T)
575	—CH₃	V <sub>P</sub>	73
576	CH <sub>3</sub>	—CH₃	
577	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	T)
578	CH <sub>3</sub>	CH <sub>3</sub>	
579 —CH	CH <sub>3</sub>	N N	
580		—CH₃	
581		H₃C CH₃	T)
582		CH <sub>3</sub> —CH <sub>3</sub>	T)
583		The state of the s	

Table 7 (continued)

584	$\sim$ N $_{\circ}$	CH <sub>3</sub>	−CH <sub>3</sub>
585			-CH <sub>3</sub>
586		H <sub>3</sub> C CH <sub>3</sub>	—CH <sub>3</sub>
587		CH <sub>3</sub>	-CH <sub>3</sub>
588		CH <sub>3</sub>	—СH <sub>3</sub>
589		н <sub>3</sub> С Сн <sub>3</sub>	—CH₃
590		CH <sub>3</sub> —CH <sub>3</sub>	—CH₃
591		N N N N N N N N N N N N N N N N N N N	—CH₃
592	——CH <sub>3</sub>	CH <sub>3</sub>	−CH <sub>3</sub>
593	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	—CH₃
594	CH <sub>3</sub>	CH <sub>3</sub>	—CH <sub>3</sub>
595	——СH <sub>3</sub>	~ h	—CH₃

Table 7 (continued)

596	ÇH₃	—-CH <sub>3</sub>	-CH <sub>3</sub>
		J. 3	3
	CH <sub>3</sub>		
597	ÇH <sub>3</sub>		—CH₃
	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
598	CH₃	CH₃	—CH₃
	Cu	CH₃	
599	CH <sub>3</sub>	0 1	—СH <sub>3</sub>
		V <sub>N</sub> V <sub>N</sub>	3.13
	CH <sub>3</sub>	H	
600	0	——CH <sub>3</sub>	—CH₃
	, , ,		
601	9		—CH <sub>3</sub>
	<b>✓ / / / / / / /</b>		
		H₃C CH₃	
602	0	CH₃	—СН <sub>3</sub>
	, , ,		
		CH₃	
603	0	0	−CH <sub>3</sub>
	, v	l ~~ l	
604		—CH₃	
	N O		
605			
	N O		
		H <sub>3</sub> C CH <sub>3</sub>	~~
606		CH <sub>3</sub>	
		CH <sub>3</sub>	

Table 7 (continued)

	·		
607		~ H	
608		—−CH <sub>3</sub>	
609		H₃C CH₃	
610		CH <sub>3</sub>	
611			
612	—CH₃	CH₃	
613	CH <sub>3</sub>	H₃C CH₃	
614	—-CH₃	CH₃ —CH₃	
615	——CH₃	HZ - Z	
616	CH <sub>3</sub>	CH <sub>3</sub>	
617	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	

Table 7 (continued)

618	ÇH₃	CH <sub>3</sub>	
	CH <sub>3</sub>	—CH₃	
619	CH <sub>3</sub>		
	CH <sub>3</sub>	√µ√,"	
620		——CH₃	
621			
		H <sub>3</sub> C CH <sub>3</sub>	
622	, Î	CH₃	
		СН₃	
623	, Î		
		, Å, Å,	
624		CH <sub>3</sub>	CH <sub>3</sub>
			VCH₃
625	-N 0		CH₃ .O.
		H <sub>3</sub> C CH <sub>3</sub>	VCH₃
626	-N 0	CH <sub>3</sub>	CH₃
		—CH₃	O, CH3
627	-N 0		CH₃
		, H, 0, /	O CH <sub>3</sub>

Table 7 (continued)

628	/==N	—CH <sub>3</sub>	ÇH₃
		-7.3	O, CH3
629		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>
630		CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>
631			CH <sub>3</sub> O CH <sub>3</sub>
632	CH <sub>3</sub>	——CH₃	ÇH <sub>3</sub> O CH <sub>3</sub>
633	—−CH <sub>3</sub>	н <sub>3</sub> С СН <sub>3</sub>	CH <sub>3</sub> OCH <sub>3</sub>
634	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>
635	——CH <sub>3</sub>	N N N	CH <sub>3</sub> O,CH <sub>3</sub>
636	CH <sub>3</sub>	—CH₃	CH <sub>3</sub> O CH <sub>3</sub>
637	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>
638	CH₃ CH₃	CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>

Table 7 (continued)

639	CH <sub>3</sub>		CH <sub>3</sub> O CH <sub>3</sub>
640		CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>
641		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>
642		CH <sub>3</sub> CH <sub>3</sub>	CH <sub>3</sub> O CH <sub>3</sub>
643			CH <sub>3</sub> O CH <sub>3</sub>
644		-—CH₃	
645	N O	H <sub>3</sub> C CH <sub>3</sub>	
646		CH <sub>3</sub>	
647		N N	
648		CH <sub>3</sub>	
649		H <sub>3</sub> C CH <sub>3</sub>	

Table 7 (continued)

650	N	CH₃	
		CH₃	
651		~ A	
652	CH₃	CH <sub>3</sub>	
653	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
654	——CH₃	CH <sub>3</sub>	
655	——CH₃		
656	CH <sub>3</sub>	—CH₃	
657	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
658	CH <sub>3</sub>	CH <sub>3</sub>	
659	CH <sub>3</sub>	~ H	
660		CH <sub>3</sub>	

Table 7 (continued)

	***		
661		H <sub>3</sub> C CH <sub>3</sub>	
662		CH <sub>3</sub> CH <sub>3</sub>	
663		· NH NH NH NH NH NH NH NH NH NH NH NH NH	
664		——CH₃	SCI
665		H₃C CH₃	
666		CH <sub>3</sub> —CH <sub>3</sub>	SCI
667			SCI
668		——CH₃	SCI
669		н₃с сн₃	-\s\_c_i
670		CH <sub>3</sub>	-\(\s\)_CI
671		N N	-\s\_c_i

Table 7 (continued)

672			
	CH₃	—-CH₃	S CI
673	——CH₃	H <sub>3</sub> C CH <sub>3</sub>	SCI
674	CH <sub>3</sub>	CH <sub>3</sub>	SCI
675	——CH₃	~ A	- S CI
676	CH <sub>3</sub>	—CH₃	SCI
677	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	- S CI
678	CH <sub>3</sub>	CH <sub>3</sub>	SCI
679	CH <sub>3</sub>	H N	SCI
680		—CH₃	SCI
681		H <sub>3</sub> C CH <sub>3</sub>	SCI
682		CH <sub>3</sub> CH <sub>3</sub>	SCI

Table 7 (continued)

683	0 1		
		~~ h	SCI
684	$\sim$ N O	——CH₃	CH <sub>3</sub>
685		H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
686	$\sim$ N $\bigcirc$ O	CH <sub>3</sub> —CH <sub>3</sub>	CH <sub>3</sub>
687		~ H	CH <sub>3</sub>
688		—CH₃	CH₃
689		H₃C CH₃	CH₃
690		CH <sub>3</sub> —CH <sub>3</sub>	CH₃
691		N N	CH₃
692	—CH₃	——CH₃	CH <sub>3</sub>
693	——CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	CH <sub>3</sub>
694	—CH₃	CH <sub>3</sub>	CH₃

Table 7 (continued)

695	CH <sub>3</sub>	~	CH <sub>3</sub>
696	CH <sub>3</sub>	— <sup>™</sup> CH <sub>3</sub>	CH <sub>3</sub>
697	CH <sub>3</sub>		CH <sub>3</sub>
698	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
036	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>
699	CH <sub>3</sub>		CH <sub>3</sub>
700		CH₃	CH <sub>3</sub>
701		H₃C CH₃	CH₃
702		CH <sub>3</sub> —CH <sub>3</sub>	CH₃
703		N N	CH₃
704		—CH₃	<del>-</del>
705		H <sub>3</sub> C CH <sub>3</sub>	

Table 7 (continued)

706	$\sim$ N $_{\odot}$	CH <sub>3</sub>	<b>─</b>
707	~~~	~ h	<b>−</b> √¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬¬
708		——CH <sub>3</sub>	$-\sqrt{}$
709		H <sub>3</sub> C CH <sub>3</sub>	<b>→</b>
710		CH <sub>3</sub> —CH <sub>3</sub>	<b>−</b> ⟨ <b>&gt;</b> −⟨
711			<b>─</b>
712	—CH₃	—CH₃	
713	CH <sub>3</sub>	н₃с сн₃	<b>─</b>
714	CH <sub>3</sub>	CH <sub>3</sub>	
715	——CH₃	M N	<b>—</b>
716	CH <sub>3</sub>	—CH₃	<b>─</b>

Table 7 (continued)

010			
717	CH <sub>3</sub>	H₃C CH₃	-
718	CH <sub>3</sub>	CH <sub>3</sub>	<b>→</b>
719	CH <sub>3</sub>	) ZI Z	<del>-</del>
720		—CH₃	
721		H₃C CH₃	<del>-</del>
722		CH <sub>3</sub> CH <sub>3</sub>	
723		H N	
724		—CH₃	
725		H <sub>3</sub> C CH <sub>3</sub>	TT.
726		СН3	TT.
727		~ h ~	TT.

Table 7 (continued)

728	N I	CU	
		—CH₃	
729		H <sub>3</sub> C CH <sub>3</sub>	
730		CH <sub>3</sub>	
731	$\boxed{}$		
732	—CH₃	—CH₃	
733	—-CH₃	H <sub>3</sub> C CH <sub>3</sub>	
734	——CH₃	CH₃ —CH₃	
735	——CH₃	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	
736	CH <sub>3</sub>	——CH₃	
737	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
738	CH <sub>3</sub>	CH <sub>3</sub>	

Table 7 (continued)

739	CII.	0 1	
	CH <sub>3</sub>	N N N	
740		—−CH <sub>3</sub>	
741		H <sub>3</sub> C CH <sub>3</sub>	
742		CH <sub>3</sub>	
743			
744		—-CH₃	
745		H₃C CH₃	$\overline{}$
746		CH <sub>3</sub> —CH <sub>3</sub>	
747		H N	
748		CH₃	
749		H <sub>3</sub> C CH <sub>3</sub>	

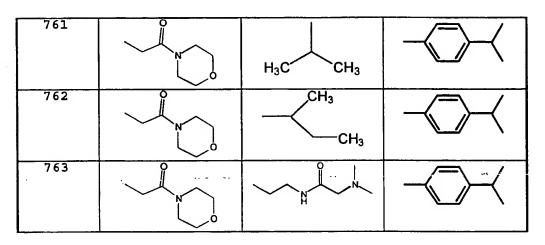
Table 7 (continued)

750		011	·
750		CH <sub>3</sub> —CH <sub>3</sub>	
751		→ N	
752	——CH₃	——CH₃	$-\bigcirc$
753	—CH₃	H <sub>3</sub> C CH <sub>3</sub>	
754	—−CH <sub>3</sub>	CH <sub>3</sub> CH <sub>3</sub>	$\overline{}$
755	CH <sub>3</sub>		
756	CH <sub>3</sub>	—CH₃	
757	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
758	CH <sub>3</sub>	CH <sub>3</sub> —CH <sub>3</sub>	
759	CH <sub>3</sub>	H N	
760		—CH₃	

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Table 7 (continued)



5

Table 8

HO 
$$R_3$$
 H O O

5

_				
	Example	R <sup>2</sup> .	R³	D.
	764		—-CH₃	N. H.
	765		H <sub>3</sub> C CH <sub>3</sub>	N H
	766		CH <sub>3</sub>	H
	767			N. A.
	768		—CH₃	A A
	769		н₃с Сн₃	\\
	770		CH <sub>3</sub> —CH <sub>3</sub>	1
	771		The state of the s	H

Table 8 (continued)

772	—-CH₃	CH <sub>3</sub>	
		- · · · · · ·	N
773	—CH₃		н \
		H <sub>3</sub> C CH <sub>3</sub>	H
774	—CH₃	CH₃	
	·· ·	CH₃	N-
775	—−CH <sub>3</sub>	NA NA	N
776	CH <sub>3</sub>	—-CH₃	
	CH <sub>3</sub>		H
777.	CH <sub>3</sub>		
	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	H
778	CH <sub>3</sub>	CH₃	
	CH <sub>3</sub>	CH₃	N N
779	CH <sub>3</sub>	H. T.	N N
780	9	CH <sub>3</sub>	
			1
781	l .		
		H₃C CH₃	H
782	l ,	CH₃	
		CH₃	H

Table 8 (continued)

783	<u> </u>	9 1	1
		VN N N N N N N N N N N N N N N N N N N	H
784		—−CH <sub>3</sub>	$-$ N $\bigcirc$
785		H <sub>3</sub> C CH <sub>3</sub>	
786		CH <sub>3</sub>	$-$ N $\bigcirc$
787		N N	$ \sim$
788		——CH₃	_<
789		H₃C CH₃	_k
790		CH <sub>3</sub> CH <sub>3</sub>	_<_<
791		The state of the s	_<
792	—CH₃	—−CH <sub>3</sub>	_K

Table 8 (continued)

793	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	-N
794	—CH₃	CH <sub>3</sub>	<b>−</b> √
795	——CH₃		_K
796	CH <sub>3</sub>	—CH₃	_N
797	CH₃ CH₃	H₃C CH₃	_<<
798	CH <sub>3</sub>	CH <sub>3</sub>	\_\_\_\
799	CH <sub>3</sub>	The state of the s	_N
800		—−CH <sub>3</sub>	_K
801		H <sub>3</sub> C CH <sub>3</sub>	-N
802		CH <sub>3</sub>	_N

Table 8 (continued)

002	0	0 1	
803		~~ n	$-$ N $\bigcirc$
804	~~~	—CH₃	_h
805		H <sub>3</sub> C CH <sub>3</sub>	H
806		CH <sub>3</sub>	-H-(_)d
807		FI	_H
808		CH <sub>3</sub>	-H-(_)d
809		H₃C CH₃	_h
810		CH <sub>3</sub>	-h-<
811		NA NA	_hd
812	——CH₃	——CH₃	_h
813	—CH₃	H <sub>3</sub> C CH <sub>3</sub>	-H-(_)<

Table 8 (continued)

814	—CH₃	CH₃	N
815	—−CH <sub>3</sub>		— <u>h</u> —(
816	ÇH₃	— CH <sub>3</sub>	H \
		3	_hd
817	CH₃ CH₃		
	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	_td
818	CH₃	CH <sub>3</sub>	-h-()-d
	CH <sub>3</sub>	СH₃	\
819	CH <sub>3</sub>	N N	—    — Q
820		——CH₃	-H-()<
821		H <sub>3</sub> C CH <sub>3</sub>	_H
822		CH <sub>3</sub>	-H-()-<
823		H N	-H-(-)
824		CH <sub>3</sub>	_N_>0

Table 8 (continued)

825	$\sim$ N $\sim$ 0	H <sub>3</sub> C CH <sub>3</sub>	
826		CH <sub>3</sub>	
827			-N
828		-—CH₃	-N
829		H₃C CH₃	$-$ N $\bigcirc$ O
830		CH <sub>3</sub>	$-\mathbf{N}$
831			
832	—−CH <sub>3</sub>	—-CH₃	7
833	—CH₃	н₃с сн₃	_N_O
834	——CH₃	CH <sub>3</sub>	_NO
835	CH <sub>3</sub>	No.	_N

-**ii**1-

Table 8 (continued)

836	ÇН₃	·CU	
030	УП3 	——CH <sub>3</sub>	./ \_
			—N 0
	CH <sub>3</sub>		
837	CH₃		
		$\downarrow$	—N 0
	° CH₃	H <sub>3</sub> C CH <sub>3</sub>	
838	CH₃	CH₃	
		, <del></del>	—-N )O
	° CH₃	CH₃	
839	ÇH₃	0	
		~~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	— й о
	CH₃	n	
840	P	CH₃	
			— Ń , o
1			
841	9	1	
			N O
1	\ \\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	H₃C CH₃	
	Ö		
842	0	CH <sub>3</sub>	
		$\overline{}$	—-N O
		СН <sub>3</sub>	
843	P P	9 1	
			—N 0
	N	H	'\ /
	<u> </u>		
844		—CH₃	NH <sub>2</sub>
	~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~		
845		1	NH <sub>2</sub>
	N O		2
		H <sub>3</sub> C CH <sub>3</sub>	
L	<u></u>	1 130 013	1

Table 8 (continued)

846		,CH₃	NH <sub>2</sub>
			14112
		CH <sub>3</sub>	
847			NH <sub>2</sub>
	N 0		
		Н	
848	/==N	—-CH <sub>3</sub>	$NH_2$
	( )		_
			179
849	N		NILI
043			NH <sub>2</sub>
		H₃C CH₃	
850	/==N	,CH₃	$NH_2$
		$\longrightarrow$	-
1		—CH₃	
053		0 1	
851	N	Ĭ	──NH <sub>2</sub>
		n	
852	—-СH <sub>3</sub>	CH <sub>3</sub>	NH <sub>2</sub>
853	CH <sub>3</sub>	-	NH <sub>2</sub>
ì	1		72
l			
		H₃C CH₃	
854	CH <sub>3</sub>	CH₃	$NH_2$
		—	
		_—СН <sub>3</sub>	
855	CH <sub>3</sub>	9 . 1	—NH <sub>2</sub>
			2
<u></u>	<u> </u>	A M. A	
856	CH <sub>3</sub>	—CH₃	$NH_2$
	CH <sub>3</sub>		
857	ÇH <sub>3</sub>	l	──NH <sub>2</sub>
			'**'2
	CH₃	H <sub>3</sub> C CH <sub>3</sub>	
	i UH <sub>2</sub>	H <sub>3</sub> C CH <sub>3</sub>	

Table 8 (continued)

858	СН₃	CH₃	——NH <sub>2</sub>
		_	_
859	CH₃	CH₃	
859	CH₃		──NH <sub>2</sub>
	CH <sub>3</sub>	, H, , ,	
860	9	CH <sub>3</sub>	NH <sub>2</sub>
•		·	
861	9.		NH <sub>2</sub>
		H <sub>3</sub> C CH <sub>3</sub>	
862	Î	CH₃	NH <sub>2</sub>
		Ch	
	<u></u>	CH₃	
863			NH <sub>2</sub>
864	, o	CU	
004	N O	CH₃	/ \
865			
	NO	u c Cu	—N >
866		H <sub>3</sub> C CH <sub>3</sub>	
	N 0	I —<	n
		CH₃	
867	-N 0		
		~ H ~ ~ . ~	-\
	<u> </u>	<u> </u>	<u> </u>

Table 8 (continued)

868	∠ N I	——CH <sub>3</sub>	
		J. 13	$-$ N $\bigcirc$
869		H <sub>3</sub> C CH <sub>3</sub>	$-$ N $\bigcirc$
870	N	CH <sub>3</sub> —CH <sub>3</sub>	_N
871	N	N N	
872	CH <sub>3</sub>	—−CH <sub>3</sub>	$-$ N $\bigcirc$
873	——CH₃	н₃с сн₃	$-$ N $\bigcirc$
874	CH <sub>3</sub>	CH <sub>3</sub> —CH <sub>3</sub>	-N
875	CH <sub>3</sub>		
876	CH <sub>3</sub>	CH₃	-\(\)
877	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	_r()
878	CH <sub>3</sub>	CH <sub>3</sub>	_\(\sigma\)

Table 8 (continued)

879	ÇH₃	0 1	
		V <sub>N</sub> √√√	_N \
	CH <sub>3</sub>		
880		—−CH <sub>3</sub>	
	$\backslash \backslash \backslash \backslash \backslash $	`	—N )
001	<u></u>		
881			<u></u>
		H <sub>3</sub> C CH <sub>3</sub>	
882		CH₃	
		CH₃	
883	9	Î J	
	N N	√ H	_N
884		——CH <sub>3</sub>	
		_	N- H
885	-NO		N
		H <sub>3</sub> C CH <sub>3</sub>	Ĥ
886	-NO	CH <sub>3</sub>	N
		—CH₃	H
887			N_
	N O	1 ~ J	H
888	/==N	—CH <sub>3</sub>	N-
			N H
889	N		N-
		н₃с Сн₃	H

Table 8 (continued)

000	<b>.</b>		
890		CH <sub>3</sub>	, K
891			HZ
892	—CH₃	—CH <sub>3</sub>	Z <sup>L</sup>
893	—CH₃	H₃C CH₃	H
894	——CH₃	CH <sub>3</sub>	HZ
895	—-CH₃	NT NT NT NT NT NT NT NT NT NT NT NT NT N	HZ
896	CH <sub>3</sub>	——CH₃	HZ
897	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	HZ
898	CH <sub>3</sub>	CH₃ —CH₃	IZ
899	CH <sub>3</sub>		HZ
900		CH <sub>3</sub>	H

Table 8 (continued)

		· · · · · · · · · · · · · · · · · · ·	
901		H <sub>3</sub> C CH <sub>3</sub>	H
902	9		··
		CH <sub>3</sub> CH <sub>3</sub>	H
903			H
904		—CH₃	
905	~~~	н₃с сн₃	
906		CH <sub>3</sub>	
907		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	$-\bigcirc$
908	$\sim$	—CH₃	
909		H <sub>3</sub> C CH <sub>3</sub>	
910	N	CH <sub>3</sub>	
911		J. J.	

Table 8 (continued)

912	CH <sub>3</sub>	CH <sub>3</sub>	<del>-</del>
913	—СН₃		
914	—-CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
075		CH <sub>3</sub>	
915	—CH₃		
916	CH <sub>3</sub>	——CH₃	
917	CH <sub>3</sub>	н <sub>3</sub> С Сн <sub>3</sub>	
918	CH <sub>3</sub>	CH <sub>3</sub>	
919	CH <sub>3</sub>	Mark Market Mark	
920		—-CH₃	
921		H <sub>3</sub> C CH <sub>3</sub>	
922		CH <sub>3</sub>	

Table 8 (continued)

923			
924		CH₃	
925		H₃C CH₃	
926		CH <sub>3</sub> —CH <sub>3</sub>	\
927		What was a second secon	
928		——CH₃	
929		H <sub>3</sub> C CH <sub>3</sub>	
930		CH <sub>3</sub>	
931		The state of the s	
932	—CH₃	CH <sub>3</sub>	
933	СН <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	

Table 8 (continued)

934	OU I	,CH₃	
334	— CH <sub>3</sub>	CH <sub>3</sub>	
935	—-СH <sub>3</sub>	V <sub>N</sub>	
936	CH <sub>3</sub>	—CH <sub>3</sub>	
937	CH <sub>3</sub>	H₃C CH₃	
938	CH <sub>3</sub>	CH <sub>3</sub>	
939	CH <sub>3</sub>		
940		—−CH <sub>3</sub>	
941		н₃с сн₃	2
942		CH <sub>3</sub> —CH <sub>3</sub>	
943		M N	
944		CH <sub>3</sub>	

Table 8 (continued)

C 04E			
945	$\sim$ N $\sim$ 0	H₃C CH₃	
946		CH <sub>3</sub>	
947	N 0	N. N. N. N. N. N. N. N. N. N. N. N. N. N	, in
948		-—CH₃	
949		H <sub>3</sub> C CH <sub>3</sub>	
950		CH <sub>3</sub>	∑ <sub>N</sub>
951		~~ H	N
952	——CH₃	—CH₃	₩ N
953	——CH₃	H <sub>3</sub> C CH <sub>3</sub>	₩ .
954	CH <sub>3</sub>	CH₃ CH₃	∑° ×
955	——CH₃	~ H	
956	CH <sub>3</sub>	CH₃	\

Table 8 (continued)

~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	······································		
957	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
958	CH <sub>3</sub>	CH₃ —CH₃	Z z
959	CH <sub>3</sub>	NH NH NH NH NH NH NH NH NH NH NH NH NH N	N N
960		——CH₃	N N
961		H <sub>3</sub> C CH <sub>3</sub>	
962		CH <sub>3</sub> —CH <sub>3</sub>	
963			
964		—CH₃	NH <sub>2</sub>
965		H <sub>3</sub> C CH <sub>3</sub>	NH <sub>2</sub>
966		CH <sub>3</sub>	NH <sub>2</sub>

Table 8 (continued)

967	$\sim$ N $_{\odot}$		S-NH <sub>2</sub>
968		CH <sub>3</sub>	S-NH <sub>2</sub>
969		H₃C CH₃	S-NH <sub>2</sub>
970		CH <sub>3</sub>	S-NH <sub>2</sub>
971	$\sim$	NA NA	S-NH <sub>2</sub>
972	CH₃	——CH₃	S-NH₂ OO
973	——CH₃	H₃C CH₃	SNH₂ OO
974	CH <sub>3</sub>	СН3	SNH₂ OOO
975	—CH₃	THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACTOR OF THE CONTRACT	NH₂ O O
976	CH <sub>3</sub>	——CH₃	S-NH <sub>2</sub>
977	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	S. NH <sub>2</sub>

Table 8 (continued)

978	ÇH₃	,CH₃	
3,0			S-NH <sub>2</sub>
		Cu	S. NH <sub>2</sub>
070	CH₃	CH₃	O'Ö
979	CH₃		
	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	~ '\\	S-NH <sub>2</sub>
	CH <sub>3</sub>		\$5.01.2 0.0
980		CH <sub>3</sub>	
	, , , , , , , , , , , , , , , , , , ,	· •	s NH2
			00
981	0		
			S-NH <sub>2</sub>
		H <sub>3</sub> C CH <sub>3</sub>	S-NF2
982	9	CH₃	
		· — (	S-NH <sub>2</sub>
		CH₃	S S S S S S S S S S S S S S S S S S S
983	9	Ŷ I	
1			S-NH <sub>2</sub>
		н	S-NH₂ O'Ö
984		—CH₃	\ c
	N O	O1 13	
			N.
985			\rangle s,
	N O		
		H₃C CH₃	, ,
986		CH <sub>3</sub>	S
	N P		
		CH <sub>3</sub>	
987			\rightarrows s_\lambda
	N O		
988	/==N	—CH <sub>3</sub>	VS.
			N
1		l	

Table 8 (continued)

989		H <sub>3</sub> C CH <sub>3</sub>	T <sub>N</sub> S
990		CH <sub>3</sub>	T <sub>N</sub>
991			. Ts
992	——CH₃	——CH₃	\[\s\]
993	—CH₃	H <sub>3</sub> C CH <sub>3</sub>	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
994	——CH <sub>3</sub>	CH <sub>3</sub> —CH <sub>3</sub>	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
995	——CH₃		) S
996	CH <sub>3</sub>	——CH₃	s ~ z
997	CH <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	S N
998	CH <sub>3</sub>	CH <sub>3</sub>	S
999	CH <sub>3</sub>	~ # ~ / · ·	S N

Table 8 (continued)

1000	0	CH <sub>3</sub>	
		OI 13	TS
			-N
1001	9		
		H₃C CH₃	
1002		CH <sub>3</sub>	S
		—CH₃	"\n"
1002	ò	0113	
1003			\rightarrows \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square \square
	) , , ,		~N
1004		—-CH₃	<b>√</b> a
	NO		
1005			\-a
	NO		
1006		H <sub>3</sub> C CH <sub>3</sub>	> -0
	N O		
1007		CH₃	~
1007	N O		79
		H	
1008	/=-N	—CH₃	79
1009	N		7-9
		H <sub>3</sub> C CH <sub>3</sub>	
1010	/=N	CH <sub>3</sub>	<u>\</u> 0
		$\vdash$	
		CH₃	

Table 8 (continued)

1011	$\sim$	~ n	TS
1012	—CH₃	—−CH <sub>3</sub>	T
1013	CH₃	H <sub>3</sub> C CH <sub>3</sub>	T3
1014	——CH₃	CH <sub>3</sub>	TS
1015	CH <sub>3</sub>		B
1016	CH <sub>3</sub>	—CH₃	T3
1017	CH <sub>3</sub>	H₃C CH₃	D
1018	CH <sub>3</sub>	CH <sub>3</sub> —CH <sub>3</sub>	
1019	CH <sub>3</sub>	EZ	
1020	J.	CH₃	
1021		H <sub>3</sub> C CH <sub>3</sub>	

Table 8 (continued)

1022	Q ·	CH₃	
		<b>─</b>	T9 1
		CH₃	
1023	Î	0	Va
		~ \\	
1024		—CH <sub>3</sub>	1.
	NO		
1025			
		H <sub>3</sub> C CH <sub>3</sub>	
1026	$\sim$	CH₃	
	N	CH <sub>3</sub>	
1027	$\sim$	Î	
	N O	→ H → II	
1028	/==N	CH <sub>3</sub>	
1029	/==N		
1030		H <sub>3</sub> C CH <sub>3</sub>	
			_/ <b>》</b>
		CH₃	
1031	/=N	l l	
1032	—СH <sub>3</sub>	CH <sub>3</sub>	

Table 8 (continued)

1033	CH <sub>3</sub>	<u> </u>	
	0, 13		_/ \\
		H <sub>3</sub> C CH <sub>3</sub>	
1034	—CH₃	CH₃	
		$\mid$ $\prec$	<b>─</b> ⟨ ⟩
		CH <sub>3</sub>	
1035	——CH <sub>3</sub>		
		#	
1036	ÇH₃	—CH₃	
			<b>─</b> (
	CH <sub>3</sub>		
1037	ÇH₃		
	CII		
1038	СН <sub>3</sub>	H <sub>3</sub> C CH <sub>3</sub>	
		— (	(
	CH₃	CH₃	
1039	ÇH₃	Î .	
			—< <u> </u>
1040	CH₃		
1010		——CH₃	_// \\
	, N		\/
1041			
1041			
		H <sub>3</sub> C CH <sub>3</sub>	
1040	, o		
1042	Î	CH₃	
	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	CH₃	
	<u></u> 6		
1043			
	N.	1 ~ h ~	

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## Treatment Process

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A process for treating a host mammal having a condition associated with pathological matrix

5 metalloprotease activity is also contemplated. That process comprises administering a metalloprotease inhibitor described hereinbefore in an MMP enzyme-inhibiting effective amount to a mammalian host having such a condition. The use of administration repeated a plurality of times is particularly contemplated.

A contemplated inhibitor compound is used for treating a host mammal such as a mouse, rat, rabbit, dog, horse, primate such as a monkey, chimpanzee or human that has a condition associated with pathological matrix metalloprotease activity.

Also contemplated is the similar use of a contemplated metalloprotease inhibitor compound in the treatment of a disease state that can be affected by the activity of metalloproteases such as TNF- $\alpha$  convertase. Exemplary of such disease states are the acute phase responses of shock and sepsis, coagulation responses, hemorrhage and cardiovascular effects, fever and inflammation, anorexia and cachexia.

In treating a disease condition associated with pathological matrix metalloproteinase activity, a contemplated MMP inhibitor compound can be used, where appropriate, in the form of an amine salt derived from an inorganic or organic acid. Exemplary acid salts include but are not limited to the following: acetate, adipate, alginate, citrate, aspartate, benzoate, benzenesulfonate, bisulfate, butyrate, camphorate, camphorsulfonate, digluconate, cyclopentanepropionate,

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dodecylsulfate, ethanesulfonate, glucoheptanoate, glycerophosphate, hemisulfate, heptanoate, hexanoate, fumarate, hydrochloride, hydrobromide, hydroiodide, 2-hydroxy-ethanesulfonate, lactate, maleate,

methanesulfonate, nicotinate, 2-naphthalenesulfonate, oxalate, palmoate, pectinate, persulfate, 3-phenylpropionate, picrate, pivalate, propionate, succinate, tartrate, thiocyanate, tosylate, mesylate and undecanoate.

Also, a basic nitrogen-containing group can be quaternized with such agents as lower alkyl (C<sub>1</sub>-C<sub>6</sub>) halides, such as methyl, ethyl, propyl, and butyl chloride, bromides, and iodides; dialkyl sulfates like dimethyl, diethyl, dibuytl, and diamyl sulfates, long chain (C<sub>8</sub>-C<sub>20</sub>) halides such as decyl, lauryl, myristyl and dodecyl chlorides, bromides and iodides, aralkyl halides like benzyl and phenethyl bromides, and others to provide enhanced water-solubility. Water or oilsoluble or dispersible products are thereby obtained as desired. The salts are formed by combining the basic compounds with the desired acid.

Other compounds useful in this invention that are acids can also form salts. Examples include salts with alkali metals or alkaline earth metals, such as sodium, potassium, calcium or magnesium or with organic bases or basic quaternary ammonium salts.

In some cases, the salts can also be used as an aid in the isolation, purification or resolution of the compounds of this invention.

Total daily dose administered to a host mammal in single or divided doses of an MMP enzyme-

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inhibiting effective amount can be in amounts, for example, of about 0.001 to about 30 mg/kg body weight daily and more usually about 0.01 to about 10 mg.

Dosage unit compositions can contain such amounts or submultiples thereof to make up the daily dose. A suitable dose can be administered, in multiple subdoses per day. Multiple doses per day can also increase the total daily dose, should such dosing be desired by the person prescribing the drug.

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The dosage regimen for treating a disease condition with a compound and/or composition of this invention is selected in accordance with a variety of factors, including the type, age, weight, sex, diet and medical condition of the patient, the severity of the disease, the route of administration, pharmacological considerations such as the activity, efficacy, pharmacokinetic and toxicology profiles of the particular compound employed, whether a drug delivery system is utilized and whether the compound is administered as part of a drug combination. Thus, the dosage regimen actually employed can vary widely and therefore can deviate from the preferred dosage regimen set forth above.

A compound useful in the present invention can be formulated as a pharmaceutical composition.

Such a composition can then be administered orally, parenterally, by inhalation spray, rectally, or topically in dosage unit formulations containing conventional nontoxic pharmaceutically acceptable carriers, adjuvants, and vehicles as desired. Topical administration can also involve the use of transdermal

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administration such as transdermal patches or iontophoresis devices. The term parenteral as used herein includes subcutaneous injections, intravenous, intramuscular, intrasternal injection, or infusion techniques. Formulation of drugs is discussed in, for example, Hoover, John E., Remington's Pharmaceutical Sciences, Mack Publishing Co., Easton, Pennsylvania; 1975 and Liberman, H.A. and Lachman, L., Eds., Pharmaceutical Dosage Forms, Marcel Decker, New York, N.Y., 1980.

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Injectable preparations, for example, sterile injectable aqueous or oleaginous suspensions can be formulated according to the known art using suitable dispersing or wetting agents and suspending agents. 15 The sterile injectable preparation can also be a sterile injectable solution or suspension in a nontoxic parenterally acceptable diluent or solvent, for example, as a solution in 1,3-butanediol. Among the acceptable vehicles and solvents that can be employed are water, Ringer's solution, and isotonic sodium chloride solution. In addition, sterile, fixed oils are conventionally employed as a solvent or suspending medium. For this purpose any bland fixed oil can be employed including synthetic mono- or diglycerides. In addition, fatty acids such as oleic acid find use in 25 the preparation of injectables. Dimethyl acetamide, surfactants including ionic and non-ionic detergents, polyethylene glycols can be used. Mixtures of solvents and wetting agents such as those discussed above are 30 also useful.

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Suppositories for rectal administration of the drug can be prepared by mixing the drug with a suitable nonirritating excipient such as cocoa butter, synthetic mono- di- or triglycerides, fatty acids and polyethylene glycols that are sold at ordinary temperatures but liquid at the rectal temperature and will therefore melt in the rectum and release the drug.

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Solid dosage forms for oral administration can include capsules, tablets, pills, powders, and granules. In such solid dosage forms, the compounds of this invention are ordinarily combined with one or more adjuvants appropriate to the indicated route of administration. If administered per os, the compounds can be admixed with lactose, sucrose, starch powder, cellulose esters of alkanoic acids, cellulose alkyl esters, talc, stearic acid, magnesium stearate, magnesium oxide, sodium and calcium salts of phosphoric and sulfuric acids, gelatin, acacia gum, sodium alginate, polyvinylpyrrolidone, and/or polyvinyl alcohol, and then tableted or encapsulated for 20 convenient administration. Such capsules or tablets can contain a controlled-release formulation as can be provided in a dispersion of active compound in hydroxypropylmethyl cellulose. In the case of 25 capsules, tablets, and pills, the dosage forms can also comprise buffering agents such as sodium citrate, magnesium or calcium carbonate or bicarbonate. Tablets and pills can additionally be prepared with enteric coatings.

30 For therapeutic purposes, formulations for parenteral administration can be in the form of aqueous

or non-aqueous isotonic sterile injection solutions or suspensions. These solutions and suspensions can be prepared from sterile powders or granules having one or more of the carriers or diluents mentioned for use in the formulations for oral administration. The compounds can be dissolved in water, polyethylene glycol, propylene glycol, ethanol, corn oil, cottonseed oil, peanut oil, sesame oil, benzyl alcohol, sodium chloride, and/or various buffers. Other adjuvants and modes of administration are well and widely known in the pharmaceutical art.

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Liquid dosage forms for oral administration can include pharmaceutically acceptable emulsions, solutions, suspensions, syrups, and elixirs containing inert diluents commonly used in the art, such as water. Such compositions can also comprise adjuvants, such as wetting agents, emulsifying and suspending agents, and sweetening, flavoring, and perfuming agents.

The amount of active ingredient that can be combined with the carrier materials to produce a single dosage form varies depending upon the mammalian host treated and the particular mode of administration.

Certain of the sulfonamide, sulfinamide or sulfenamide, compounds of this invention that are administered in accordance with an above-discussed process can serve as prodrugs to other compounds of this invention. Prodrugs are drugs that can be chemically converted in vivo or in vitro by biological systems into an active derivative or derivatives.

30 Prodrugs are administered in essentially the same manner as the other pharmaceutical compounds of the

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invention. Exemplary prodrugs correspond in structure to a compound of formula VII in which  $R^{14}$  is acyl.

## Preparation of Useful Compounds

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5 Expressly included among the individual compounds of the present invention are carboxylic acid compounds corresponding to each of the hydroxamic acid compounds of Tables 1-8. Each such carboxylic acid compound has the structure depicted for the 10 corresponding hydroxamic acid compound of the tables, except that the carboxylic acid contains an -OH group in the same location in the structure as the HO-NHgroup of the hydroxamic acid. Thus, the invention specifically includes a carboxylic acid compound 15 corresponding to each of: Examples 4-17 of Table 1; Examples 18-31 of Table 2; Examples 32-45 of Table 3; Examples 46-59 of Table 4; Examples 4-303 of Table 5; Examples 304-563 of Table 6; Examples 564-763 of Table 7; and Examples 764-1043 of Table 8. The invention 20 also specifically includes the carboxylic acid compounds corresponding to each of working Examples 1-4 that are provided hereinafter.

Schemes I and III and Schemes 1, 2, 4, 5, 6, and 7 illustrate procedures with examples of chemical transformations that may be useful for the preparation of compounds of this invention. These syntheses, as with all of the reactions discussed herein, can be carried out under a dry inert atmosphere such a nitrogen or argon if desired. Selected reactions known to those skilled in the art, can be carried out under a dry atmosphere such as dry air whereas other synthetic

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like.

steps, for example, aqueous acid or base ester or amide hydrolyses, can be carried out under laboratory air.

Thus, in general, the choices of starting material and reaction conditions can vary as is well know to those skilled in the art. Usually, no single set of conditions is limiting since variations can be applied as required. Conditions will also will be selected as desired to suit a specific purpose such as small scale preparations or large scale preparations. In either case, the use of less safe or less environmentally sound materials or reagents will usually be minimized. Examples of such less desirable materials are diazomethane, diethyl ether, heavy metal salts, dimethyl sulfide, chloroform, benzene and the

## Scheme 1

R<sub>3</sub> NHR<sub>2</sub> + CI 
$$S$$
 R<sub>1</sub>W  $THF/H_2O$  R<sub>3</sub> R<sub>1</sub>W  $R_3$   $R_1$ W  $R_3$   $R_4$   $R_4$   $R_4$   $R_5$   $R_1$ W  $R_4$   $R_4$   $R_5$   $R_1$ W  $R_4$   $R_5$   $R_1$ W  $R_4$   $R_5$   $R_4$   $R_5$   .

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Scheme 1 shows the conversion of an N-substituted alpha-amino acid, protected or unprotected, into a compound of Formula I. The amino acid may be protected with a group P such as an alkyl 5 ester such as methyl, ethyl, tert-butyl, tetrahydropyranyl and the like or arylalkyl ester such as benzyl. Treatment of this amine with a sulfonyl, sulfinyl or sulfenyl chloride would provide the corresponding amide. A base would normally be used to inactivate the HCl released from the acid chloride and 10 it would be such that it would not react with the sulfonyl chloride, i.e., ammonia, I° or II° amines would not normally be used. Examples of bases that can be used include, for example, metal hydroxides such as 15 sodium, potassium, lithium or magnesium hydroxide, oxides such as those of sodium, potassium, lithium, calcium or magnesium, metal carbonates such as those of sodium, potassium, lithium, calcium or magnesium, metal bicarbonates such as sodium bicarbonate or potassium bicarbonate, I°, II° or III° organic amines such as 20 alkyl amines, arylalkyl amines, alkylarylalkyl amines, heterocyclic amines or heteroaryl amines, ammonium hydroxides or quaternary ammonium hydroxides. As nonlimiting examples, such amines can include triethyl amine, trimethyl amine, diisopropyl amine, 25 methyldiisopropyl amine, diazabicyclononane, tribenzyl amine, dimethylbenzyl amine, morpholine, N-methylmorpholine, N, N'-dimethylpiperazine, N-ethylpiperidine, 1,1,5,5-tetramethylpiperidine, 30 dimethylaminopyridine, pyridine, quinoline, tetramethylethylenediamine and the like. Non-limiting

examples of ammonium hydroxides, usually made from amines and water, can include ammonium hydroxide, triethyl ammonium hydroxide, trimethyl ammonium hydroxide, methyldiiospropyl ammonium hydroxide, tribenzyl ammonium hydroxide, dimethylbenzyl ammonium hydroxide, morpholinium hydroxide, N-methylmorpholinium hydroxide, N, N'-dimethylpiperazinium hydroxide, N-ethylpiperidinium hydroxide, and the like. As nonlimiting examples, quaternary ammonium hydroxides can 10 include tetraethyl ammonium hydroxide, tetramethyl ammonium hydroxide, dimethyldiiospropyl ammonium hydroxide, benzymethyldiisopropyl ammonium hydroxide, methyldiazabicyclononyl ammonium hydroxide, methyltribenzyl ammonium hydroxide, N,N-15 dimethylmorpholinium hydroxide, N,N,N', N',tetramethylpiperazenium hydroxide, and N-ethyl-N'hexylpiperidinium hydroxide and the like. Metal hydrides, amide or alcoholates such as calcium hydride, sodium hydride, potassium hydride, lithium hydride, 20 sodium methoxide, potassium tert-butoxide, calcium ethoxide, magnesium ethoxide, sodium amide, potassium diisopropyl amide and the like may also be suitable reagents. Organometallic deprotonating agents such as alkyl or aryl lithium reagents such as methyl, phenyl or butyl lithium, Grignard reagents such as methylmagnesium bromide or methymagnesium chloride, organocadium reagents such as dimethylcadium and the like may also serve as bases for causing salt formation or catalyzing the reaction. Quaternary ammonium

hydroxides or mixed salts are also useful for aiding

phase transfer couplings or serving as phase transfer reagents.

The first reaction in Scheme 1 also illustrated the use of a mixed solvent THF/H2O. This is one solvent system however others may be useful also. For example, the reaction media can consist of a single solvent, mixed solvents of the same or different classes or serve as a reagent in a single or mixed solvent system. The solvents can be protic, non-protic 10 or dipolar aprotic. Non-limiting examples of protic solvents include water, methanol (MeOH), denatured or pure 95% or absolute ethanol, isopropanol and the like. Typical non-protic solvents include acetone, tetrahydrofurane (THF), dioxane, diethylether (ether), tert-butylmethyl ether (TBME), aromatics such as xylene, toluene, or benzene, ethyl acetate, methyl acetate, butyl acetate, trichloroethane, methylene chloride, ethylenedichloride (EDC), hexane, heptane, isooctane, cyclohexane and the like. Dipolar aprotic solvents include compounds such as dimethylformamide 20 (DMF), dimethylacetamide (DMAc), acetonitrile, nitromethane, tetramethylurea, N-methylpyrrolidone and the like.

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## Scheme 2

P = H, protecting group

Non-limiting examples of ammonium hydroxides,

5 usually made from amines and water, can include ammonium hydroxide, triethyl ammonium hydroxide, trimethyl ammonium hydroxide, methyldiiospropyl ammonium hydroxide, tribenzyl ammonium hydroxide, dimethylbenzyl ammonium hydroxide, morpholinium 10 hydroxide, N-methylmorpholinium hydroxide, N, N'dimethylpiperazinium hydroxide, N-ethylpiperidinium hydroxide, and the like. As non-limiting examples, quaternary ammonium hydroxides can include tetraethyl ammonium hydroxide, tetramethyl ammonium hydroxide, 15 dimethyldiiospropyl ammonium hydroxide, benzymethyldiisopropyl ammonium hydroxide, methyldiazabicyclononyl ammonium hydroxide,

methyltribenzyl ammonium hydroxide, N,Ndimethylmorpholinium hydroxide, N,N,N',

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N', -tetramethylpiperazenium hydroxide, and N-ethyl-N'hexylpiperidinium hydroxide and the like. Metal hydrides, amide or alcoholates such as calcium hydride, sodium hydride, potassium hydride, lithium hydride, sodium methoxide, potassium tert-butoxide, calcium ethoxide, magnesium ethoxide, sodium amide, potassium diisopropyl amide and the like may also be suitable reabents. Organometallic deprotonating agents such as alkyl or aryl lithium reagents such as methyl, phenyl or butyl lithium, Grignard reagents such as methylmagnesium bromide or methymagnesium chloride, organocadium reagents such as dimethylcadium and the like may also serve as bases for causing salt formation or catalyzing the reaction. Quaternary ammonium 15 hydroxides or mixed salts are also useful for aiding phase transfer couplings or serving as phase transfer reagents.

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acetate, butyl acetate, trichloroethane, methylene chloride, ethylenedichloride (EDC), hexane, heptane, isooctane, cyclohexane and the like. Dipolar aprotic solvents include compounds such as dimethylformamide (DMF), dimethylacetamide (DMAc), acetonitrile, nitromethane, tetramethylurea, N-methylpyrrolidone and the like.

Non-limiting examples of reagents that can be used as solvents or as part of a mixed solvent system include organic or inorganic mono- or multi-protic 10 acids or bases such as hydrochloric acid, phosphoric acid, sulfuric acid, acetic acid, formic acid, citric acid, succinic acid, triethylamine, morpholine, N-methylmorpholine, piperidine, pyrazine, piperazine, pyridine, potassium hydroxide, sodium hydroxide, alcohols, ammonia or amines for making esters or amides and the like.

Acids are used in many reactions during various synthesis. Scheme 1 illustrates acid use for the removal of the THP protecting group to produce the hydroxamic acid of Formula I. The acid might be mono-, di- or tri-protic organic or inorganic acids. Examples of acids include hydrochloric acid, phosphoric acid, sulfuric acid, acetic acid, formic acid, citric acid, succinic acid, hydrobromic acid, hydrofluoric acid, carbonic acid, phosphorus acid, p-toluene sulfonic acid, trifluoromethane sulfonic acid, trifluoroacetic acid, difluoroacetic acid, benzoic acid, methane sulfonic acid, benzene sulfonic acid, 2,6-30 dimethylbenzene sulfonic acid, trichloroacetic acid, nitrobenzoic acid, dinitrobenzoic acid, trinitrobenzoic

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acid, and the like. They might also be Lewis acids such as aluminum chloride, borontrifluoride, antimony pentafluoride and the like. A preferred solvent in this type reaction is dioxane eith an alcohol or water 5 however almost any solvent system with one component being a protic solvent can be useful.

Scheme I illustrates conversion of a carboxylic acid protected as an ester or amide into an hydroxamic acid derivative such as a O-arylalkylether or O-cycloalkoxyalkylether group. In particular, the this Scheme the protecting group on the hydroxylamine is the THP group. In the case where hydroxylamine is used, treatment of an ester or amide with one or more equivalents of hydroxylamine hydrochloride at room temperature or above in a solvent or solvents, usually protic or partially protic, such as those listed above can provide a hydroxamic acid directly. This exchange process may be further catalyzed by the addition of additional acid. Alternatively, a base such as a salt of an alcohol used as a solvent, for example, sodium methoxide in methanol, can be used to form hydroxylamine from hydroxylamine hydrochloride in situ which can exchange with an ester or amide. mentioned above, exchange can be carried out with a protected hydroxyl amine such as tetrahydropyranylhydroxyamine (THPONH2), benzylhydroxylamine (BnONH2), and the like in which case compounds such as shown in Scheme 1 that are tetrahydropyranyl (THP) or benzyl (Bn) hydroxamic acid derivatives are the products. 30 Removal of the protecting groups when desired, for example, following further transformations in another

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part of the molecule or following storage, is accomplished by standard methods well known in the art such as acid hydrolysis of the THP group as discussed above or reductive removal of the benzyl group with 5 hydrogen and a metal catalyst such as palladium, platinum, palladium on carbon or nickel.

In the case where P is hydrogen, i.e., where the intermediate is a carboxylic acid, standard coupling reactions can be used. For example, the acid 10 can be converted into an acid chloride, mixed anhydride or activated ester and treated with hydroxylamine or a ... protected hydroxylamine in the presence of a noncompetitive base to the nitrogen acylated compound. This is the same product as discussed above. Couplings of this nature are well known in the art and especially the art related to peptide and amino acid chemistry.

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Scheme II illustrates another possible synthesis of the compounds of Formula I starting with a protected or unprotected amino acid. Sulfonylation of the amino group is accomplished as discussed above to produce the sulfonamide II-B. This compound is a secondary sulfonamide and, as such, is acidic and can be alkylated with an R<sup>2</sup> group. Alkylation, a process well known in the art, can be carried by treatment of the sulfonamide with base to form the corresponding anion, adding an electrophilic reagent and allowing the SN2 reaction to proceed. Electrophiles include halogen derivatives, sulfonate esters, epoxides and the like. The bases and solvents discussed with regard to Scheme I are applicable in this Scheme. Preferred bases are those that are hindered such that competition with the

electrophile is minimized. Additional preferred bases are metal hydrides, amide anions or organometallic bases such as a butyl lithium. The solvents, solvent mixtures or solvent/reagent mixtures discussed are satisfactory but non-protic or dipolar aprotic solvents such as acetone, acetonitrile, DMF and the like are examples of preferred classes.

Scheme III illustrates the potential for use of a sulfonyl chloride reagent, specifically 10 nitrobenzenesulfonyl chloride, to prepare compounds of this invention. It should be noted that this reagent is for illustration and is not to be considered limiting or required. After coupling with an amino acid and alkylation of the coupling product if 15 required, the nitrosulfonamide can be reduced to provide a useful amino compound. The amino group can be alkylated if desired. It can also be acylated with an aroyl chloride, heteroaryl or other  ${\bf R}^{\bf 6}$  amine carbonyl froming agent to form a -C(=0) - or -S(=0) n-20 compound of this invention. The amino sulfonamide can also be reacted with a carbonic acid ester chloride as shown in Scheme IV, a sulfonyl chloride as shown in Scheme V or in Scheme VII or a carbamoyl chloride or isocyanate as shown in Scheme VI to produce the 25 corresponding carbamate, sulfonamides, or ureas of this invention. Acylation of amines of this type are well known in the art and the reagents are also well known. Usually these reactions are carried out in aprotic solvents under an inert or/and dry atmosphere at about 30 45°C to about -10°C. An equivalent of a non-

competitive base is usually used with sulfonyl

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chloride, acid chloride or carbonyl chloride reagents. Following this acylation step, synthesis of the hydroxamic acid products of this invention can proceed as discussed above for Scheme I and Scheme II.

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Schemes II through VI also illustrate the possible reduction of a nitrobenzenesulfonamide to produce an amino sulfonamide. The reduction of nitro groups to amines is will know in the art with a preferred method being hydrogenation. There is usually 10 a metal catalyst such as Rh, Pd, Pt, Ni or the like with or without an additional support such as carbon, barium carbonate and the like. Solvents can be protic or non-protic pure solvents or mixed solvents as required. The reductions can be carried out at atmospheric pressure to a pressure of multiple atmospheres with atmospheric pressure to about 40 pounds per square inch (psi) preferred.

Other sulfonyl chloride reagents can also be used in the preparation of compounds of this invention as outline in the Schemes. Examples are fluoroaryl or fluoroheteroaryl sulfonyl chlorides, azidoaryl or azidoheteroaryl or amide, carbonate, carbamate or urea substituted aryl or heteroaryl sulfonyl chloride reagents. Azides, for example, can be reduced to an amino group using hydrogen with a metal catalyst or metal chelate catalyst or activated hydride transfer reagent. The fluoro substituted sulfonic acid or sulfonamide can be treated with a nucleophile such as ammonia or a primary amine, under pressure if desired, to provide an amino or substituted (R5) amino group

that can then be reacted a reagent as outline in Scheme III and in Schemes 4-7 inclusive.

#### Scheme III

# Scheme 4

# Scheme 5

Formula V

### Scheme 6

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#### Scheme 7

P = H, protecting group

Intermediate VI-C

Compounds of the present can possess one or more asymmetric carbon atoms and are thus capable of 5 existing in the form of optical isomers as well as in the form of racemic or nonracemic mixtures thereof. The optical isomers can be obtained by resolution of the racemic mixtures according to conventional processes well known in the art, for example by 10 formation of diastereoisomeric salts by treatment with an optically active acid or base. Examples of appropriate acids are tartaric, diacetyltartaric, dibenzoyltartaric, ditoluoyltartaric and camphorsulfonic acid and then separation of the mixture 15 of diastereoisomers by crystallization followed by liberation of the optically active bases from these salts. A different process for separation of optical isomers involves the use of a chiral chromatography column optimally chosen to maximize the separation of 20 the enantiomers. Still another available method involves synthesis of covalent diastereoisomeric

molecules, e.g., esters, amides, acetals, ketals, and the like, by reacting compounds of Formula I with an optically active acid in an activated form, a optically active diol or an optically active isocyanate. synthesized diastereoisomers can be separated by conventional means such as chromatography, distillation, crystallization or sublimation, and then hydrolyzed to deliver the enantiomericaly pure compound. In some cases hydrolysis to the parent optically active drug is not necessary prior to dosing the patient since the compound can behave as a prodrug. The optically active compounds of Formula I can likewise be obtained by utilizing optically active starting materials.

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Contemplated equivalents of the general formulas set forth above for the MMP inhibitor compounds and derivatives as well as the intermediates are compounds otherwise corresponding thereto and having the same general properties such as tautomers thereof and compounds wherein one or more of the various R groups are simple variations of the substituents as defined therein, e.g., wherein R is a higher alkyl group than that indicated. In addition, where a substituent is designated as, or can be, a hydrogen, the exact chemical nature of a substituent which is other than hydrogen at that position, e.g., a hydrocarbyl radical or a halogen, hydroxy, amino and the like functional group, is not critical so long as it does not adversely affect the overall activity and/or synthesis procedure. For 30 example, two hydroxyl groups, two amino groups, two thiol groups or a mixture of two hydrogen-heteroatom groups on

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the same carbon are know not to be stable without protection or as a derivative.

The chemical reactions described above are generally disclosed in terms of their broadest application to the preparation of the compounds of this invention. Occasionally, the reactions may not be applicable as described to each compound included within the disclosed scope. The compounds for which this occurs will be readily recognized by those skilled in the art. In all such cases, either the reactions can be 10 successfully performed by conventional modifications known to those skilled in the art, e.g., by appropriate protection of interfering groups, by changing to alternative conventional reagents, by routine modification of reaction conditions, and the like, or 15 other reactions disclosed herein or otherwise conventional, will be applicable to the preparation of the corresponding compounds of this invention. preparative methods, all starting materials are known or readily preparable from known starting materials. 20

### Best Mode for Carrying Out the Invention

Without further elaboration, it is believed that one skilled in the art can, using the preceding description, utilize the present invention to its fullest extent. The following preferred specific embodiments are, therefore, to be construed as merely illustrative, and not limiting of the remainder of the disclosure in any way whatsoever.

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Example 1. N-hydroxy-2(R)-[[(4-benzoylamino)-benzenesulfonyl][(4-ethylmorpholino)amino]3-methylbutanamide, hydrochloride

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Part A: To a stirred solution of D-valine (10.0q) in a 2:1 mixture of THF/water (200mL) containing 3 equiv. of triethylamine at 5°C is added 10 4-(benzoylamino)benzenesulfonyl chloride (0.9 equiv.) and the reaction is stirred to room temperature overnight. The resulting mixture is diluted with dichloromethane, and washed with 1N HCl, and water. The organic layer is dried over magnesium sulfate, 15 filtered and concentrated in vacuo to provide the desired N-[(4-benzoylamino)benzenesulfonyl]-(D)-valine as a crude product. A solution of the crude acid (10 grams) in 80 mL of dry toluene is charged with dimethylformamide di-tert-butylacetal (45 mL) and 20 heated to 100°C for several hours. The resulting cooled solution is concentrated by rotary evaporation and purified by silica gel chromatography to provide

the desired N-[(4-benzoylamino)benzenesulfonyl]-(D)-valine, tert-butyl ester.

Part B: To a dimethylformamide solution (100 mL) of the tert-butyl ester from Part A (5.0 grams) is added 4-(2-chloroethyl)-morpholine hydrochloride (1.3 equivalents) and potassium carbonate (3 equivalents); this suspension is heated to 70°C for five hours. resulting suspension is cooled and diluted with water (500 mL) and extracted with ethyl acetate. The organic layer is washed with water (2 x 200 mL), saturated 10 sodium bicarbonate (2 x 200mL), and brine, dried over magnesium sulfate, filtered and concentrated to yield the desired product, t-butyl 2(R)-[[(4-benzoylamino)benzenesulfonyl] [(4-ethylmorpholino)amino]-3methylbutanoate, which can be purified by silica gel 15 chromatography.

Part C: t-Butyl 2(R)-[[(4-benzoylamino)benzenesulfonyl][(4-ethylmorpholino)amino]-3methylbutanoate from Part A (4.0 grams) in 100 mL of anhydrous dichloromethane, is cooled to  $-5\,^{\circ}\text{C}$  , and dry 20 HCl gas is bubbled into the reaction flask for 20 minutes. The flask is the sealed and allowed to warm to room temperature over three to four hours. solvent is removed by rotary evaporation and the desired acid, hydrochloride salt, (2(R)-[[(4-25 benzoylamino)benzenesulfonyl][(4ethylmorpholino) amino] -3-methylbutanoic acid hydrochloride) is triturated with ether hexane and filtered. The dried precipitate is used without 30 further purification.

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Part D: 2(R)-[[(4-benzoylamino)benzenesulfonyl] [(4-ethylmorpholino)amino]-3-methylbutanoic acid hydrochloride from Part C (3.8 grams) and 1-hydroxybenzotriazole (1.5 equivalents) are dissolved in anhydrous dimethylformamide (50 mL) and cooled to 10°C. To this is added N-methylmorpholine (3 equivalents) followed by EDC (1.1 equivalents) and this solution is stirred for two hours at 10°C. To this is added O-tetrahydropyranyl (O-THP) hydroxylamine (2 equivalents) and the solution is stirred overnight 10 at room temperature. The resulting mixture is diluted with 200 mL of water and extracted with ethyl acetate. The organic layer is washed with saturated sodium bicarbonate and brine, dried over magnesium sulfate, filtered and concentrated to yield the desired O-THP-15 protected hydroxamate, N-tetrahydropyranyloxy-2(R)-[[(4-benzoylamino)-benzenesulfonyl][(4ethylmorpholino) -amino] -3-methylbutanamide. Further purification by silica gel chromatography provides a mixture of desired diastereomers which are combined. 20

N-tetrahydropyranyloxy-2(R)-[[(4-benzoylamino)-benzenesulfonyl][(4-ethylmorpholino)amino]-3-methylbutanamide (4.0 grams), is dissolved in a solution of dioxane(150 mL), and ethanol (1mL) and cooled to -5°C. To this is added HCl (5 equivalents; 4N in dioxane) and the solution is stirred for one hour. The contents are concentrated and the desired product is triturated from ether/hexane and filtered to yield the desired N-hydroxy-2(R)-[[(4-benzoylamino)benzenesulfonyl]-[(4-ethylmorpholino)-amino]-3-methylbutanamide, hydrochloride.

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Example 2. N-hydroxy-2(R)-[[(4-benzoylamino)benzene-sulfonvl][(3-picolyl)aminolpropanamide

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Part A: To a solution of D-alanine methyl 10 ester hydrochloride (6.0 grams) in methanol (25 mL) is added 1.1 equivalents of 3-Pyridinecarboxaldehyde. solution is stirred at room temperature for several hours and then the solvent removed by rotary evaporation. The resulting imine is redissolved in 15 acetic acid (10 mL) and methanol (2 mL), and sodium cyanoborohydride (1.5 equivalents) is added in several portions over 10 minutes. The mixture is then stirred for 16 hours and then concentrated by rotary evaporation. The resulting residue is partitioned between ethyl acetate and aqueous sodium carbonate (10%). The organic layer is dried over magnesium sulfate, filtered and concentrated to yield the desired N-3-picolyl-D-alanine methyl ester.

Part B: To a stirred solution of N-3picolyl-D-alanine methyl ester (5.0g) in a 2:1 mixture
of THF/water (200mL) containing triethylamine (3
equiv.) at 5°C is added 4-(benzoylamino)benzenesulfonyl
chloride and the reaction is stirred to room

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temperature overnight. The resulting mixture is diluted with dichloromethane, and washed with 1 N HCl, and water. The organic layer is dried over magnesium sulfate, filtered and concentrated in vacuo to provide 5 the desired methyl 2(R)-[[(4-benzoylamino)benzenesulfonyl] [(3-picolyl)amino]propionate as a crude product, which may be purified by silica gel chromatography using ethyl acetate/hexanes as eluant.

Part C: To a solution of methyl ester from Part B (2.0 grams) in tetrahdrofuran (100 mL) is added 10 a solution of aqueous 1N sodium hydroxide (1.2 equiv.) and the reaction mixture is stirred for 20 The solvents are removed by rotary evaporation and the remaining oil is partitioned between ethyl acetate and 1N hydrochloric acid. The organic layer is 15 separated and dried over magnesium sulfate, filtered and concentrated to yield 2(R)-[[(4-benzoylamino)benzenesulfonyl][(3-picolyl)amino]propionic acid.

2(R)-[[(4-benzoylamino)benzenesulfonyl]-[(3-picolyl)amino]propionic acid (1.3 grams) and 1-hydroxybenzotriazole (1.5 equivalents) are dissolved in anhydrous dimethylformamide (25 mL) and cooled to 10°C. To this is added N-methylmorpholine (3 equivalents) followed by EDC (1.1 equivalents) and this solution is stirred for two hours at 10°C. To this 25 solution is added O-tetrahydropyranyl hydroxylamine (2 equivalents) and this solution is stirred overnight to room temperature. The solution is diluted with water (100 mL) and extracted with ethyl acetate. The organic layer is washed with saturated sodium 30 bicarbonate and brine, dried over magnesium sulfate,

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filtered and concentrated to yield the desired O-THP protected hydroxamate which is further purified by silica gel chromatography to give a mixture of desired diastereomers which are combined.

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N-tetrahydropyranyloxy-2(R)-[[(4benzoylamino) - benzenesulfonyl] [(3-picolyl)amino] propanamide (1.0 gram), is dissolved in a solution of dioxane (50 mL), and ethanol (1mL) and cooled to -5°C. To this is added 5 equivalents of HCl (4N in dioxane) and the solution is stirred for one hour. The contents are concentrated and the desired product is triturated with ether/hexane and filtered to yield the desired N-hydroxy-2(R)-[[(4-benzoylamino)benzenesulfonyl]-[(3-picolyl)amino]propanamide, hydrochloride. 15 crude product may be purified by silica gel chromatography using methanol and methylene chloride as the eluant, after neutralization to the free base.

Example 3. N-hydroxy-2(R)-[[(4-(benzenesulfonyl)amino)-20 benzenesulfonyl] [(4-ethylmorpholino)amino]-3-methylbutanamide, hydrochloride

Part A: To a stirred solution of D-valine (10.0g) in a 2:1 mixture of THF/water (200mL) 5 containing triethylamine (3 equiv.) at 5°C is added 4-nitrobenzene-sulfonyl chloride (0.9 equiv.) and the reaction is stirred at room temperature overnight. resulting mixture is diluted with dichloromethane, and washed with 1N HCl, and water. The organic layer is 10 dried over magnesium sulfate, filtered and concentrated in vacuo to provide the desired N-(nitrobenzenesulfonyl) - (D) -valine as a crude amino acid product. A solution of the crude amino acid (10 grams) in dry toluene (80 mL) is charged with dimethylformamide di-tertbutylacetal (45 mL) and heated to 100°C for several hours. The resulting cooled solution is concentrated by rotary evaporation and purified by silica gel chromatography to provide the desired N-(4-nitrobenzenesulfonyl)-(D)-valine, tertbutyl ester. 20

Part B: To a dimethylformamide solution (100 mL) of the tert-butyl ester from Part A (5.0 grams) is added 4-(2-chloroethyl)-morpholine hydrochloride (1.3 equivalents) and potassium carbonate (3 equivalents); and this suspension is heated to 70°C for five hours. The resulting suspension is cooled and diluted with water (500 mL) and extracted with ethyl acetate. The organic layer is washed with water (2 x 200 mL), saturated sodium bicarbonate (2 x 200 mL), and brine, dried over magnesium sulfate, filtered and concentrated to yield the desired product, t-butyl 2(R)-[4-

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nitrobenzenesulfonyl] [(4-ethylmorpholino)amino]-3methylbutanoate, which can be purified by silica gel chromatography.

Part C: t-Butyl 2(R)-[4-nitrobenzene-5 sulfonyl] [(4-ethylmorpholino)amino]-3-methylbutanoate (1.5 grams) is added to a Fisher® porter bottle containing ethanol/THF (50 mL each) and 200 mg of 10% palladium on carbon. The bottle is flushed with nitrogen while stirring and then charged with hydrogen 10 at a pressure of 50 psig. After one hour the bottle is flushed with nitrogen and the resulting suspension is filtered through Celite. The filtrate is concentrated to yield t-butyl 2(R)-[4-aminobenzenesulfonyl]-[(4-ethylmorpholino)amino]-3-methylbutanoate.

Part D: To a solution of t-butyl 2(R)-[4aminobenzenesulfonyl] - [(4-ethylmorpholino)amino] -3methylbutanoate from Part C (1.30 grams) in tetrahydrofuran containing N-methylmorpholine (1.2 equivalents) is added benzenesulfonyl chloride (1.0 equivalents); this solution was stirred for 16 hours. 20 The contents were diluted with ethyl acetate and washed with 5% KHSO4, saturated sodium bicarbonate and brine, dried over magnesium sulfate, filtered and concentrated to yield the desired t-butyl-2(R)-[[(4-

(benzenesulfonyl) amino) -benzenesulfonyl] [(4-25 ethylmorpholino)amino]-3-methylbutanoate (1.5 grams).

Part E: t-Butyl-2(R)-[[(4-(benzenesulfonyl)amino)-benzenesulfonyl][(4ethylmorpholino) amino] - 3-methylbutanoate from Part D (1.3 grams) in 50 mL of anhydrous dichloromethane is cooled to -5°C, and dry HCl gas is bubbled into the

reaction flask for 20 minutes. The flask is the sealed and allowed to warm to room temperature over three to four hours. The solvent is removed by rotary evaporation and the desired acid, hydrochloride salt is triturated with ether hexane and filtered. The dried precipitate is used without further purification.

t-Butyl-2(R)-[[(4-(benzenesulfonyl)amino)benzenesulfonyl][(4-ethylmorpholino)amino]-3methylbutanoate (1.0 grams) and 1-hydroxybenzotriazole (1.5 equivalents) are dissolved in anhydrous 10 dimethylformamide(25 mL) and cooled to 10°C. To this is added N-methylmorpholine (3 equivalents) followed by EDC (1.1 equivalents) and this solution is stirred for two hours at 10°C. To this solution is added 0tetrahydropyranyl (0-THP) hydroxylamine (2 15 equivalents); this solution is stirred overnight at room temperature. The solution is diluted with water (100 mL) and extracted with ethyl acetate. The organic layer is washed with saturated sodium bicarbonate and brine, dried over magnesium sulfate, filtered and 20 concentrated to yield the desired O-THP-protected hydroxamate which is further purified by silica gel chromatography to give a mixture of desired diastereomers which are combined.

N-tetrahydropyranyloxy-2(R)-[[(4-(benzenesulfonyl)-amino)benzenesulfonyl][(4-ethylmorpholino)amino]-3-methylbutanoic acid (0.80 grams), is dissolved in a solution of dioxane (20mL), and ethanol (1mL) and cooled to -5°C. To this is added HCl (5 equivalents; 4N in dioxane) and the solution is stirred for one hour. The contents are concentrated

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and the desired product is triturated from ether/hexane and filtered to yield the desired N-hydroxy-2(R)-[[(4-(benzenesulfonyl)amino)benzenesulfonyl][(4-ethylmorpholino)amino]-3-methylbutanamide,

- 5 hydrochloride. This crude product may be purified by silica gel chromatography using methanol and methylene chloride as the eluant, on the free base.
- 10 Example 4: (R)-N-[4-[[[2-(hydroxyamino)-1-methyl-2-oxoethyl][2-(4-morpholinyl)ethyl]

  aminolsulfonyllphenyllbenzamide

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Part A: To a solution of D-alanine, t-butyl ester hydrochloride (9.80 g, 53.9 mmol) in H<sub>2</sub>O (64 mL) and acetone (26 mL) was added triethylamine (17.3 mL, 124 mmol) and the solution was cooled to zero degrees Celsius. To this solution was added 4-nitrobenzenesulfonyl chloride (11.1 g, 50.2 mmol) dropwise in acetone (25 mL). The solution was stirred for 72 hours. The solution was concentrated in vacuo and the residue was dissolved into ethyl acetate. The solution was washed with 5% KHSO<sub>4</sub>, saturated NaHCO<sub>3</sub> and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Recrystallization

(ethyl acetate/hexane) provided the sulfonamide as a solid (10.87 g, 66 %).

Part B: To a solution of the sulfonamide of part A (10.8 g, 32.7 mmol) in DMF (60 mL) was added 4-(2-chloroethyl)morpholine(12.2 g, 65.4 mmol) and K<sub>2</sub>CO<sub>3</sub> (13.6 g, 98.0 mmol) and the solution was heated to seventy degrees Celsius for 7 hours. The solution was partitioned between ethyl acetate and H<sub>2</sub>O. The organic layer was washed with H<sub>2</sub>O, saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Concentration in vacuo followed by tritration (ethyl ether) provided the morpholine compound as a solid (8.48 g, 59 %). MS(CI) MH<sup>+</sup> calculated for C<sub>19</sub>H<sub>29</sub>N<sub>3</sub>O<sub>7</sub>S: 444, found: 444.

Part C: To a solution of the morpholine

15 compound of part B (8.49 g, 19.1 mmol) in THF (100 mL)

under atmosphere of 50 psi of hydrogen was added 4%

Pd/C and the solution was stirred for 2 hours until

uptake stopped. The solution was filtered through

Celite and concentration in vacuo of the filtrate

20 provided the aniline as a solid (8.5 g, quantitative

yield). MS(CI) MH<sup>+</sup> calculated for C<sub>19</sub>H<sub>31</sub>N<sub>3</sub>O<sub>5</sub>S: 414,

found: 414.

Part D: To a solution of the aniline of part C (2.0 g, 4.8 mmol) in THF (16 mL) was added

25 triethylamine (3.0 mL, 21.3 mmol) and the solution was cooled to zero degrees Celsius. To this solution was added benzoyl chloride (1.46 mL, 12.6 mmol) and the solution was stirred for 18 hours at ambient temperature. The solution was concentrated in vacuo and the residue was partitioned between ethyl acetate and saturated NaHCO3. The organic layer was washed with

saturated NaCl and dried over  $Na_2SO_4$ . Chromatography (on silica gel, ethyl acetate/methanol) provided the benzamide as a solid (2.0 g, 80 %). MS(CI) MH<sup>+</sup> calculated for  $C_{26}H_{35}N_3O_6S$ : 518, found: 518.

Part E: To a solution of the benzamide of part D (2.0 g, 3.9 mmol) in anisole (9 mL) was added trifluoroacetic acid (26 mL) and the solution was stirred for 18 hours. The solution was concentrated in vacuo to remove the trifluoroacetic acid. The remaining solution was poured into ethyl ether and the resulting solid was collect by vacuum filtration to provide the acid as a white solid (1.09 g, 50 %).

MS(CI) MH+ calculated for C<sub>22</sub>H<sub>27</sub>N<sub>3</sub>O<sub>6</sub>S: 462, found: 462.

Part F: To a solution of the acid of part E

(1.09 g, 1.89 mmol) in methanol (3 mL) cooled to zero
degrees Celsius was added thionyl chloride (0.18 mL,
2.4 mmol) and the solution was stirred at ambient
temperature for 18 hours. The solution was partitioned
between ethyl acetate and saturated NaHCO3. The organic
layer was washed with saturated NaHCO3 and saturated
NaCl and dried over Na2SO4. Reverse phase
chromatography (on silica; acetonitrile/H2O) provided
the methyl ester as a white solid (650 mg, 72 %).
MS(CI) MH+ calculated for C23H29N3O6S: 476, found: 476.

Part G: To a solution of the methyl ester of part F (650 mg, 1.4 mmol) in methanol (1.6 mL) and THF (1.6 mL) was added 50% aqueous hydroxylamine (1.6 mL). The solution was stirred for 18 hours. The solution was concentrated in vacuo and the residue was partitioned between ethyl acetate and  $\rm H_2O$ . The organic layer was washed with saturated NaCl and dried over

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Na<sub>2</sub>SO<sub>4</sub>. Concentration in vacuo provided (R)-N-[4-[[[2-(hydroxyamino)-1-methyl-2-oxoethyl] [2-(4-morpholinyl)ethyl]amino]sulfonyl]phenyl]benzamide as a white solid (380 mg, 58 %). MS(CI) MH<sup>+</sup> calculated for C<sub>22</sub>H<sub>28</sub>N<sub>4</sub>O<sub>6</sub>S: 477, found: 477.

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Example 4a: (R)-N-[4-[[[2-(hydroxyamino)-1-methyl-2-oxoethyl] [2-4-morpholinyl)ethyl]

amino]sulfonyl]phenyl]benzamide,

monohydrochloride

To a solution of N-hydroxy-2(R)-[[(4-

benzoylamino) benzenesulfonyl] [(4ethylmorpholino) amino] -3-methylbutanamide,

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hydrochloride of Example 1 (380 mg, 0.8 mmol) in acetonitrile (30 mL) was added (0.13 mL; 1.59 mmol) 12N HCl and the solution was stirred for 10 minutes. The solution was concentrated *in vacuo* and the residue was triturated with ethyl ether to provide the hydrochloride salt as a white solid (349 mg, 85 %).

MS(CI) MH<sup>+</sup> calculated for C<sub>22</sub>H<sub>28</sub>N<sub>4</sub>O<sub>6</sub>S: 477, found: 477.

Example 5: (R)-4-bromo-N-[4-[[[2-(hydroxyamino)-110 methyl-2-oxoethyl][2-(4-morpholinyl)ethyl]
amino]sulfonyl]phenyl]benzamide,
mono(trifluoroacetate) (salt)

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Part A: To a solution of D-alanine, t-butyl ester hydrochloride (9.80 g, 53.9 mmol) in H<sub>2</sub>O (64 mL) and acetone (26 mL) was added triethylamine (17.3 mL, 124 mmol) and the solution was cooled to zero degrees Celsius. To this solution was added 4-nitrobenzenesulfonyl chloride (11.1 g, 50.2 mmol) dropwise in acetone (25 mL). The solution was stirred for 72 hours. The solution was concentrated in vacuo and the residue was dissolved into ethyl acetate. The solution was washed with 5% KHSO<sub>4</sub>, saturated NaHCO<sub>3</sub> and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Recrystallization

(ethyl acetate/hexane) provided the sulfonamide as a solid (10.87 g, 66 %).

Part B: To a solution of the sulfonamide of part A (10.8 g, 32.7 mmol) in DMF (60 mL) was added 4-(2-chloroethyl)morpholine(12.2 g, 65.4 mmol) and K<sub>2</sub>CO<sub>3</sub> (13.6 g, 98.0 mmol) and the solution was heated to seventy degrees Celsius for 7 hours. The solution was partitioned between ethyl acetate and H<sub>2</sub>O. The organic layer was washed with H<sub>2</sub>O, saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Concentration in vacuo followed by tritration (ethyl ether) provided the morpholine compound as a solid (8.48 g, 59 %). MS(CI) MH<sup>+</sup> calculated for C<sub>19</sub>H<sub>29</sub>N<sub>3</sub>O<sub>7</sub>S: 444, found: 444.

Part C: To a solution of the morpholine compound of part B (8.49 g, 19.1 mmol) in THF (100 mL) under atmosphere of 50 psi of hydrogen was added 4% Pd/C and the solution was stirred for 2 hours until uptake stopped. The solution was filtered through Celite and concentration in vacuo of the filtrate provided the aniline as a solid (8.5 g, quantitative yield). MS(CI) MH<sup>+</sup> calculated for C<sub>19</sub>H<sub>31</sub>N<sub>3</sub>O<sub>5</sub>S: 414, found: 414.

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Part D: To a solution of the aniline of part C (2.84 g, 6.87 mmol) in THF (40 mL) cooled to zero

25 degrees Celsius was added triethylamine (2.1 mL, 15.1 mmol) followed by 4-bromobenzoyl chloride (1.96 g, 9.93 mmol) in THF (5 mL). The solution was stirred at zero degrees Celsius for 1 hour. The solution was concentrated in vacuo and the residue was partitioned

30 between ethyl acetate and saturated NaHCO3 and the organic is washed with saturated NaCl and dried over

 $Na_2SO_4$ . Chromatography (on silica gel, ethyl acetate/methanol) provided the benzamide as a solid (3.3 g, 81 %). MS(CI) MH $^+$  calculated for  $C_{26}H_{34}N_3O_6SBr$ : 596, found: 596.

Part E: To a solution of the benzamide of part D (2.84 g, 4.76 mmol) in anisole (11 mL) was added trifluoroacetic acid (32 mL) and the solution was stirred at ambient temperature for 18 hours. The solution was concentrated in vacuo to remove the trifluoroacetic acid and the residue was poured into ethyl ether. Filtration provided the acid as an off-white solid (2.8 g, quantitative yield). MS(CI) MH<sup>†</sup> calculated for C<sub>22</sub>H<sub>26</sub>N<sub>3</sub>O<sub>6</sub>SBr: 541, found 541.

Part F: To a solution of the acid of part E (2.71 g, 4.14 mmol) in methanol (10 mL) cooled to zero degrees Celsius was added thionyl chloride (0.38 mL, 5.25 mmol). The solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and saturated NaHCO3 and the organic layer was washed with saturated NaCl and dried over Na2SO4. Chromatography (on silica, ethyl acetate/methanol) provided the methyl ester as a solid (1.96 g, 85 %).

Part G: To a solution of the methyl ester of

25 part F (1.96 g, 3.53 mmol) in THF (2 mL) and methanol

(2 mL) was added 50% aqueous hydroxylamine (4.2 mL,

70.7 mmol) was added and the solution was stirred for

18 hours at ambient temperature. The solution was

concentrated in vacuo and reverse phase chromatography

30 (on silica, acetonitrile/H<sub>2</sub>O (0.05% TFA)) provided

(R)-4-bromo-N-[4-[[[2-(hydroxyamino)-1-methyl-2-

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oxoethyl] [2-(4-morpholinyl)ethyl]amino]sulfonyl]-phenyl]benzamide, mono(trifluoroacetate) salt as a white solid (350 mg, 18 %). MS(EI)  $M^{+}$  calculated for  $C_{22}H_{27}N_4O_6SBr$ : 555, found: 555.

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Example 6: (R)-N-[4-[[[2-(hydroxyamino)-1-methyl-2-oxoethyl] [2-(4-morpholinyl)ethyl]amino]-sulfonyl]phenyl]cyclohexanecarboxamide,
mono(trifluoroacetate) (salt)

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Part A: To a solution of D-alanine, t-butyl ester hydrochloride (9.80 g, 53.9 mmol) in H<sub>2</sub>O (64 mL) and acetone (26 mL) was added triethylamine (17.3 mL, 124 mmol) and the solution was cooled to zero degrees Celsius. To this solution was added 4-nitrobenzenesulfonyl chloride (11.1 g, 50.2 mmol) dropwise in acetone (25 mL). The solution was stirred for 72 hours. The solution was concentrated in vacuo and the residue was dissolved into ethyl acetate. The solution was washed with 5% KHSO<sub>4</sub>, saturated NaHCO<sub>3</sub> and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Recrystallization (ethyl

acetate/hexane) provided the sulfonamide as a solid (10.87 g, 66 %).

Part B: To a solution of the sulfonamide of part A (10.8 g, 32.7 mmol) in DMF (60 mL) was added 5 4-(2-chloroethyl)morpholine (12.2 g, 65.4 mmol) and K<sub>2</sub>CO<sub>3</sub> (13.6 g, 98.0 mmol) and the solution was heated to seventy degrees Celsius for 7 hours. The solution was partitioned between ethyl acetate and H<sub>2</sub>O. The organic layer was washed with H<sub>2</sub>O, saturated NaCl, and dried over Na<sub>2</sub>SO<sub>4</sub>. Concentration in vacuo followed by tritration (ethyl ether) provided the morpholine compound as a solid (8.48 g, 59 %). MS(CI) MH<sup>+</sup> calculated for C<sub>19</sub>H<sub>29</sub>N<sub>3</sub>O<sub>7</sub>S: 444, found: 444.

Part C: To a solution of the morpholine

15 compound of part B (8.49 g, 19.1 mmol) in THF (100 mL)

under atmosphere of 50 psi of hydrogen was added 4%

Pd/C and the solution was stirred for 2 hours until

uptake stopped. The solution was filtered through

Celite and concentration in vacuo of the filtrate

20 provided the aniline as a solid (8.5 g, quantitative

yield). MS(CI) MH<sup>+</sup> calculated for C<sub>19</sub>H<sub>31</sub>N<sub>3</sub>O<sub>5</sub>S: 414,

found: 414.

Part D: To a solution of the aniline of part C (2.70 g, 6.53 mmol) in THF (40 mL) was added

25 triethylamine (3.6 mL, 26.1 mmol) and the solution was cooled to zero degrees Celsius. To this solution was added cyclohexane carbonyl chloride (2.3 mL, 17.0 mmoL) and the solution was stirred for 30 minutes. The solution was concentrated in vacuo and the residue was partitioned between ethyl acetate and saturated NaHCO3. The organic layer was washed with saturated NaHCO3 and

saturated NaCl, and dried over  $Na_2SO_4$ . Chromatography (on silica, ethyl acetate/methanol) provided the benzamide as a solid (2.09 g, 61 %). MS(CI) MH<sup>+</sup> calculated for  $C_{26}H_{41}N_3O_6S$ : 524, found: 524.

Part E: To a solution of the benzamide of part D (2.0 g, 3.82 mmol) in anisole (10 mL) was added trifluoroacetic acid (18 mL). The solution was stirred for 18 hours at ambient temperature. The solution was concentrated in vacuo to remover the trifluoroacetic acid. The remaining solution was diluted with ethyl ester and the resulting white solid was collected by vacuum filtration to provide the acid (2.48 g, quantitative yield). MS(CI) MH+ calculated for C22H33N3O6S: 468, found: 468.

Part F: To a solution of the acid of part E 15 (1.27 g, 2.18 mmol) in DMF (10 mL) was added N-hydroxybenzotriazole (353 mg, 2.62 mmol) followed by 4-methylmorpholine (1.4 mL, 13.1 mmol), tetrahydropyranyl hydroxylamine (791 mg, 6.76 mmol) and 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide 20 hydrochloride (590 mg, 3.05 mmol). The solution was stirred for 72 hours. The solution was partitioned between ethyl acetate and saturated NaHCO3 and the organic was washed with H2O and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (on silica, ethyl 25 acetate/methanol) provided the ester as a white solid (1.2 g, quantitative yield). MS(CI) MH calculated for  $C_{27}H_{42}N_4O_7S$ : 567, found 567.

Part G: A solution of the ester of part F

30 (1.2 g, 2.12 mmol) in 4N HCl in dioxane (25 mL) was
stirred for 1 hour. The solution was diluted with

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ethyl ether and resulting white solid was collected by vacuum filtration. The solid was suspended into ethyl acetate and was washed with saturated NaHCO3 and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Reverse phase chromatography (on silica, acetonitrile/H<sub>2</sub>O(0.05% TFA)) 5 provided (R)-N-[4-[[[2-(hydroxyamino)-1-methyl-2oxoethyl] [2-(4-morpholinyl)ethyl]amino]sulfonyl]phenyl]cyclohexane-carboxamide, mono(trifluoroacetate) salt as a white solid (312 mg, 25 %). MS(CI) MH<sup>+</sup> calculated for  $C_{22}H_{34}N_4O_6S$ : 483, found 483.

(R) -N-4-[[[1-[(hydroxyamino)carbonyl]-2-Example 7: methylpropyl] [2-(4-morpholinyl)ethyl]amino] sulfonyl]-phenyl]-4-propylbenzamide, monohydrochloride

20 Part A: To a solution to D-valine (25.0 g, 213 mmol) in  $H_2O$  (180 mL) and acetone (96 mL) was added triethylamine (62 mL, 448 mmol) and was cooled to zero degrees Celsius. To this solution was added 4nitrobenzenesulfonyl chloride (45.3 g, 204 mmol) in 25 acetone (100 mL) dropwise. The solution was stirred

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for 72 hours. The solution was concentrated in vacuo and the resulting aqueous layer was extracted with toluene and acidified to pH = 2 with 2N HCl. The aqueous layer was extracted with ethyl acetate three times and the combined organic layers were washed with saturated NaCl and dried over MgSO<sub>4</sub>. Concentration in vacuo provided the sulfonamide as a light brown solid (37.15 g, 61 %).

Part B: A solution of the sulfonamide of

10 part A (37.15 g, 123 mmol) and a catalytic amount of

H<sub>2</sub>SO<sub>4</sub> in dichloromethane/dioxane (1L) was subjected to

isobutylene for 18 hours. The solution was cooled to

zero degrees Celsius and quenched with saturated NaHCO<sub>3</sub>.

The aqueous layer was extracted with ethyl acetate and

15 the organic layer was washed with saturated NaCl and

dried over MgSO<sub>4</sub>. Chromatography (on silica, ethyl

acetate/hexane) provided the t-butyl ester as a solid

(16.7 g, 38 %).

Part C: To a solution of the t-butyl ester

20 of part B (16.5 g, 46 mmol) in DMF (60 mL) was added
4-(2-chloroethyl)morpholine hydrochloride (17.2 g, 92

mmol) and K<sub>2</sub>CO<sub>3</sub> (25.5 g, 184 mmol) and the solution was
heated to sixty degrees Celsius for 7 hours. The
solution was partitioned between ethyl acetate and H<sub>2</sub>O

25 and the organic layer was washed with saturated NaCl
and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromagraphy (on silica, ethyl
acetate/hexane) provided the morpholine compound as a
solid (21.5 g, 99 %).

Part D: To a solution of the morpholine compound of part C (21.5 g, 45.6 mmol) in THF (200 mL) in a flask purged with  $H_2$  was added 4% Pd/C (3.04 g) and

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the solution was hydrogenated until uptake ceased. The solution was filtered through Celite® to remove the excess catalyst and the filtrate was concentrated in vacuo to provide the aniline as an oil (19.2 g, 95 %).

D (2.9 g, 6.6 mmol) in THF (20 mL) was added triethylamine (3.66 mL, 26.3 mmol) and cooled to four degrees Celsius. To this solution was added 4-propylbenzoyl chloride (2.0 g, 11.0 mmol) and the solution was stirred for 1 hour at ambient temperature. The solution was concentrated in vacuo and the residue was partitioned between ethyl acetate and saturated NaHCO<sub>3</sub>. The organic layer was washed with saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (on silica, ethyl acetate/hexane) provided the benzamide as a solid (3.3 g, 85 %).

Part F: To a solution of the benzamide of part E (3.2 g, 5.4 mmol) in dichloromethane (20 mL) was added trifluoroacetic acid (80 mL) and the solution was stirred for 30 minutes. The solution was concentrated in vacuo and the residue was dissolved into warm toluene/ethyl acetate and was added dropwise to ethyl ether to produce a yellow precipitate. Vacuum filtration provided the acid as a yellow solid (2.58 g, 84 %).

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Part G: To a solution of the acid of part F (2.04 g, 3.6 mmol) in DMF (5 mL) was added N-hydroxybenzotriazole (583 mg, 4.32 mmol), 4-methylmorpholine (2.37 mL, 21.6 mmol), tetrahydropyranyl hydroxylamine (1.31 g, 11.2 mmol) and 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide

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hydrochloride (966 mg, 5.04 mmol). The solution was stirred for 18 hours at ambient temperature. The solution was partitioned between ethyl acetate and saturated NaHCO<sub>3</sub>. The organic layer was washed with saturated NaHCO<sub>3</sub>, H<sub>2</sub>O and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (on silica, ethyl acetate/methanol) provided the ester as a solid (2.15 g, 95 %).

Part H: Into a solution of the ester of part G (2.15 g, 3.4 mmol) in methanol (30 mL) was bubbled HCl gas. After 1 hour the solution was concentrated in vacuo to a reduced volume (5 mL) and this solution was dropped into cooled ethyl ether to produce a precipitate. Vacuum filtration provided (R)-N-4-[[[1-[(hydroxyamino)carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]amino]sulfonyl]-phenyl]-4-propylbenzamide, monohydrochloride as a white solid (1.64 g, 83 %). MS(CI) MH\* calculated for C27H36N4O6S: 547, found 547.

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Example 8: (R)-4-butyl-N-[4-[[[1-[(hydroxyamino)-carbonyl]-2-methylpropyl][2-(4-morpholinyl)-ethyl]amino]sulfonyl]phenyl]-benzamide.monohydrochloride

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Part A: To a solution of the aniline of
Example 7, part D (2.53 g, 5.73 mmol) in THF (20 mL)

5 was added triethylamine (3.2 mL, 22.9 mmol) and the
solution was cooled to four degrees Celsius. To this
solution was added 4-butylbenzoyl chloride (1.9 g, 9.7
mmol) and the solution was stirred at ambient
temperature for 18 hours. The solution was

10 concentrated in vacuo and the residue was partitioned
between ethyl acetate and saturated NaHCO3. The organic
layer was washed with saturated NaHCO3 and saturated
NaCl and dried over Na3SO4. Chromatography (on silica,
ethyl acetate/hexane) provided the benzamide as a solid

15 (2.8 g, 82 %).

Part B: A solution of the benzamide of part A (2.8 g, 4.6 mmol) in 4N HCl in dioxane (20 mL) was stirred for 72 hours at ambient temperature. The solution was concentrated in vacuo and the residue was dissolved in dioxane (3 mL) and dropped into stirring ethyl ether. The resulting precipitate was collected by vacuum filtration to provide the acid as a white solid (2.7 g, quantitative yield).

Part C: To a solution of the acid of part B 25 (2.0 g, 3.4 mmol) in DMF (5 mL) was added N-

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hydroxybenzotriazole (557 mg, 4.13 mmol) and the solution was cooled to four degrees Celsius. To this solution was added 4-methylmorpholine (2.27 mL, 20.6 mmol), 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide

5 hydrochloride (923 mg, 4.81 mmol) and tetrahydropyranyl hydroxylamine (604 mg, 5.16 mmol) and the solution was stirred for 1 hour. The solution was partitioned between ethyl acetate and saturated NaHCO3. The organic layer was washed with saturated NaHCO3, H2O and

10 saturated NaCl, and dried over Na2SO4. Chromatography (on silica, ethyl acetate/methanol) provided the ester as a solid (2.0 g, 91 %).

Part D: To a solution of the ester of part C (2.0 g, 3.1 mmol) in methanol (1.5 mL) was added 4N HCl in dioxane (10 mL) and the solution was stirred for 18 hours. The solution was concentrated in vacuo to a smaller volume and dropped into ethyl ether. The resulting precipitate was collected by vacuum filtration to provide (R)-4-butyl-N-[4-[[1-[(hydroxyamino)-carbonyl]-2-methylpropyl][2-(4-morpholinyl)-ethyl]amino]sulfonyl]phenyl]benzamide, monohydrochloride as a white solid (1.8 g, 96 %).

MS(CI) MH\* calculated for C28H40N4O6S: 561, found: 561.

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Example 9: R-N-[4-[[[1-(hydroxyamino)carbonyl]-2methylpropyl][2-(4-orpholinyl)ethyl]amino]sulfonyl]phenyl]-4-pentylbenzamide,
monohydrochloride

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Part A: To a solution of the aniline of Example 7, part D (2.60 g, 5.88 mmol) in THF (20 mL) 5 was added triethylamine (3.2 mL, 22.8 mmol) and the solution was cooled to four degrees Celsius. solution was added 4-pentylbenzoyl chloride (2.1 g, 10.0 mmol) and the solution was stirred for 18 hours at ambient temperature. The solution was concentrated in vacuo and the residue was partitioned between ethyl acetate and saturated NaHCO3. The organic layer was washed with saturated NaHCO3 and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (ethyl acetate/hexane) provided the benzamide as a solid (2.09 q, 58 %):

Part B: A solution of the benzamide of part A (2.09 g, 3.4 mmol) in 4N HCl (20 mL) was stirred for 72 hours. The solution was concentrated in vacuo and the residue was dissolved into ethyl acetate (5 mL) and dropped into ethyl ether. The resulting precipitate was collected by vacuum filtration to provide the acid as a solid (1.9 g, 94 %).

Part C: To a solution of the acid of part B (1.52 g, 2.56 mmol) in DMF (5 mL) was added N-hydroxybenzotriazole (414 mg, 3.07 mmol) and the

solution was cooled to four degrees Celsius. To this solution was added 4-methylmorpholine (1.69 mL, 15.6 mmol), 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (687 mg, 3.58 mmol) and tetrahydropyranyl hydroxylamine (449 mg, 3.84 mmol) and was stirred for 1 hour at ambient temperature. The solution was partitioned between ethyl acetate and saturated NaHCO3 and the organic layer was washed with saturated NaHCO3, saturated NaCl and H2O and dried over Na2SO4.

10 Chromatography (ethyl acetate/methanol) provided the ester as a solid (1.54 g, 91 %).

Part D: To a solution of the ester of part C

(1.54 g, 2.34 mmol) in methanol (1 mL) was added 4N HCl

(10 mL) and the solution was stirred for 18 hours at

15 ambient temperature. The solution was concentrated in vacuo. Reverse phase chromatography (on silica, acetonitrile/H<sub>2</sub>O(HCl) provided the title compound, R-N
[4-[[[1-(hydroxyamino)carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]amino]sulfonyl]phenyl]-4
20 pentylbenzamide, monohydrochloride, as a white solid

(745 mg, 52 %). MS(CI) MH+ calculated for C<sub>29</sub>H<sub>42</sub>N<sub>4</sub>O<sub>6</sub>S:

575, found: 575.

25 Example 10: (R)-4-hexyl-N-[4-[[[1-(hydroxyamino)-carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]amino]sulfonyl]-phenylbenzamide, monohydrochloride

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Part A: To a solution of the aniline of
Example 7, part D (2.5 g, 5.7 mmol) was added

5 triethylamine (3.2 mL, 22.8 mmol) and the solution was
cooled to four degrees Celsius. To this solution was
added 4-hexylbenzoyl chloride (2.18 g, 9.69 mmol) and
the solution was stirred overnight at ambient
temperature. The solution was concentrated in vacuo
and the residue was partitioned between ethyl acetate
and saturated NaHCO<sub>3</sub>. The organic layer was washed with
saturated NaHCO<sub>3</sub> and saturated NaCl and dried over
Na<sub>2</sub>SO<sub>4</sub>. Chromatography (on silica, ethyl
acetate/hexane) provided the benzamide as a solid (2.76

15 g, 77 %).

Part B: A solution of the benzamide of part A (2.7 g, 4.3 mmol) in 4N HCl in dioxane (20 mL) was stirred for 72 hours. The solution was concentrated in vacuo and the residue was dissolved into ethyl acetate (5 mL). This solution was dropped into ethyl ether. The resulting precipitate was collected by vacuum filtration to provide the acid as a solid (2.5 g, 95%).

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Part C: To a solution of the acid of part B

25 (2.03 g, 3.33 mmol) in DMF (5 mL) was added

N-hydroxybenzotriazole (540 mg, 4.00 mmol) and the

solution was cooled to four degrees Celsius. To this solution was added 4-methylmorpholine (2.19 mL, 20.0 mmol), 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (894 mg, 4.66 mmol) and tetrahydropyranyl hydroxylamine (615 mg, 5.00 mmol) and the solution was stirred for 1 hour at ambient temperature. The solution was partitioned between ethyl acetate and saturated NaHCO3 and the organic layer was washed with saturated NaHCO3, saturated NaCl and H2O and dried over Na2SO4. Chromatography (on silica, ethyl acetate/methanol) provided the ester as a solid (2.01 g, 90 %).

Part D: To a solution of the ester of part C (2.01 g, 3.24 mmol) in methanol (1 mL) was added 4N HCl (10 mL) and the solution was stirred for 18 hours at ambient temperature. Reverse phase chromatography (on silica, acetonitrile/H<sub>2</sub>O(0.05% HCl)) provided the title compound, (R)-4-hexyl-N-[4-[[[1-(hydroxyamino)carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]amino]sulfonyl]-phenylbenzamide, monohydrochloride, as a white solid (1.23 g, 61 %).

MS(CI) MH\* calculated for C<sub>30</sub>H<sub>44</sub>N<sub>4</sub>O<sub>6</sub>S: 589, found: 589.

- 25 Example 10a: (R)-4-hexyl-N-[4-[[[1-(hydroxyamino)-carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]amino]sulfonyl]
  phenylbenzamide
- To a solution of the methyl ester of Example 10, part C (1.4 mmol) in methanol (1.6 mL) and THF (1.6

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mL) is added 50% aqueous hydroxylamine (1.6 mL). The solution is stirred for 18 hours. The solution is concentrated in vacuo and the residue was partitioned between ethyl acetate and H2O. The organic layer is 5 washed with saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Concentration of the dried organic layer in vacuo provides "10-Hydrochloride. MS(CI) MH+ calculated for C30H44N4O6S: 589.

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Example 11: (R)-N-[4-[[[1-[(hydroxyamino)carbonyl]-2methylpropyl] (3-pyridinylmethyl) aminolsulfonyllphenyll-4-propylbenzamide

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Part A: To a solution of aniline (3.3 g, 35.7 mmol) and triethylamine (8.0 g, 79 mmol) in THF, cooled to zero degrees Celsius, was added benzoyl chloride (5.0 g, 27 mmol) and the solution was stirred for 18 hours at ambient temperature. The solution was diluted with H2O and extracted with ethyl acetate. organic layer was washed with 1N HCl and saturated NaHCO3 and dried over MgSO4. Recrystallization (ethyl acetate/hexane) provided the benzamide as an off-white 25 solid (4.91 g, 64 %).

part B: To chlorosulfonic acid (2.0 g, 17.3 mmol) cooled to five degrees Celsius was added the benzamide of part A (4.91 mg, 17.3 mmol). The solution was heated to sixty-five degrees Celsius for 1 hour.

5 The solution was cooled to ambient temperature and diluted with dichloromethane. The solution was poured into cold H<sub>2</sub>O and extracted with ethyl acetate. The organic layer was washed with saturated NaHCO<sub>3</sub> and dried over MgSO<sub>4</sub>. Concentration in vacuo provided the sulfonyl chloride as a yellow solid (4.89 g, 74 %).

Part C: To a solution of D-valine, t-butyl ester (2.6 g, 15.1 mmol) in THF (25 mL) was added the sulfonyl chloride of part B (4.8 g, 12.5 mmol) followed by triethylamine (6.3 mL, 44.5 mmol) and the solution was stirred for 18 hours at ambient temperature. The solution was diluted with H<sub>2</sub>O and the resulting precipitate was collected by vacuum filtration. The solid was dissolved into ethyl acetate and dichloromethane and dried over Na<sub>2</sub>SO<sub>4</sub>. Concentration in vacuo provided the sulfonamide (4.0 g, 67 %).

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Part D: To a solution the sulfonamide of part C (2.0 g, 4.1 mmol) in DMF (10 mL) was added  $K_2CO_3$  (2.2 g, 16 mmol) followed by picolyl chloride hydrochloride (860 mg, 5.0 mmol) and the solution was stirred for 40 hours at ambient temperature. The solution was partitioned between ethyl acetate and  $H_2O$ . The organic layer was chromatographed (on silica, ethyl acetate/hexane) to provide the picolyl compound (1.3 g, 57 %).

30 Part E: To a solution of the picolyl compound of part D (1.1 g, 2 mmol) in dichloromethane

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(15 mL) was added trifluoroacetic acid (20 mL) and the solution was stirred for 20 minutes. Concentration in vacuo provided the acid as a solid (1.24 mg, quantitative yield).

Part F: To a solution of the acid of part E 5 (1.2 g, 2.0 mmol) in DMF (20 mL) was added 4-methylmorpholine (1.2 g, 12 mmol), Nhydroxybenzotriazole (800 mg, 3 mmol) and 1-(3dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (780 mg, 4 mmol). After 10 minutes of stirring, 10 tetrahydropyranyl hydroxylamine (720 mg, 6 mmol) was added and the solution was stirred for 18 hours. solution was concentrated in vacuo and the residue was partitioned between ethyl acetate and 20. The organic layer was dried over MgSO4. Chromatography (on silica, 15 ethyl acetate/hexane/methanol) provided the ester as an oil (1.4 g, quantitative yield).

Part G: A solution of the ester of part F

(1.4 g, 2 mmol) in dioxane (5 mL) and 4M HCl in dioxane

(10 mL) was stirred for 30 minutes. Dilution with
ethyl ether which precipitated a white solid followed
by collection by vacuum filtration provided the title
compound, (R)-N-[4-[[[1-[(hydroxyamino)carbonyl]-2methylpropyl](3-pyridinylmethyl)amino]sulfonyl]phenyl]
4-propylbenzamide (1.28 g, 2.3 mmol). MS(CI) MH

calculated for C<sub>27</sub>H<sub>32</sub>N<sub>4</sub>O<sub>5</sub>S: 525, found: 525.

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Example 12: (R)-N-[4-[[[1-(hydroxyamino)carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]-amino]sulfonyl]phenyl]benzamide,
monohydrochloride

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Part A: To a solution of the aniline of Example 7, part D (2.2 g, 5.0 mmol) in dichloromethane (5 mL) was added triethylamine (1.0 g, 10 mmol) and the solution was cooled to zero degrees Celsius. To this solution was added benzoyl chloride (717 mg, 5.1 mmol) in dichloromethane (5 mL). The solution was stirred for 16 hours at ambient temperature. The solution was diluted with dichloromethane and washed with saturated NaHCO3 and saturated NaCl and dried over Na2SO4. Chromatography (on silica, ethyl acetate/hexane) provided the benzamide as a solid (2.7 g, 99 %).

Part B: To a solution of the benzamide of part A (2.56 g, 4.69 mmol) in dichloromethane (50 mL) was added trifluoroacetic acid (12 mL) and the solution was stirred for 18 hours at ambient temperature. The solution was concentrated *in vacuo*. Chromatography (on silica, ethyl acetate/methanol) provided the acid as a solid (1.64 g, 71 %).

Part C: To a solution of the acid of part C (1.24 g, 2.53 mmol) in DMF (15 mL) was added Nhydroxybenzotriazole (513 mg, 3.8 mmol) and 4methylmorpholine (1.5 g, 15.2 mmol) and the solution was cooled to zero degrees Celsius. To this solution was added 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (534 mg, 2.78 mmol) and tetrahydropyranyl hydroxylamine (444 mg, 3.8 mmol) and the solution was stirred for 18 hours at ambient temperature. The solution was diluted with H<sub>2</sub>O and the solution was 10 extracted with ethyl acetate. The organic layer was washed with saturated NaHCO3 and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (on silica, ethyl acetate/hexane/methanol) provided the ester as a solid 15 (815 mg, 55 %).

Part D: To a solution of the ester of part C (750 mg, 1.27 mmol) in dioxane (5 mL) was added 4N HCl in dioxane (10 mL) and the solution was stirred for 20 minutes at ambient temperature. Trituration (hexane) provided the title compound, (R)-N-[4-[[[1-(hydroxyamino)carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]-amino]sulfonyl]phenyl]benzamide, monohydrochloride, as a white solid (590 mg, 86 %).
MS(CI) MH<sup>+</sup> calculated for C<sub>24</sub>H<sub>32</sub>N<sub>4</sub>O<sub>6</sub>S: 505, found: 505.

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Example 13: (R)-4-bromo-N-[4-[[[1-(hydroxyamino)-carbonyl]-2-methylpropyl][2-(4-morpholinyl)ethyl]amino]sulfonyl]-phenylbenzamide, monohydrochloride

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Part A: To a solution of the aniline of Example 7, part D (2.2 g, 5 mmol) in dichloromethane (50 mL) was added triethylamine (1.5 g, 15 mmol) and 4-bromobenzoyl chloride (1.65 g, 7.5 mmol) and the solution was stirred for 12 hours at ambient temperature. The solution was diluted with dichloromethane and washed with saturated NaHCO3 and saturated NaCl and dried over Na2SO4. Chromatography (on silica, ethyl acetate/hexane) provided the benzamide as a solid (2.8 g, 90 %).

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Part B: To a solution of the benzamide of part A (2.5 g, 4.0 mmol) in dichloromethane (4 mL) was added trifluoroacetic acid (16 mL) and the solution was stirred for 16 hours at ambient temperature. The solution was concentrated in vacuo and chromatography (on silica, ethyl acetate/methanol) provided the acid as a solid (1.68 g, 74 %).

Part C: To a solution of the acid of part B

20 (1.2 g, 2.11 mmol) in DMF (20 mL) was added

N-hydroxybenzotriazole (428 mg, 3.16 mmol) and

4-methylmorpholine (1.3 g, 12.7 mmol) and the solution

was cooled to zero degrees Celsius. To this solution

was added 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide

25 hydrochloride (445 mg, 2.32 mmol) and tetrahydropyranyl

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hydroxylamine (371 mg, 3.16 mmol) and the solution was stirred for 18 hours at ambient temperature. The solution was diluted with  $\rm H_2O$  and the solution was extracted with ethyl acetate. The organic layer was washed with saturated NaHCO<sub>3</sub> and saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (on silica, ethyl acetate/hexane/methanol) provided the ester as a solid (940 mg, 67 %).

Part D: To a solution of the ester of part C

(800 mg, 1.2 mmol) in dioxane (5 mL) was added 4N HCl
in dioxane (10 mL) and the solution was stirred for 20
minutes at ambient temperature. Trituration (hexane)
provided the title compound, (R)-4-bromo-N-[4-[[[1(hydroxyamino)carbonyl]-2-methylpropyl][2-(4morpholinyl)ethyl]amino]sulfonyl]phenylbenzamide,
monohydrochloride, as a white solid (668 mg, 90 %).
MS(CI) MH<sup>+</sup> calculated for C<sub>24</sub>H<sub>31</sub>N<sub>4</sub>O<sub>6</sub>SBr: 584, found:
584.

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Example 14: (R)-N-hydroxy- $\alpha$ -[[[4-(4-pentylbenzoyl)-amino]phenyl]sulfonyl]amino

Part A: To a solution of aniline (3.3 g, 35.7 mmol) and triethylamine (8.0 g, 79 mmol) in THF, cooled to zero degrees Celsius, was added 4-pentylbenzoyl chloride (5.7 g, 27 mmol) and the solution was stirred for 18 hours at ambient temperature. The solution was diluted with H2O and extracted with ethyl acetate. organic layer was washed with 1N HCl and saturated NaHCO3 and dried over MgSO4. Recrystallization (ethyl acetate/hexane) provided the benzamide as an off-white solid.

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Part B: To chlorosulfonic acid (2.3 g, 20 mmol) cooled to five degrees Celsius was added the benzamide of part A. The solution was heated to sixty-five degrees Celsius for 1 hour. The solution was cooled to ambient temperature and diluted with dichloromethane. The solution was poured into cold H<sub>2</sub>O and extracted with ethyl acetate. The organic layer was washed with saturated NaHCO3 and dried over MgSO4. Concentration in vacuo provided the sulfonyl chloride as a yellow solid (5.5 g, 55 %, two steps).

Part C: To a solution of R-phenylalanine (2.47 g, 15 mmol) in THF (100 mL) and H<sub>2</sub>O (30 mL) cooled to zero degrees Celsius was added triethylamine (4.54 g, 45 mmol) and the sulfonyl chloride of part B (5.5 g, 15 mmol) and the solution was stirred for 16 hours at ambient temperature. The solution was concentrated in vacuo and extracted with ethyl acetate. The organic layer was washed with saturated NaCl. Chromatography (on silica, ethyl acetate/hexane) provided the 30 sulfonamide as a solid (4.5 g, 61 %).

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Part D: To a solution of the sulfonamide of part C (494 mg, 1.0 mmol) in DMF (20 mL) was added N-hydroxybenzotriazole (203 mg, 1.5 mmol) and the solution was cooled to ten degrees Celsius. To this solution was added 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (211 mg, 1.1 mmol) and tetrahydropyranyl hydroxylamine (351 mg, 3.0 mmol) and the solution was stirred for 12 hours at ambient temperature. The solution was diluted with H<sub>2</sub>O and the resulting precipitate was extracted with ethyl acetate and washed with saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (ethyl acetate/hexane) provided the ester as a solid (327 mg, 57 %).

Part E: To a solution of the ester of part D

(200 mg, 0.34 mmol) in dioxane (3 mL) was added 4N HCl
in dioxane (5 mL) and the solution was stirred for 1
hour at ambient temperature. The solution was
concentrated in vacuo and reverse phase chromatography
(acetonitrile/H<sub>2</sub>O) provided the title compound, (R)-N
hydroxy-α-[[[4-(4-pentylbenzoyl)amino]phenyl]sulfonyl]amino benzenepropanamide, as a white solid (80
mg, 26 %). MS(CI) MH<sup>+</sup> calculated for C<sub>27</sub>H<sub>31</sub>N<sub>3</sub>O<sub>5</sub>S: 510,
found: 510.

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Example 15: (R)-N-hydroxy-α-[[2-(4-morpholinyl)ethyl][[4-[(4-pentylbenzoyl)amino]phenyl]sulfonyl]amino]benzenepropanamide,
monohydrochloride

Part A: To a solution of the sulfonamide of Example 14, part C (7.0 g, 15 mmol) in toluene (120 mL) was added DMF di-tert-butylacetal (6.1 g, 30 mmol) and the solution was heated at one hundred degrees Celsius for 1 hour. Concentration in vacuo followed by recrystallization (cold methanol) provided the ester as a solid (3.7 g, 45 %).

Part B: To a solution of the ester of part A

10 (3.5 g, 6.35 mmol) and K<sub>2</sub>CO<sub>3</sub> (5.53 g, 40 mmol) in DMF

(100 mL) was added 4-(2-chloroethyl)morpholine

hydrochloride (1.77 g, 9.52 mmol) and the solution was

stirred for 16 hours at sixty degrees Celsius. The

solution was partitioned between ethyl acetate and H<sub>2</sub>O

15 and the organic layer was washed with saturated NaCl

and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (ethyl

acetate/hexane) provided the morpholine compound as a

solid (1.8 g, 43 %).

Part C: To a solution of the morpholine

20 compound of part B (1.4 g, 2.1 mmol) in dichloromethane

(5 mL) was added trifluoroacetic acid (12 mL) and the

solution was stirred for 16 hours at ambient

temperature. Concentration in vacuo followed by

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reverse phase chromatography (acetonitrile/ $H_2O$ ) provided the acid as a solid (1.12 g, 87 %).

Part D: To a solution of the acid of part C (607 mg, 1.0 mmol) in DMF (40 mL) was added N
hydroxybenzotriazole (207 mg, 1.5 mmol) and the solution was cooled to two degrees Celsius. To this solution was added 1-(3-dimethylaminopropyl)-3ethylcarbodiimide hydrochloride (211 mg, 1.1 mmol) and tetrahydropyranyl hydroxylamine (585 mg, 5.0 mmol) and the solution was stirred for 6 hours at ambient temperature. The solution was diluted with H<sub>2</sub>O and the resulting precipitate was extracted with ethyl acetate and washed with saturated NaCl and dried over Na<sub>2</sub>SO<sub>4</sub>. Chromatography (on silica, ethyl acetate/hexane)

provided the ester as a solid.

Part E: To a solution of the ester of part D in dioxane (3 mL) was added 4N HCl in dioxane (5 mL) and the solution was stirred for 1 hour at ambient temperature. The solution was concentrated in vacuo and reverse phase chromatography (acetonitrile/ $H_2O$ ) provided the title compound, (R)-N-hydroxy- $\alpha$ -[[2-(4-morpholinyl)ethyl]-[[4-[(4-pentylbenzoyl)amino]phenyl]-sulfonyl]amino]benzenepropanamide, monohydrochloride, as a white solid (80 mg, 12 %, two steps). MS(CI) MH<sup>+</sup> calculated for  $C_{33}H_{42}N_4O_6S$ : 660, found: 660.

### Example 16: In Vitro Metalloprotease Inhibition

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The compounds prepared in the manner described in Examples 1 to 15 were assayed for activity

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by an in vitro assay. Following the procedures of Knight et al., FEBS Lett. 296(3):263 (1992). Briefly, 4-aminophenylmercuric acetate (APMA) or trypsin activated MMPs were incubated with various concentrations of the inhibitor compound at room temperature for 5 minutes.

More specifically, recombinant human MMP-13 and MMP-1 enzymes were prepared in laboratories of the assignee. MMP-13 was expressed in baculovirus as a proenzyme, and purified first over a heparin agarose column and then over a chelating zinc chloride column. The proenzyme was activated by APMA for use in the assay. MMP-1 expressed in transfected HT-1080 cells was provided by Dr. Howard Welgus of Washington University, St. Louis, MO. The enzyme was also activated using APMA and was then purified over a hydroxamic acid column.

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The enzyme substrate is a methoxycoumarincontaining polypeptide having the following sequence:

MCA-ProLeuGlyLeuDpaAlaArgNH², wherein MCA is methoxycoumarin and Dpa is 3-(2,4-dinitrophenyl)-L-2,3-diaminopropionyl alanine. This substrate is commercially available from Baychem as product M-1895.

The buffer used for assays contained 100 mM 25 Tris-HCl, 100 mM NaCl, 10 mM CaCl<sub>2</sub> and 0.05 percent polyethyleneglycol (23) lauryl ether at a pH value of 7.5. Assays were carried out at room temperature, and dimethyl sulfoxide (DMSO) at a final concentration of 1 percent was used to dissolve inhibitor compound.

The assayed inhibitor compound in DMSO/buffer solution was compared to an equal amount of DMSO/buffer

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with no inhibitor as control using Microfluor White Plates (Dynatech). The inhibitor or control solution was maintained in the plate for 10 minutes and the substrate was added to provide a final concentration of 4  $\mu M$ .

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In the absence of inhibitor activity, a fluorogenic peptide was cleaved at the gly-leu peptide bond, separating the highly fluorogenic peptide from a 2,4-dinitrophenyl quencher, resulting in an increase of fluorescence intensity (excitation at 328 nm/emission at 415 nm). Inhibition was measured as a reduction in fluorescent intensity as a function of inhibitor concentration, using a Perkin Elmer L550 plate reader. The IC50 values were calculated from those values. The results are set forth in the Inhibition Table below as Table 9, reported in terms of IC50.

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Table 9

#### INHIBITION TABLE

Example	hMMP-1	hMMP-2	hMMP-3	hMMP-8	hMMP-9	hMMP-13
-	(MM)	(Mac)	(Ma)	(MA)	(MM)	(MA)
4	>10,000				20.0	2.5
4a	>10,000	0.2			24.0	2.2
5	4,850	-	- , -	*		0.35
6	>10,000				3,500	250
7	4,000	0.2	90.0	9.0	4.5	0.3
8	4,000	0.2	50.0	25.8	31.0	0.1
9	4,000	<0.1	55.0	8.0	200	<0.1
10	5,600				350	<0.1
10a		<0.1	180	2.6		
11	>10,000	0.1	225	39.0	24.0	0.5
12	3,400		800	12.3	245	11.4
13	880				1.9	0.4
14 .	>10,000	<0.1	30.0	34.0	161	<0.1
15	>10,000	<0.1	140	241	286	<0.1

•

Example 17: <u>In Vivo Angiogenesis Assay</u>

The study of angiogenesis depends on a reliable and reproducible model for the stimulation and inhibition of a neovascular response. The corneal micropocket assay provides such a model of angiogenesis in the cornea of a mouse. See, A Model of Angiogenesis in the Mouse Cornea; Kenyon, BM, et al., Investigative

PCT/US98/04299 WO 98/39329

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Ophthalmology & Visual Science, July 1996, Vol. 37, No. 8.

In this assay, uniformly sized Hydron<sup>TM</sup> pellets containing bFGF and sucralfate are prepared and surgically implanted into the stroma mouse cornea adjacent to the temporal limbus. The pellets are formed by making a suspension of 20 µL sterile saline containing 10 µg recombinant bFGF, 10 mg of sucralfate and 10 µL of 12 percent Hydron<sup>TM</sup> in ethanol. The slurry is then deposited on a 10  $\times$  10 mm piece of sterile nylon mesh. After drying, the nylon fibers of the mesh are separated to release the pellets.

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The corneal pocket is made by anesthetizing a 7 week old C57Bl/6 female mouse, then proptosing the 15 eye with a jeweler's forceps. Using a dissecting microscope, a central, intrastromal linear keratotomy of approximately 0.6 mm in length is performed with a #15 surgical blade, parallel to the insertion of the lateral rectus muscle. Using a modified cataract knife, a lamellar micropocket is dissected toward the 20 temporal limbus. The pocket is extended to within 1.0 mm of the temporal limbus. A single pellet is placed on the corneal surface at the base of the pocket with a jeweler's forceps. The pellet is then advanced to the 25 temporal end of the pocket. Antibiotic ointment is then applied to the eye.

Mice are dosed on a daily basis for the duration of the assay. Dosing of the animals is based on bioavailability and overall potency of the compound. an exemplary dose is 50 mg/kg bid, po. Neovascularization of the corneal stroma begins at

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about day three and is permitted to continue under the influence of the assayed compound until day five. At day five, the degree of angiogenic inhibition is scored by viewing the neovascular progression with a slit lamp microscope.

The mice are anesthetized and the studied eye is once again proptosed. The maximum vessel length of neovascularization, extending from the limbal vascular plexus toward the pellet is measured. In addition, the contiguous circumferential zone of neovascularization is measured as clock hours, where 30 degrees of arc equals one clock hour. The area of angiogenesis is calculated as follows.

$$area = \frac{(0.4 \times clock\ hours \times 3.14 \times vessel\ length\ (in\ mm))}{2}$$

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The studied mice are thereafter compared to control mice and the difference in the area of neovascularization is recorded. A contemplated compound typically exhibits about 25 to about 75 percent inhibition, whereas the vehicle control exhibits zero percent inhibition.

Without further elaboration, it is believed

25 that one skilled in the art can, using the preceding
description, utilize the present invention to its fullest
extent. The foregoing specific embodiments are,
therefore, to be construed as merely illustrative, and

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not limitative of the remainder of the disclosure in any way whatsoever.

From the forgoing description, one skilled in the art can easily ascertain the essential

5 characteristics of this invention, and without departing from the spirit and scope thereof, can make various changes and modifications of the invention to adapt it to various usages and conditions.

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What is Claimed is:

### 1. A compound of formula VII

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$$R^{14}O \longrightarrow R^{2} \longrightarrow R^{1} \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O \longrightarrow R^{14}O$$

wherein:

n is zero, 1 or 2;

10

W is independently selected from the group consisting of  $-NR^5COR^6$ ,  $-NR^5S(O)_zR^7$  where z is zero, 1, or 2,  $-NR^5COOR^8$ ,  $-NR^5CONR^8R^9$ , and  $-NR^{11}R^{12}$ ;

R1 is cycloalkylene, arylene or heteroarylene;

15 R<sup>2</sup> is selected from the group consisting of a hydrido, alkyl, aralkyl, heteroaralkyl, cycloalkylalkyl, heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl, hydroxycarbonylalkyl, aroylalkyl, and heteroaroylalkyl group, -(CH<sub>2</sub>)<sub>X</sub>-NR<sup>11</sup>R<sup>12</sup>, or -(CH<sub>2</sub>)<sub>X</sub>-

20 C(O)-NR11R12 wherein x is an integer from zero to 6;

R<sup>3</sup> is selected from the group consisting of a hydrido, alkyl, aryl, aralkyl, thioalkyl, heteroaralkyl, heteroaryl, alkoxyalkoxyalkyl, trifluoromethylalkyl, alkoxycarbonylalkyl,

25 aralkoxycarbonylalkyl, hydroxycarbonylalkyl, alkoxyalkyl, heterocycloalkylalkyl, aryloxyalkyl, alkylthioalkyl, arylthioalkyl, heteroarylthioalkyl group, or a sulfoxide or sulfone of any of said thiocontaining groups, a -(CH<sub>2</sub>)<sub>X</sub>-C(O)NR<sup>11</sup>R<sup>12</sup> group, wherein x is an integer from zero to 6, and a -(CH<sub>2</sub>)<sub>Y</sub>-W group, wherein y is an integer from 1 to 6 and W is defined above:

or  $\mathbb{R}^2$  and  $\mathbb{R}^3$  together with the atom chain to which they are attached form a 3-8 membered ring;

10 R4 is a hydrido or C<sub>1</sub>-C<sub>4</sub> alkyl group;

R<sup>5</sup> is a hydrido or C<sub>1</sub>-C<sub>4</sub> alkyl group;

R6 is selected from the group consisting of a hydrido, cycloalkyl, heterocycloalkyl, aryl,

heteroaryl, aralkyl, heteroaralkyl, cycloalkylalkyl,

15 heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl group, and a  $-(CH_2)_{X}-NR^{11}R^{12}$  group wherein x is an

integer from zero to 6, wherein an aryl or heteroaryl

group of R6 is optionally substituted with one or more substituents independently selected from the group

consisting of a halo,  $C_1$ - $C_4$  alkyl,  $C_1$ - $C_4$  alkoxy, nitro,

cyano, hydroxy, carboxy, hydroxycarbonylalkyl,

-(CH $_2$ ) $_{\rm X}$ -NR11R12, wherein x is an integer from zero to

6, trifluoromethyl, alkoxycarbonyl, aminocarbonyl,

thio, alkylsulfonyl, carbonylamino, aminosulfonyl,

25 alkylsulfonamino, alkoxyalkyl, cycolalkyloxy,

alkylthioalkyl and alkylthio group;

or R5 and R6 together with the atom chain to which they are bonded form a 5- to 7-membered a cyclic amide or imide that is substituted or

30 unsubstituted;

 $\rm R^7$  is selected from the group consisting of  $\rm R^6$  and alkyl; or  $\rm R^5$  and  $\rm R^7$  together with the atom chain to which they are bonded form a 5- to 7-membered a cyclic sulfonamide that is substituted or

5 unsubstituted;

R8 and R9 are independently selected from the group consisting of R6 and alkyl; or R8 and R9 together with the depicted nitrogen atom form a 5- to 7-membered ring containing zero or one heteroatom that is oxygen, nitrogen or sulfur;

R11 and R12 are independently selected from the group consisting of a hydrido, alkyl, aryl, aralkyl, heteroaryl, heteroaralkyl, cycloalkyl, alkanoyl, aralkanoyl, and heteroaralkanoyl group; or R11 and R12 taken together form a 5 to 8-membered heterocyclo or heteroaryl ring;

 $$\rm R13$  is a hydrido or C1-C6 alkyl group; and  $$\rm R14$  is H, C1-C6 alkyl, aryl, substituted aryl, arylalkyl or substituted arylalkyl.

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- $\label{eq:compound} \textbf{2.} \quad \textbf{The compound according to claim 1 wherein} \\ \textbf{n is 2.}$
- 3. The compound according to claim 1 wherein  $\mathbb{R}^{1}$  is arylene.
  - 4. The compound according to claim 1 wherein W is  $-NR^5COR^6$ .

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- 5. The compound according to claim 1 wherein W is  $-NR^5S(0)zR^7$ .
- 6. The compound according to claim 1 wherein  $5 \text{ W is -NR}^5\text{COOR}^7$ .

#### 7. A compound of formula I

HO 
$$R^{13}$$
  $R^4$   $R^3$   $\binom{\parallel}{0}_n$ 

10

wherein:

n is zero, 1 or 2;

W is independently selected from the group consisting of -NR5COR6, -NR5S(O)  $_{\rm Z}{\rm R}^7$  where z is zero, 1,

15 or 2, -NR5COOR8, -NR5CONR8R9, and -NR11R12;

R1 is cycloalkylene, arylene or heteroarylene;

R<sup>2</sup> is selected from the group consisting of a hydrido, alkyl, aralkyl, heteroaralkyl,

cycloalkylalkyl, heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl, hydroxycarbonylalkyl, aroylalkyl, and heteroaroylalkyl group, -(CH<sub>2</sub>)<sub>X</sub>-NR<sup>11</sup>R<sup>12</sup>, or -(CH<sub>2</sub>)<sub>X</sub>-C(O)-NR<sup>11</sup>R<sup>12</sup> wherein x is an integer from zero to 6;

R3 is selected from the group consisting of a hydrido, alkyl, aryl, aralkyl, thioalkyl, heteroaralkyl, heteroaryl, alkoxyalkoxyalkyl,

trifluoromethylalkyl, alkoxycarbonylalkyl, aralkoxycarbonylalkyl, hydroxycarbonylalkyl, alkoxyalkyl, heterocycloalkylalkyl, aryloxyalkyl, alkylthioalkyl, arylthioalkyl, heteroarylthioalkyl group, or a sulfoxide or sulfone of any of said thiocontaining groups, a -(CH<sub>2</sub>)<sub>X</sub>-C(O)NR<sup>11</sup>R<sup>12</sup> group, wherein x is an integer from zero to 6, and a -(CH<sub>2</sub>)<sub>Y</sub>-W group, wherein y is an integer from 1 to 6 and W is defined above;

or  $R^2$  and  $R^3$  together with the atom chain to which they are attached form a 3-8 membered ring;

R<sup>4</sup> is a hydrido or C<sub>1</sub>-C<sub>4</sub> alkyl group; R<sup>5</sup> is a hydrido or C<sub>1</sub>-C<sub>4</sub> alkyl group;

15 R<sup>6</sup> is selected from the group consisting of a hydrido, cycloalkyl, heterocycloalkyl, aryl, heteroaryl, aralkyl, heteroaralkyl, cycloalkylalkyl, heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl group, and a -(CH<sub>2</sub>)<sub>X</sub>-NR<sup>11</sup>R<sup>12</sup> group wherein x is an 20 integer from zero to 6, wherein an aryl or heteroaryl group of R<sup>6</sup> is optionally substituted with one or more substituents independently selected from the group consisting of a halo, C<sub>1</sub>-C<sub>4</sub> alkyl, C<sub>1</sub>-C<sub>4</sub> alkoxy, nitro, cyano, hydroxy, carboxy, hydroxycarbonylalkyl,

25 -(CH<sub>2</sub>)<sub>X</sub>-NR<sup>11</sup>R<sup>12</sup>, wherein x is an integer from zero to 6, trifluoromethyl, alkoxycarbonyl, aminocarbonyl, thio, alkylsulfonyl, carbonylamino, aminosulfonyl, alkylsulfonamino, alkoxyalkyl, cycolalkyloxy, alkylthioalkyl and alkylthio group;

or R5 and R6 together with the atom chain to which they are bonded form a 5- to 7-membered a cyclic amide or imide that is substituted or unsubstituted;

R7 is selected from the group consisting of R6 and alkyl; or R5 and R7 together with the atom chain to which they are bonded form a 5- to 7-membered a cyclic sulfonamide that is substituted or unsubstituted;

10 R<sup>8</sup> and R<sup>9</sup> are independently selected from the group consisting of R<sup>6</sup> and alkyl, or R<sup>8</sup> and R<sup>9</sup> together with the depicted nitrogen atom form a 5- to 7-membered ring containing zero or one heteroatom that is oxygen, nitrogen or sulfur;

15 R11 and R12 are independently selected from the group consisting of a hydrido, alkyl, aryl, aralkyl, heteroaryl, heteroaralkyl, cycloalkyl, alkanoyl, aralkanoyl, and heteroaralkanoyl group, or R11 and R12 taken together form a 5 to 8-membered
20 heterocyclo or heteroaryl ring; and

R13 is a hydrido or C1-C6 alkyl group.

8. The compound according to claim 7 wherein n is 2.

- 9. The compound according to claim 8 wherein  $\mathbb{R}^1$  is arylene.
- 10. The compound according to claim 9  $\,$  30 wherein  $R^4$  is hydrido.

- 11. The compound according to claim 7 wherein W is  $-NR^5COR^6$ .
- 5 12. The compound according to claim 7 that corresponds in structure to formula III

HO 
$$R^3$$
  $R^4$   $R^3$   $(0)_2$ 

- 10 wherein  $R^2$ ,  $R^3$ ,  $R^4$ ,  $R^6$  and  $R^{13}$  are defined above.
  - $\mbox{13.} \quad \mbox{The compound according to claim 12} \\ \mbox{wherein $R^6$ and $R^4$ are both hydrido.}$
  - $\label{eq:compound} \mbox{14.} \quad \mbox{The compound according to claim 13}$  wherein  $\mbox{R}^{3}$  is heterocycloalkylalkyl.
- 20 15. The compound according to claim 14 wherein  $R^6$  is aryl.

16. The compound according to claim 15 corresponding in structure to the formula

- 17. The compound according to claim 14 wherein  ${\sf R}^6$  is heteroarylaryl.
- 18. The compound according to claim 1510 corresponding in structure to the formula

- 19. The compound according to claim 7 wherein W is  $-NR^5S(0)_{\mathbf{Z}}R^7$  where z is 2.
- 15 20. The compound according to claim 7 that corresponds in structure to formula  $\boldsymbol{V}$

HO N R<sup>3</sup> 
$$\begin{pmatrix} R^2 \\ R^3 \end{pmatrix}_2$$
  $\begin{pmatrix} R^7 \\ Q \end{pmatrix}_2$ 

wherein  $\mathbb{R}^2,~\mathbb{R}^3,~\mathbb{R}^4,~\mathbb{R}^7$  and  $\mathbb{R}^{13}$  are defined above.

5 21. The compound according to claim 18 that corresponds in structure to the formula

10 22. A process for treating a host mammal having a condition associated with pathological matrix metalloprotease activity that comprises administering a compound corresponding in structure to Formula I in an MMP enzyme-inhibiting effective amount to a mammalian host having such a condition:

HO 
$$R^{13}$$
  $R^4$   $R^3$   $\binom{\parallel}{0}_n$ 

#### wherein:

n is zero, 1 or 2;

W is independently selected from the group consisting of  $-NR^5COR^6$ ,  $-NR^5S(O)_ZR^7$  where z is zero, 1, or 2,  $-NR^5COOR^8$ ,  $-NR^5CONR^8R^9$ , and  $-NR^{11}R^{12}$ ;

R1 is cycloalkylene, arylene or heteroarylene;

10 R<sup>2</sup> is selected from the group consisting of a hydrido, alkyl, aralkyl, heteroaralkyl, cycloalkylalkyl, heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl, hydroxycarbonylalkyl, aroylalkyl, heteroaroylalkyl, -(CH<sub>2</sub>)<sub>X</sub>-NR<sup>11</sup>R<sup>12</sup>, and -(CH<sub>2</sub>)<sub>X</sub>-C(O)-

NR11R12 group, wherein x is an integer from zero to 6;

R3 is selected from the group consisting of a hydrido, alkyl, aryl, aralkyl, thioalkyl, heteroaralkyl, heteroaryl, alkoxyalkoxyalkyl, trifluoromethylalkyl, alkoxycarbonylalkyl,

aralkoxycarbonylalkyl, hydroxycarbonylalkyl, alkoxyalkyl, heterocycloalkylalkyl, aryloxyalkyl, alkylthioalkyl, arylthioalkyl, heteroarylthioalkyl group, or a sulfoxide or sulfone of any of said thiocontaining groups, a - (CH<sub>2</sub>)<sub>X</sub>-C(O)NR<sup>11</sup>R<sup>12</sup> group, wherein

25 x is an integer from zero to 6, and a -  $(CH_2)_y$ -W group,

wherein y is an integer from 1 to 6 and W is defined above;

or  $\mathbb{R}^2$  and  $\mathbb{R}^3$  together with the atom chain to which they are attached form a 3-8 membered ring;

5

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 $R^4$  is a hydrido or  $C_1$ - $C_4$  alkyl group;  $R^5$  is a hydrido or  $C_1$ - $C_4$  alkyl group;

R6 is selected from the group consisting of a hydrido, cycloalkyl, heterocycloalkyl, aryl,

- heteroaryl, aralkyl, heteroaralkyl, cycloalkylalkyl, heterocycloalkylalkyl, alkoxyalkyl, alkylthioalkyl group, and a -(CH<sub>2</sub>)<sub>X</sub>-NR<sup>11</sup>R<sup>12</sup> group wherein x is an integer from zero to 6, wherein an aryl or heteroaryl group of R<sup>6</sup> is optionally substituted with one or more substituents independently selected from the group consisting of a halo, C<sub>1</sub>-C<sub>4</sub> alkyl, C<sub>1</sub>-C<sub>4</sub> alkoxy, nitro, cyano, hydroxy, carboxy, hydroxycarbonylalkyl,
  - -( $\rm CH_2$ ) $_{\rm X}$ -NR<sup>11</sup>R<sup>12</sup>, wherein x is an integer from zero to 6, trifluoromethyl, alkoxycarbonyl, aminocarbonyl, thio, alkylsulfonyl, carbonylamino, aminosulfonyl,
  - alkylsulfonamino, alkoxyalkyl, cycolalkyloxy, alkylthioalkyl and alkylthio group;

or R5 and R6 together with the atom
chain to which they are bonded form a 5- to 7-membered
25 a cyclic amide or imide that is substituted or
unsubstituted;

 $$\rm R^{7}$  is selected from the group consisting of  $\rm R^{6}$  and alkyl; or  $\rm R^{5}$  and  $\rm R^{7}$  together with the atom chain to which they are bonded form a 5- to 7-membered a

cyclic sulfonamide that is substituted or unsubstituted;

R<sup>8</sup> and R<sup>9</sup> are independently selected from the group consisting of R<sup>6</sup> and alkyl, or R<sup>8</sup> and R<sup>9</sup> together with the depicted nitrogen atom form a 5- to 7-membered ring containing zero or one heteroatom that is oxygen, nitrogen or sulfur;

R11 and R12 are independently selected from the group consisting of a hydrido, alkyl, aryl, 10 aralkyl, heteroaryl, heteroaralkyl, cycloalkyl, alkanoyl, aralkanoyl, and heteroaralkanoyl group, or R11 and R12 taken together form a 5 to 8-membered heterocyclo or heteroaryl ring; and

R13 is a hydrido or C1-C6 alkyl group.

- 23. The process according to claim 22 wherein n is 2.
- 24. The process according to claim 22 20 wherein  $\mathbb{R}^1$  is arylene.
  - \$25.\$ The process according to claim 22 wherein W is  $-NR^5COR^6.$
- 26. The process according to claim 22 wherein W is -NR<sup>5</sup>S(O)zR<sup>7</sup>.
- 27. The process according to claim 22 wherein said compound corresponds in structure to 30 formula III

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HO 
$$R^3$$
  $R^4$   $R^3$   $(0)_2$ 

wherein R2, R3, R4, R6 and R13 are.

- 5 defined above.
  - \$28.\$ The process according to claim 22 wherein said compound corresponds in structure to formula  $V\/$

HO 
$$R^3$$
  $R^3$   $R$ 

10

wherein  ${\rm R}^2,~{\rm R}^3,~{\rm R}^4,~{\rm R}^7$  and  ${\rm R}^{13}$  are defined above.

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29. The process according to claim 22 wherein said compound is administered a plurality of times.

## INTERNATIONAL SEARCH REPORT

Form PCT/ISA/210 (second sheet)(July 1992)\*

International application No. PCT/US98/04299

	SSIFICATION OF SUBJECT MATTER :CO7D 413/12, 265/30; A61K 31/535, 31/44						
US CL. :Please See Extra Sheet.							
	o International Patent Classification (IPC) or to both n	ational classification and IPC					
	DS SEARCHED ocumentation searched (classification system followed	by classification symbols)					
	Please See Extra Sheet.	• • • • • • • • • • • • • • • • • • • •					
Documentat	ion searched other than minimum documentation to the	extent that such documents are included	in the fields searched				
Electronic d	ata base consulted during the international search (nam	ne of data base and, where practicable,	search terms used)				
	S ONLINE						
c. Doc	UMENTS CONSIDERED TO BE RELEVANT						
Category*	Citation of document, with indication, where app	ropriate, of the relevant passages	Relevant to claim No.				
Y, P	US 5,646,167 A (MACPHERSON en document.	al.) 08 July 1997, entire	1-29				
Y	US 5,506,242 A (MACPHERSON et document.	1-29					
Y	US 5,552,419 A (MACPHERSON et al.) 03 September 1996, entire document.						
A	US 4,337,197 A (GORDON et al.) 29 June 1982, entire document. 1-29						
X	EP 0 757 984 A1 (ONO PHARMACEUTICAL CO., LTD.) 12 1-29 February 1997, entire document.						
X Furt	X Further documents are listed in the continuation of Box C. See patent family annex.						
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Name and mailing address of the ISA/US Commissioner of Patents and Trademarks Box PCT Washington, D.C. 20231		Authorized officer  MATTHEW V. GRUMBLING	March _				
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## INTERNATIONAL SEARCH REPORT

International application No. PCT/US98/04299

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Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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WO 95/35276 A1 (BRITISH BIOTECH PHARMACEUTICALS LIMITED) 28 December 1995, entire document.	1-29
WO 98/07742 A1 (ZENECA LIMITED) 26 February 1998, entire document.	1-29
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## INTERNATIONAL SEARCH REPORT

International application No. PCT/US98/04299

A. CLASSIFICATION OF SUBJECT MATTER: US CL :

544/85, 133 146, 130, 131, 137, 141, 148; 546/247, 233, 234, 236, 237, 265; 548/204; 514/231.5, 231.8, 237.2, 316, 318, 236, 342, 343, 365

B. FIELDS SEARCHED
Minimum documentation searched
Classification System: U.S.

544/85, 133 146, 130, 131, 137, 141, 148; 546/247, 233, 234, 236, 237, 265; 548/204; 514/231.5, 231.8, 237.2, 316, 318, 236, 342, 343, 365

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